

PROGRAMME STRUCTURE

School of Pharmacy B.Pharm

Programme Code: SOP0101

Batch: 2024-2028





School of Pharmacy Programme Structure

S. No	Course Code	Course Name	Category	L	Т	P	Credits
	ester I				T		
1	BP101T	Human Anatomy and Physiology I— Theory	Core	3	1	-	4
2	BP102T	Pharmaceutical Analysis I–Theory	Core	3	1	-	4
3	BP103T	Pharmaceutics I— Theory	Core	3	1	-	4
4	BP104T	Pharmaceutical Inorganic Chemistry– Theory	Core	3	1	-	4
5	BP105T	Communication skills– Theory*	Skill Enhance ment course	2	-	-	2
6	BP106RBT BP106RMT	Remedial Biology/ Remedial Mathematics—Theory*	Core	2	-	-	2
7	BP107P	Human Anatomy and Physiology— Practical	Practical	-	-	4	2
8	BP108P	Pharmaceutical Analysis I–Practical	Practical	-	-	4	2
9	BP109P	Pharmaceutics I— Practical	Practical	-	-	4	2
10	BP110P	Pharmaceutical Inorganic Chemistry Practical	Practical	-	-	4	2
11	BP111P	Communication skills–Practical*	Practical	-	-	2	1
12	BP112RBP	BP112RBP Remedial Biology— Practical		-	2	1	
		Total Credits:	: 27			_	





Seme	ester II						
1	BP201T	Human Anatomy and Physiology II – Theory	Core	3	1	-	4
2	BP202T	Pharmaceutical Organic Chemistry I -Theory	Core	3	1	-	4
3	BP203T	Biochemistry– Theory	Core	3	1	-	4
4	BP204T	Pathophysiology— Theory	Core	3	1	ı	4
5	BP205T	Computer Applications in Pharmacy—Theory*	Core	3	-	-	3
6	BP206T	Environmental sciences –Theory*	Skill mental Enhance		-	-	3
7	BP207P	Human Anatomy and Physiology II – Practical	Core	-	-	4	2
8	BP208P	Pharmaceutical Organic Chemistry I–Practical	Core	-	-	4	2
9	BP209P	Biochemistry– Practical	Core	-	-	4	2
10	BP210P	Computer Applications in Pharmacy–Practical*	Ability Enhance ment course	-	_	2	1
		Total Cred	its: 29				





Semest	er-III		T		,	T	
1		Pharmaceutical					
	BP301T	Organic	Core	3	3 1	_	4
		Chemistry II –					'
2		Theory Physical					
	BP302T	Pharmaceutics I–	Core	3	1	_	4
	210021	Theory			_		4
3		Pharmaceutical					
	BP303T	Microbiology-	Core	3	1	-	4
		Theory					
4	BP304T	Pharmaceutical	Core	3	1	_	4
-		Engineering-Theory	4 7 474				7
5.	KVE301	Universal Human value & Professional	Ability enhancem	3	0	0	
	KVE301	Ethics	ent course	3	U	U	3
5		Pharmaceutical	cht course				
		Organic Chemistry					
	BP305P	II–	Practical	Practical -	-	4	2
		Practical					
6		Physical					
	BP306P	Pharmaceutics I—	Practical	-	-	4	2
_		Practical					
7	DD207D	Pharmaceutical					
	BP307P	Microbiology-	Practical	-	-	4	2
8		Practical Pharmaceutical					
0	BP308P	Engineering—	Core	_		4	
	Dr 300r	Practical	Core	-		4	2
		Total Cred	its: 27			1	1





Say	mester IV						
1 1	mester IV	Pharmaceutical	T		I		
1	BP401T	Organic Chemistry III Theory	Core	3	1	-	4
2	BP402T	Medicinal Chemistry I–Theory	Core	3	1	-	4
3	BP403T	Physical Pharmaceutics II— Theory	Core	3	1	-	4
4	BP404T	Pharmacology I— Theory			1	-	4
5	BP405T	Pharmacognosy and PhytochemistryI —Theory	Core	3	1	-	4
6	BP406P	Medicinal Chemistry I– Practical	Practical	-	-	4	2
7	BP407P	Physical Pharmaceutics II— Practical	Practical	-	-	4	2
8	BP408P	Pharmacology I— Practical	Practical	-	-	4	2
9	BP409P	Pharmacognosy and Phytochemistry I —Practical	Practical	-	-	4	2
	Total Credits: 28						





Sem	iester-V						
1	BP501T	Medicinal Chemistry II–Theory	' LARA 3		1	-	4
2	BP502T	Industrial Pharmacy I—Theory	Core	3	1	-	4
3	BP503T	Pharmacology II— Theory	Core	3	1	-	4
4	BP504T	Pharmacognosy and Phytochemistry II–Theory	Pharmacognosy and Phytochemistry Core		1	-	4
5	BP505T	Pharmaceutical Jurisprudence Theory	naceutical rudence– Core		1	-	4
6	BP506P	Industrial Pharmacy I— Practical	Practical	-	-	4	2
7	BP507P	Pharmacology II— Practical	Practical	-	-	4	2
8	BP508P	Pharmacognosy and Phytochemistry II— Practical	Practical	-	-	4	2
9	PCC301	Community connect	Practical	-	-	4	2
		Total Credits: 28					





Seme	ester-VI						
1	BP601T	Medicinal Chemistry III–Theory	Core	3	1	-	4
2	BP602T	Pharmacology III— Theory	Core	3	1	-	4
3	BP603T	Herbal Drug Technology–Theory	Core	3	1	-	4
4	BP604T	Biopharmaceutics and Pharmacokinetics— Theory	Biopharmaceutics and Pharmacokinetics— Core 3 1 -				4
5	BP605T	Pharmaceutical Biotechnology— Theory	Pharmaceutical Biotechnology— Core		1	-	4
6	BP606T	Quality Assurance— Theory	Core	3	1	-	4
7	BP607P	Medicinal chemistry III–Practical	Practical	-	-	4	2
8	BP608P	Pharmacology III— Practical	Practical	-	-	4	2
9	BP609P	Herbal Drug Technology— Practical -		-	4	2	
		Total Credit	ts: 30				





Se	mester-VII						
1	BP701T	Instrumental Methods of Analysis–Theory	Core	3	1	-	4
2	BP702T	Industrial Pharmacy II–Theory	Core	3	1	-	4
3	BP703T	Pharmacy Practice— Theory	Core	3	1	-	4
4	BP704T	Novel Drug Delivery System–Theory	Core	3	1	-	4
5	BP705P	Instrumental Methods of Analysis–Practical	Practical	-	-	4	2
6	BP706PS	Practice School* 12		6			
		Total Credits:	24				





Sem	ester-VIII						
1	BP801T	Biostatistics and Research Methodology	Core	3	1		4
2	BP802T	Social and Preventive Pharmacy	Core	3	1	ı	4
3	BP803ET	Pharma Marketing Management	Discipline specific Elective				
4	BP804ET	Pharmaceutical Regulatory Science	Elective	3+	1+1		
5	BP805ET	Pharmacovigilance	Discipline specific Elective	3= 6	=2		
6	BP806ET	Quality Control and Standardization Of Herbals	Elective				
7	BP807ET	Computer Aided Drug Design	Elective				4+4=8
8	BP808ET	Cell and Molecular Biology	Discipline specific Elective				
9	BP809ET	Cosmetic Science	Elective	1			
10	BP810ET	Experimental Pharmacology	Elective				
11	BP811ET	Advanced Instrumentation Techniques	Elective				
12	BP812ET	Dietary Supplements and Nutraceuticals	Elective				
13	BP813PW	Project Work	Project -	-	-	12	6
	_	Total Credits:	22				
·		Programme Credits: 21	5				



Course Module





Sch	ool:	SOP
	gramme:	B. Pharm
	nch:	Semester: 1
1	Course Code	BP101 T
2	Course Title	Human Anatomy & Physiology I – Theory
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Compulsory
5	Course Objective	Upon completion of this course the student should be able to 1. Understand the mechanism of drug action and its relevance in the treatment of different diseases 2. Demonstrate isolation of different organs/tissues from the laboratory animals by simulated experiments 3. Demonstrate the various receptor actions using isolated tissue preparation 4. Appreciate correlation of pharmacology with related medical sciences
6	Course Outcomes	CO1: The students will be able to understand the structure and functions of various tissues and organs of the body. CO2: The student will be able to summarize the functioning of various body systems and their homeostasis. CO3: The student will be able to apply the knowledge of the anatomy and physiology of different body parts in explaining the working patterns of different body systems. CO4: The students will analyze the structures of various tissues and their origin to evaluate their damage and repair process. CO5: The students will evaluate the mechanisms of various processes on which the functioning of the various body organs depend and will observe the anatomical differentiation of different body parts. CO6: The students will be able to conclude about the mechanisms of various functioning of the body organs.
7	Course Description	This subject is designed to impart fundamental knowledge on the structure and functions of the various systems of the human body. It also helps in understanding both homeostatic mechanisms. The subject provides the basic knowledge required to understand the various disciplines of pharmacy.
8	Outline syllabus	
	1	





	UNIT-I
	A.Introduction to human body
	Definition and scope of anatomy and physiology, levels of structural organization
	and body systems, basic life processes, homeostasis, basic anatomical
	terminology.
	B.Cellular level of organization
	Structure and functions of cell, transport across cell membrane, cell division, cell junctions. General principles ofcell communication, intracellular signaling pathway activation by extracellular signal molecule, Forms of intracellular signaling: a) Contact-dependent b) Paracrine c) Synaptic d) Endocrine C.Tissue level of organization
	Classification of tissues, structure, location and functions of epithelial,
	muscular and nervous and connective tissues.
2	UNIT-II
	A. Integumentary system
	Structure and functions of skin
	B. Skeletal system
	Divisions of skeletal system, types of bone, salient features and functions of bones of axial and appendicular skeletal system Organization of skeletal muscle, physiology of muscle contraction, neuromuscular junction C. Joints
	Structural and functional classification, types of joints movements and its
	Articulation
3	UNIT-III
	A. Body fluids and blood
	Body fluids, composition and functions of blood, hemopoeisis, formation of hemoglobin, anemia, mechanisms of coagulation, blood grouping, Rh factors, transfusion, its significance and disorders of blood, Reticulo endothelial system.
	B. Lymphatic system
	Lymphatic organs and tissues, lymphatic vessels, lymph circulation and C. Functions of lymphatic system





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4						
	UNIT-IV					
		al nervous sys	tem:			
	-	-				
		Classification of peripheral nervous system: Structure and functions of sympathetic and parasympathetic nervous system.				
		B. Origin and functions of spinal and cranial nerves.				
	_	C .Special senses				
	_	Structure and functions of eye, ear, nose and tongue and their disorders.				
5	UNIT-V	Tunctions of C	ye, car, nose and tongue and men disorders.			
3		scular system				
		•	blood circulation, blood vessels, structure and			
	functions of	,	,,,			
	artery, vein	and capillaries	s, elements of conduction system of heart and			
	heart beat, its	-	•			
	B. Regulation	n by autonon	nic nervous system, cardiac output, cardiac			
	cycle.					
	C. Regulation	n ofblood pres	sure, pulse, electrocardiogram and disorders of			
	heart.					
Mode of	Theory					
examination						
Weightage	Continuous	Sessional	ESE			
Distribution	Mode	Exam				
	Assessment					
	10 Marks	15	75			
Text book/s*			Physiology by K. Sembulingam and P. prother's medical publishers, New Delhi.			
	•		gy in Health and Illness by Kathleen J.W. gstone, New York			
		cal basis of N o,Riverview,M	Medical Practice-Best and Tailor. Williams & II USA			
		of Medical P	hysiology- Arthur C,Guyton andJohn.E. Hall.			
		of Anatomy GA, U.S.A.	and Physiology by Tortora Grabowski.			
Other						
References						









Sch	ool:	SOP
	gramme:	B.Pharm
	inch:	Semester: 1
1	Course Code	BP102T
2	Course Title	Pharmaceutical Analysis I- Theory
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Compulsory
5	Course Objective	Upon completion of this course the student should be able to 1. Understand the mechanism of drug action and its relevance in the treatment of different diseases 2. Demonstrate isolation of different organs/tissues from the laboratory animals by simulated experiments 3. Demonstrate the various receptor actions using isolated tissue preparation 4. Appreciate correlation of pharmacology with related medical sciences
6	Course Outcomes	CO1: Students will be able to understand about the Pharmaceutical analysis and its importance in Pharmacy. CO2: Students will able to evaluate about different types of analytical techniques. CO3: Students can apply their anlaytical knowledge regarding Gravimetry. CO4: Students will able to explain difference between volumetric, quantitative and qualitative analysis. CO5: Students will be able to conclude about Electrochemical methods of analysis CO6: Students will be able to generalize about Non aqueous titrations.
7	Course Description	This course deals with the fundamentals of analytical chemistry and principles of electrochemical analysis of drug
8	Outline syllabi	IS





1	UNIT-I
	A. Pharmaceutical Analysis
	Definition and scope
	Definition and scope i) Different techniques of analysis ii) Methods of expressing concentration iii) Primary and secondary standards. iv) Preparation and standardization of various molar and normal solutions Oxalic acid, sodium hydroxide, hydrochloric acid, sodium thiosulphate, sulphuric acid, potassium permanganate and ceric ammonium sulphate. B. Errors: Sources of errors, types of errors, C. Methods of minimizing errors, accuracy, precision and significant figures.
2	UNIT-II
2	A. Acid base titration & Non aqueous titration
	Theories of acid base indicators, classification of acid base titrations and theory
	involved in titrations of strong, weak, and very weak acids and bases, neutralization curves
	B. Non aqueous titration: Solvents,
	C. acidimetry and alkalimetry titration and estimation of Sodium benzoate and Ephedrine HCl
3	UNIT-III
	A. Precipitation titrations, Complexometric titration &
	ravimetry, Diazotization
	Precipitation titrations: Mohr's method, Volhard's, Modified Volhard's, Fajans method, estimation of sodium chloride. Complexometric titration: Classification, metal ion indicators, masking and demasking reagents, estimation of Magnesium sulphate, and calcium gluconate.
	B. Gravimetry : Principle and steps involved in gravimetric analysis. Purity of the precipitate: co-precipitation and post precipitation, Estimation of barium sulphate.
	C. Define the Diazotization with their principle, methodology and
	their uses
4	UNIT-IV
	A. Redox titrations
	Concepts of oxidation and reduction
	B. Types of redox titrations (Principles and applications) Cerimetry
	C. Iodimetry, Iodometry, Bromatometry, Dichrometry, Titration with potassium
	iodate.





5	UNIT-V				
3	A. Electrochemical methods of analysis				
		Conductometry- Introduction, Conductivity cell, Conductometric titrations,			
	applications.				
	B. Potentiometry - Electrochemical cell, construction and working of reference				
			chloride electrode and calomel electrode) and		
			strodes and glass electrode), methods to determine		
			ation and applications.		
			Ilkovic equation, construction and working of		
			androtating platinum substituent on Basicity.		
			anduses of Ethanolamine, Ethylenediamine,		
	Amphetamine		·		
	electrode, applic	cations			
Mode of	Theory				
examination					
Weightage	Continuous	Sessional	ESE		
Distribution	Mode	Exam			
	Assessment				
	10 Marks	15	75		
Text book/s*	1. A.H. Be	ckett & J.B. St	tenlake's, Practical Pharmaceutical		
	Chemist	ry Vol I & II,	Stahlone Press of University of		
	London				
	2. A.I. Vog	gel, Text Book	of Quantitative Inorganic analysis		
	3. P. Gund	u Rao, Inorgar	nic Pharmaceutical Chemistry		
	4. Bentley	and Driver's T	extbook of Pharmaceutical Chemistry		
	5. John H.	Kennedy, Ana	llytical chemistry principles		
	6. Indian P	harmacopoeia			





School:		SOP			
	gramme:	B.Pharm			
	nch:	Semester: 1			
1	Course Code	BP103T			
2	Course Title	Pharmaceutics – Theory			
3	Credits	4			
4	Contact Hours (L-T-P)	3-1-0			
	Course Type	Compulsory			
5	Course Objective	Upon completion of this course the student should be able to:			
		 Know the history of profession of pharmacy 			
		 Understand the basics of different dosage forms, pharmaceutical incompatibilities andpharmaceutical calculations 			
		 Understand the professional way of handling the prescription 			
		Preparation of various conventional dosage forms			
6	Course Outcomes	Student will be able CO1: to understand about general formulation, classification of different dosage forms and various Pharmacopoeias-IP, BP, USP etc. CO2: to apply the professional way of handling the prescription, excipients used in different dosage forms, various factors affecting Posology and solubility enhancement techniques. CO3: to illustrate different methods of preparation of various semisolid dosage forms and how to calculate the dose of pediatric patients, different calculations based on the Imperial & Metric system. CO4: to distinguish between various Monophasic and biphasic liquids. CO5: to explain about different types of semisolid dosage forms like suspension, emulsion, ointments, pastes, creams etc. CO6: to predict stability problems in different dosage forms.			
7	Course Description	This course is designed to impart a fundamental knowledge on the preparatory pharmacy with arts and science of preparing the different conventional dosage forms.			





8	Outline syllab	DUS
	1	
		 UNIT-I A. Historical background and development of profession of pharmacy: History of profession of Pharmacy in India in relation to pharmacy education, industry and organization, Pharmacy as a career, Pharmacopoeias: Introduction to IP, BP, USP and Extra Pharmacopoeia. B. Dosage forms: Introduction to dosage forms, classification and definitions C. Prescription: Definition, Parts of prescription, handling of Prescription and Errors in prescription. Posology: Definition, Factors affecting posology. Pediatric dose calculations based on age, body weight and body surface area.
	2	UNIT-II A. Pharmaceutical Calculations, Powders, Liquid dosage forms Weight & Measures, Calculation involving Percentage solution etc. B. Definition of Powders, Eutectic Mixtures, Geometric Dilutions.
		C.Solubility enhancement techniques Advantages & disadvantage of liquid dosage forms.
	3	 UNIT-III A. Monophasic and Biphasic liquids Introduction of various monophasic liquids such as gargles, syrups, liniments, Eardrops etc. B. Suspensions, different types of suspension & stability problems & methods to overcome C. Emulsions, classification & different methods of preparation, stability problems & methods to overcome
	4	 A. Pharmaceutical Incompatibilities & Suppositories B. Definition and classification of different pharmaceutical incompatibilities C. Suppositories: types, methods of preparartion, types of base, Evaluation and Displacement value.





5	UNIT-V			
	A. Semisolid Dosage Forms			
		Definition, classification,		
	B.Mechanism Preparation of ointments, paste, creams, gels.			
	C.Excipients (C.Excipients used and Evaluation of semi solid dosage forms.		
Mode of	Theory			
examination				
Weightage	Continuous	Sessional	ESE	
Distribution	Mode	Exam		
	Assessment			
	10 Marks	15	75	
Text book/s*	1. H.C. Ans	el et al., Phai	maceutical Dosage Form and I	Orug Delivery
			iams and Walkins, New Delhi.	
			and Gunn's-Dispensing for P	harmaceutical
		CBS publisher		
	·	-	eutics, The Science & Dosage	Form Design
		Livingstone, E	_	2 cm 2 cm,
	4. Indian ph		amourgn.	
	-	-		
	_	5. British pharmacopoeia.		
		<u>-</u>	Practice of Industrial Pharmacy, I sy of Michigan.	Lea & Febiger
	·		Remington. The Science and	Practice of
			'illiams, New Delhi.	Tractice of
			Gunn's. Tutorial Pharmacy, CBS	Publications
	New Delh	-	Guilli S. Tutoriai Tharmacy, CDS	or domeations,
	9. E.A. Rav	wlins, Bentley	y's Text Book of Pharmaceu	itics, English
	Language	Book Society	, Elsevier Health Sciences, USA.	
	10. Isaac Gh	ebre Sellassie	e: Pharmaceutical Pelletization	Technology,
	Marcel De	ekker, INC, Ne	ew York.	
	11. Dilip M	. Parikh: I	Handbook of Pharmaceutical	Granulation
	Technolog	gy, Marcel Del	kker, INC, New York.	
	12.	Francoise	Nieloud and	Gilberte
Other				
References				





Sc	chool:	SOP			
	ogramme:	B.Pharm			
	ranch:	Semester: 1			
1	Course	BP104T			
	Code				
2	Course	Pharmaceutical Inorganic Chemistry -Theory			
	Title				
3	Credits	4			
4	Contact	3-1-0			
	Hours				
	(L-T-P)				
	Course	Compulsory			
	Type				
5	Course	Upon completion of this course the student should be able to			
	Objective	1. Understand the mechanism of drug action and its relevance in the treatment of			
		different diseases			
		2. Demonstrate isolation of different organs/tissues from the laboratory animals by simulated experiments			
		3. Demonstrate the various receptor actions using isolated tissue preparation			
		4. Appreciate correlation of pharmacology with related medical sciences			
6	Course	CO1 Students shall be able to illustrate various sources of impurities and their			
	Outcomes	control in inorganic drugs and pharmaceuticals.			
		CO2 Students shall be able to explain concept of acids, bases and buffers			
		andmethods of calculating and adjusting isotonicity.			
		CO3 Students shall be able to discuss major intra and extracellular ions,			
		replacement therapy and physiological acid-base balance.			
		CO4 Students shall be able to evaluate various inorganic compounds, like			
		gastrointestinal agents, dentalproducts and antimicrobials.			
		CO5 Students will be able to apply knowledge about radiopharmaceuticals, their			
		handling, hazards and uses.			
		CO6 Students shall be able to understand importance of inorganic compounds			
7	Course	which can be used as useful medicinal compounds.			
'	Course Description	This subject deals with the monographs of inorganic drugs and pharmaceuticals.			
8	Outline sylla				
0	1	Oub			
	•	UNIT-I			
		A. Impurities in pharmaceutical substances			
		History of Pharmacopoeia,			
		B.Sources and types of impurities, principle involved in the limit test for			
		Chloride, Sulphate, Iron, Arsenic, Lead C. Heavy metals, modified limit test for			
		Chloride and Sulphate			





	2	UNIT-II					
		A. Acids, Bases and Dental products	Buffers Major ex	tra and intracellular electrolytes,			
		B . Buffer equations and buffer capacity in general, buffers in pharmaceutical systems, preparation, stability, buffered isotonic solutions, measurements of tonicity, calculations and methods of adjusting tonicity.					
		therapy: Sodium chl Rehydration Salt Dentifrices, role of flo	oride*, Potassium uoride in the treatm	ons, Electrolytes used in the replacement chloride, Calcium gluconate* and Oral ent of dental caries, Desensitizing agents, d Zinc eugenol cement.			
	3	UNIT-III		-			
		A.Gastrointestinal a	_				
		Acidifiers: Ammoniu					
				, ombinations of antacids, Sodium 40			
		Bicarbonate*, Alumin					
				thartics: Magnesium sulphate, Sodium			
		orthophosphate, Kaolin and Bentonite Antimicrobials:					
		-	•	m permanganate, Boric acid, Hydrogen			
peroxide*, Chlorinated lime*, Iodine and its preparation							
	4	UNIT-IV	nnounds				
A.Miscellaneous compounds Expectorants: Potassium iodide, Ammonium chloride*.			nium chloride*				
		Emetics: Copper sulp					
		B.Sodium potassium					
Haematinics: Ferrous sulphate*,							
		C. Ferrous gluconate Poison and Antidote: Sodium thiosulpha					
				c Sulphate, Potash Alum			
	5	UNIT-V					
		A. Radiopharmaceu					
		Radio activity, Measu					
		B.Properties of α , β , γ		-			
		C. study of radio isotopes - Sodium iodide I131, Storage conditions, precautic & pharmaceutical application of radioactive substances.					
	Mode of	Theory					
	examinatio						
	n Weightage	C 4' M 1 Consistent EGE					
	Distributio	Continuous Mode Assessment	Sessional Exam	ESE			
	n	10 Marks	15	75			
		10 Marks	13	13			





Text book/s*	1.	A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London, 4 th edition.
	2.	A.I. Vogel, Text Book of Quantitative Inorganic analysis
	3.	P. Gundu Rao, Inorganic Pharmaceutical Chemistry, 3 rd Edition
	4.	M.L Schroff, Inorganic Pharmaceutical Chemistry
	5.	Bentley and Driver's Textbook of Pharmaceutical Chemistry
	6.	Anand & Chatwal, Inorganic Pharmaceutical Chemistry





School:		SOP			
Prog	gramme:	B.Pharm			
Bra	nch:	Semester: 1			
1	Course Code	BP105T			
2	Course Title	Communications Theory Theory			
3	Credits	2			
4	Contact Hours (L-T-P)	2-0-0			
	Course Type	Compulsory			
5	Course Objective	Upon completion of the course the student shall be able to			
	3	Understand the behavioral needs for a Pharmacist to function effectively in theareas of pharmaceutical operation			
		Communicate effectively (Verbal and Non Verbal)			
		Effectively manage the team as a team player			
		Develop interview skills			
		Develop Leadership qualities and essentials			
6	Course Outcomes	CO1: students will be able to Understand the behavioral needs for a Pharmacist to function effectively in theareas of pharmaceutical operation			
		CO2: students will be able to plan how to Communicate effectively (Verbal and Non Verbal)			
		CO3: students will be able to apply Effectively to manage the team as a team player			
		CO4: students will be able to Develop interview skills			
		CO5: students will be able to generalize about Leadership qualities and essentials CO6: students will be able to plan about Group discussion			
7	Course Description	This course will prepare the young pharmacy student to interact effectively with doctors, nurses, dentists, physiotherapists and other health workers. At the end of this course the student will get the soft skills set to work cohesively with the team as a team player and will add value to the pharmaceutical business.			
8	Outline syllabu	is			





1	UNIT-I
	A. Communication Skills: Introduction, Definition, The Importance of Communication, The Communication Process – Source, Message, Encoding, Channel, Decoding, Receiver, Feedback, Context
	B. Barriers to communication: Physiological Barriers, Physical Barriers, Cultural Barriers, Language Barriers, Gender Barriers, Interpersonal Barriers, Psychological Barriers, Emotional barriers
	C. Perspectives in Communication: Introduction, Visual Perception, Language, Other factors affecting our perspective - Past Experiences, Prejudices, Feelings, Environment
2	UNIT-II
	A. Elements of Communication: Introduction, Face to Face Communication - Tone of Voice, Body Language (Non-verbal communication), Verbal Communication, Physical Communication
	B. Communication Styles: Introduction, The Communication Styles Matrix with example for each –
	C. Direct Communication Style, Spirited Communication Style, Systematic Communication Style, Considerate Communication Style
3	UNIT-III
	A. Basic Listening Skills: Introduction, Self-Awareness, Active Listening, Becoming anActive Listener, Listening in Difficult Situations
	B. Effective Written Communication: Introduction, When and When Not to Use Written Communication - Complexity of the Topic, Amount of Discussion' Required, Shades of Meaning, Formal Communication
	C. Writing Effectively: Subject Lines, Put the Main Point First, Know Your Audience, Organization of the Message





4	UNIT-IV		
	A. an inte		lls: Purpose of an interview, Do's and Don'ts of
		Giving Preser tation,	ntations: Dealing with Fears, Planning your
			ourPresentation, Delivering Your ques of Delivery
5 UNIT-V			
	Α.	Group Discus	ssion:
	В.	Introduction,	Communication skills in group discussion,
	C.	Do's and Don	ts of group discussion
Mode of	Theory		
examination			
Weightage	Continuous	Sessional	ESE
Distribution	Mode	Exam	
	Assessment		
	05 Marks	10	35





Text book/	/s* 1.	Basic communication skills for Technology,
	1.	Andreja. J. Ruther Ford, 2 nd Edition, Pearson
		Education, 2011
	2.	Communication skills, Sanjay Kumar, Pushpalata, 1 st Edition,
	2.	Oxford Press, 2011
	3.	Organizational Behaviour, Stephen .P. Robbins, 1 st Edition, Pearson, 2013
	4.	Brilliant- Communication skills, Gill Hasson, 1 st Edition, Pearson Life, 2011
	5.	The Ace of Soft Skills: Attitude, Communication and
		Etiquette for success, GopalaSwamy Ramesh,
		5 th Edition, Pearson, 2013
	6.	Developing your influencing skills, Deborah Dalley,
		Lois Burton, Margaret, Greenhall, 1st Edition
		Universe of Learning LTD, 2010
	7.	Communication skills for professionals, Konar nira, 2 nd Edition, New arrivals –
		PHI, 2011
	8.	Personality development and soft skills, Barun K
		Mitra, 1stEdition, Oxford Press,2011
	9.	Soft skill for everyone, Butter Field, 1st Edition,
		Cengage Learning india pvt.ltd,2011
	10.	Soft skills and professional communication, Francis
		Peters SJ, 1st Edition, Mc GrawHill Education, 2011
	11.	Effective communication, John Adair, 4 th Edition, Pan Mac Millan,2009
	12.	Bringing out the best in people, Aubrey Daniels, 2 nd Edition, Mc Graw Hill, 1999





School:		SOP		
Programme:		B.Pharma		
	anch:	Semester: 1		
1	Course	BP106 RBT		
	Code			
2	Course	Remedial biology - Theory		
	Title			
3	Credits	4		
4	Contact	2-0-0		
	Hours			
	(L-T-P)			
	Course	Compulsory		
	Type			
5	Course	Upon completion of the course, the student shall be able to		
	Objective	-Know the classification and salient features of five kingdoms of life		
		-Understand the basic components of anatomy & physiology of plant		
		-Know understand the basic components of anatomy & physiology animal		
		with special reference to human		
6	Course	CO1: Students will be able to generalize would acquire knowledge of five		
	Outcomes	kingdoms of life, morphology and anatomy of flowering plants, anatomy and		
		physiology of plants and humans and various plant growth regulators.		
		CO2: Students would be able to understand the anatomy and physiology of		
		plants and humans.		
		prants and namans.		
		CO3: Students will be able to apply the knowledge of the anatomy and		
		physiology of different body parts in explaining the working patterns of		
		different body systems.		
		CO4: The students will analyze the structures of various tissues and their		
		origin.		
		CO5: The students would evaluate the mechanisms of various processes on		
		which the functioning of the various body organs and plants depend.		
		Moreover, will observe the anatomical differentiation of different body parts of human.		
		CO6: The students will be able to predict about the mechanisms of various		
		body organs and plants		
		as a figure and brance		
7	Course	To learn and understand the components of living world, structure and		
	Description	functional system of plant and animal kingdom. Scope: To learn and		
	_	understand the components of living world, structure and functional system of		
	0 11	plant and animal kingdom.		
8	Outline syllab	ous		





	1	UNIT-I	
		A.Living world:	
		Definition and characters of living organisms Diversity in the living world Binomial nomenclature Five kingdoms of life and basis of classification. Salient features of Monera, Potista, Fungi, Animalia and Plantae, Virus, B.Morphology of Flowering plants Morphology of different parts of flowering plants – Root, stem, inflorescence, flower, leaf, fruit, seed. C.General Anatomy of Root, stem, leaf of monocotyledons & Dicotylidones	
	2	UNIT-II	
	2	A.Body fluids and circulation	
		Coposition of blood, blood groups, coagulation of blood Composition and functions of lymph Human circulatory system Structure of human heart and blood vessels Cardiac cycle, cardiac output and ECG B.Digestion and Absorption	
		Human alimentary canal and digestive glands Role of digestive enzymes	
		Digestion, absorption and assimilation of digested food	
		C.Breathing and respiration	
		Human respiratory system	
		Mechanism of breathing and its regulation	
		Exchange of gases, transport of gases and regulation of respiration	
		Respiratory volumes	
	3	UNIT-III	
		A.Excretory products and their elimination	
		Modes of excretion	
		Human excretory system- structure and function Urine formation	
		Rennin angiotensin system	
		B.Neuralcontrol and coordination	
		Definition and classification of nervous system Structure of a neuron Generation and conduction of nerve impulse Structure of brain and spinal cord Functions of cerebrum, cerebellum, hypothalamus and medulla oblongata C.Chemical coordination and regulation	
		Endocrine glands and their secretions Functions of hormones secreted by	
		endocrine glands	
		Human reproduction	
		Parts of female reproductive system Parts of male reproductive system	
		Spermatogenesis and Oogenesis	
		Menstrual cycle	





	4	UNIT-IV		
A.Plants and mineral nutrition:				
	Essential mineral, macro and micronutrients			nutrients
		B. Nitrogen metabolism, Nitrogen cycle, biological nitrogen fixation		
		C.Photosynthesis		
		Autotrophic nutrition, photosynthesis, Photosynthetic pigments,		
}	<u>-</u>	affecting photosynthesis.		
	5	UNIT-V A.Plant Respiration: Respiration, glycolysis, fermentation (anaerobic).		
		B.Plant growth and development		
		_	_	ndition of growth,Introduction to plant
		growth regulators	piani growin, co	nation of growin, introduction to plant
		C.Cell - The unit	of life	
				cell organelles.Cell division
		D.Tissues		8
		Definition, types of	f tissues, location	n and functions.
	Mode of	Theory		
	examination			
	Weightage	Continuous	Sessional	ESE
	Distribution	Mode	Exam	
		Assessment		
		05 Marks	10	35
		a.Text book of Biol	ogy by S. B. Gol	khale
	book/s*	A Text book of B	iology by Dr. Th	ulajappa and Dr. Seetaram.
	Other	a. A Text book of		
	References	b. A Text book of Biology by Naidu and Murthy		
		c. Botany for Degree students By A.C.Dutta.		
		d. Outlines of Zoology by M. Ekambaranatha ayyer and T. N.		
		Ananthakrishnan.		
		e. A manual for pharmaceutical biology practical by S.B. Gokhale and C. K.		
		Kokate		





School:		SOP
Pro	gramme:	B.Pharm
Branch:		Semester: 1
1	Course	BP106 RMT
	Code	
2	Course	Remedial Mathematics - Theory
	Title	
3	Credits	2
4	Contact	2-0-0
	Hours	
	(L-T-P)	
	Course	Compulsory
	Type	
5	Course	Upon completion of the course, the student shall be able to
	Objective	-Know the classification and salient features of five kingdoms of life
		-Understand the basic components of anatomy & physiology of plant
		-Know understand the basic components of anatomy & physiology animal with
special reference to human		1
		CO1: Students would acquire knowledge of partial fraction and logarithms
	Outcome	CO2: Students would be able to understand about matrices and determinants
	S	CO3: Students will be able to apply the knowledge of Differentiation.
		CO4: Students will be able to analyze the knowledge of Integration
		CO5: Students will be able to interpret differential equations.
7	Cook: Students will be able to apply mathematical techniques.	
'	Course	This is an introductory course in mathematics. This subject deals with the
	Descripti	introduction to Partial fraction, Logarithm, matrices and Determinant,
	on	Analytical geometry, Calculus, differential equation and Laplace transform
8	Outline syllabus	





		UNIT – I		
		A.Partial fraction		
		Introduction, Polynomial, Rational fractions, Proper and Improper		
		fractions, Partial fraction, Resolving into Partial fraction,		
		Application of Partial Fraction in Chemical Kinetics and		
		Pharmacokinetics		
		B.Logarithms		
		Introduction, Definition, Theorems/Properties of logarithms,		
		Common logarithms, Characteristic and Mantissa, worked		
		examples, application of logarithm to solve pharmaceutical		
		problems.		
		C Franchica		
		C.Function:		
		Real Valued function, Classification of real valued functions,		
		Limits and continuity: Introduction, Limit of a function, Definition of limit of a function		
2		UNIT -II		
		A.Matrices and Determinant:		
		Introduction matrices, Types of matrices, Operation on matrices,		
		Transpose of a matrix, Matrix Multiplication, Determinants,		
		Properties of determinants,		
		B.Product of determinants, Minors and co-Factors, Adjoint or		
		adjugate of a square matrix, Singular and non-singular matrices,		
		Inverse of a matrix,		
		C.Solution of system of linear of equations using matrix method, Cramer's rule, Characteristic equation and roots of a square		
		matrix, Cayley—Hamilton theorem, Application of Matrices in		
		solving Pharmacokinetic equations		
3				
		UNIT - III A.Calculus		
		Differentiation : Introductions, Derivative of a function,		
		Derivative of a constant, B. Derivative of a product of a constant and a function, Derivative		
		of the sum or difference of two functions, Derivative of the product		
		of two functions (product formula),		
		C. Derivative of the quotient of two functions (Quotient formula) –		
	Without Proof,			
		Derivative of x^n w.r.tx, where n is any rational number,		
		Derivative of e^x , Derivative of $\log_e x$, Derivative of a^x , Derivative		
		of trigonometric functions from first principles (without Proof),		
		Successive Differentiation, Conditions for a function to be a		
		maximum or a minimum at a point. Application		





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4					
	UNIT - IV A.Analytical Geometry Introduction: Signs of the Coordinates, Distance formula, B.Straight Line: Slope or gradient of a straight line, Conditions for parallelism and perpendicularity of two lines, Slope of a line joining two points, Slope – intercept form of a straight line C.Integration: Introduction, Definition, Standard formulae, Rules of integration,				
5	Method of substitution, Method of Partial fractions, Integration by parts, definite integrals, application				
Mode of	A. Differential Equations: Some basic definitions, Order and degree, Equations in separable form, Homogeneous equations, Linear Differential equations, Exact equations, Application in solving Pharmacokinetic equations B. Laplace Transform: Introduction, Definition, Properties of Laplace transform, Laplace Transforms of elementary functions, Inverse Laplace transforms, Laplace transform of derivatives, C.Application to solve Linear differential equations, Application in solving Chemical kinetics and Pharmacokinetics equations Theory				
examinat ion	Theory				
Weightag e	Continuous Mode Assessment	Sessional Exam	ESE		
Distributi on	5 Marks	10	35		
Text book/s*	 Differential Calculus by Shanthinarayan Pharmaceutical Mathematics with application to Pharmacy by Panchaksharappa Gowda D.H. Integral Calculus by Shanthinarayan Higher Engineering Mathematics by Dr.B.S.Grewal 				
Other Referenc					





School:		SOP			
Programme:		B.Pharm			
	anch:	Semester: 1			
1	Course	BP107P			
	Code				
2	Course	Human Anatomy and Physiology- Practical			
	Title				
3	Credits	2			
4	Contact	0-0-4			
	Hours				
	(L-T-P)				
	Course	Compulsory			
_	Type				
5	Course	1. To understand how to handle the microscope in Human Anatomy &			
	Objective	Physiology lab			
		2. To calculate Hb content, WBC & RBC count and Erythrocyte			
		3. To identify axial and skeletal bones of Human skeleton			
		4. To learn and practice how to record Blood Pressure of given subject			
6	6 Course CO1: Student will be able to Understand how to handle the microscope in				
	Outcomes	Anatomy & Physiology lab			
		CO2: Student will be able to analyze Hb content, WBC & RBC count and			
		Erythrocyte sedimentation rate			
		CO3: Student will be able to Identify axial and skeletal bones of Human skeleton			
		CO4: Student will be able to interpret Blood Pressure of given subject			
		CO5: Student will be able to skeletal bones of Human skeleton			
		CO6: Students shall be able to understand and know body parts			
7	Course	Practical physiology is complimentary to the theoretical discussions in physiology.			
	Description	Practicals allow the verification of physiological processes discussed in theory classes			
		through experiments on living tissue, intact animals or normal human beings. This is			
0	Outling avillat	helpful for developing an insight on the subject.			
8	Outline sylla	bus UNIT-I			
	1	a). Study of compound microscope			
		b). Microscopic study of epithelial and connective tissue			
		c). Microscopic study of muscular and nervous tissue			
	2	UNIT-II			
	-	a). Identification of axial bones			
		b). Identification of appendicular bones			





4	a). Introduction to hemocytometry and enumeration of white blood cell (WBC) count b). Enumeration of total red blood corpuscles (RBC) count c). Determination of bleeding time 10. Determination of clotting time UNIT-IV a). Determination of blood group		
	b). Estimation of hemoglobin content c). Determination of erythrocyte sedimentation rate (ESR)		
5 UNIT-V a). Determination of heart rate and pulse rate b). Recording of blood pressure			
Mode of examinatio			
Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE
	05	10	35
Text book/s*	1. Essentials of Medical Physiologyy K.Sembulingam and P.Sembulingam. Jaype brothers medical publishers, NewDelhi. 2.Anatomy and Physiology in Health and Illness by Kathleen W.Wilson ,ChurchilLivingstone,NewYork		
Other References		of Medical Practic	eeBestandTailor.Williams&Wilkins





School:		SOP		
Programme:		B.Pharm		
	nch:	Semester: 1		
1	Course Code	BP108P		
2	Course Title	Pharmaceutical Analysis – Practical		
3	Credits	2		
4	Contact Hours (L-T-P)	0-0-4		
5	Course Type Course	Compulsory Upon completion of the course the student shall be ship to		
3	Objective	Upon completion of the course, the student shall be able to -Know the classification and salient features of five kingdoms of life -Understand the basic components of anatomy & physiology of plant -Know understand the basic components of anatomy & physiology animal with special reference to human		
6	Course Outcomes	with special reference to human CO1: Upon completion of course student shall be able to understand about limits of impurities in a particular drug and to perform limit test to identify and determine the impurities in pharmaceuticals. CO2: Students shall be able to perform standardization and analyze given sample strength of drug or pharmaceuticals. CO3: Students shall be able to know the purity testing of drugs and pharmaceuticals. They can apply these strength tests to analyze and evaluate the sample. CO4: Students shall be able to understand about electrochemical analysis for pharmaceutical sample. CO5: Students shall be able to perform standardization and analyze given sample strength of drug or pharmaceuticals.		
		CO6: Students shall be able toapply purity testing of pharmaceuticals. Deals with the fundamentals of analytical chemistry and principles of		
′	Description	electrochemical analysis of drugs		
8	Outline syllab	· ·		
	1	UNIT-I a). Limit test for Chlorides and Sulphates b). Modified limit test for Chlorides and Sulphates c). Limit test for Iron d). Limit test for Heavy metals e). Limit test for Lead Limit test for Arsenic		





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2	UNIT-II		
	a). Sodium hy	droxide	
	b). Sulphuric a	acid	
	c). Sodium thiosulfate		
	d). Potassium permanganate		
	e). Ceric amm	onium sulphat	te
3	UNIT-III		
	a). Ammoniun	n chloride by a	acid base titration
	b). Sodium Ch	nloride by prec	eipitation titration
4	UNIT-IV		
	Conductometr	ric titration of	strong acid against strong base
5	UNIT-V		
	a). Sodium hy	droxide	
	b). Sulphuric a		
	c). Sodium thi	osulfate	
Mode of	Theory/Jury/P	ractical/Viva	
examination			
Weightage	Continuous	Sessional	ESE
Distribution	Mode	Exam	
	Assessment		
	05	10	35
Text book/s*	Practical hu	ıman anatom	y and physiology.by S.R.Kale and
	R.R.Kale.		
	A Manual	of pharmac	eutical biology practical by S.B.Gokhale,
		and S.P.Shriwa	•
			l according to National core curriculum
			_
	.Biology foru	iii oi Karnatai	ka. ProfM.J.H.Shafi





Sc	chool:	SOP
Pr	ogramme	B.Pharm
:		
Bı	ranch:	Semester: 1
1	Course	BP109P
	Code	
2	Course	Pharmaceutics –I Practical
	Title	
3	Credits	2
4	Contact	0-0-4
	Hours	
	(L-T-P)	C1
	Course Type	Compulsory
5	Course	This course will impart basic knowledge in the area of pharmaceutics and
	Objectiv	formulation of different pharmaceutical dosage forms. The students will get
	e	hands-on training in the preparation of such dosage forms in the laboratory.
6	Course	CO1: Upon completion of course student shall be able to understand about
	Outcome	different methods of preparation of various monophasic and biphasic liquid
	S	dosage forms.
		CO2: Students shall be able to apply specific types of excipients used in
		preparation of semisolid dosage forms.
		CO3: Students shall be able to interpret different types of pharmaceutical dosage
		forms like
		syrups, elixirs, solutions, paints, gargles, mouthwashes, suspensions, emulsions, powd
		ers, ointments, pastes etc.
		CO4: Students shall be able to differentiate between different methods of preparation of pharmaceutical dosage forms.
		CO5: Students shall be able to interpret specific types of excipients used in
		preparation of semisolid dosage forms.
		CO6: Students shall be able to understand Different types of pharmaceutical
		formulations for the human use.
7	Course	This course is designed to impart knowledge on preparatory pharmacy and
	Descripti	
	on	professional way of preparing various dosage forms such as monophasic liquids,
		biphasic liquids,
	0 41'	semisolid dosage forms etc.
8	Outline syl	Habus





	, , , , , , , , , , , , , , , , , , , ,
1	UNIT-I
	A. Syrups
	Syrup IP'66
	Compound syrup of Ferrous Phosphate BPC'68
	B. Elixirs
	Piperazine citrate elixir
	Paracetamol pediatric elixir
2	TO LICE II
2	UNIT-II A. Linctus
	Terpin Hydrate Linctus IP'66
	Iodine Throat Paint (Mandles Paint)
	B. Solutions
	Strong solution of ammonium acetate
	Cresol with soap solution Lugol's solution
	Lugoi s solution
3	UNIT-III
3	A. Suspensions
	Calamine lotion
	Magnesium Hydroxide mixture
	Aluminimum Hydroxide gel
	B. Emulsions
	Turpentine Liniment
	Liquid paraffin emulsion
4	UNIT-IV
•	A. Powders and Granules
	ORS powder (WHO)
	Effervescent granules
	Dusting powder
	Divided powders
	B. Suppositories
	Glycero gelatin suppository
	Coca butter suppository
	Zinc Oxide suppository





	5	UNIT-V	UNIT-V		
		A. Semisolids			
		Sulphur ointment			
		Non staining-iodine o	intment with methyl	salicylate	
		Carbopal gel			
		B. Gargles and M	l outhwashes		
		Iodine gargle			
		Chlorhexidine mouth	wash		
	Mode of	Practical/Viva			
	examinat				
	ion				
	Weighta	Continuous Mode	Sessional Exam	ESE	
	ge	Assessment			
	Distribut	05Marks	10	35	
	ion				
	Text	H.C. Ansel et al., F	Pharmaceutical Dosa	ge Form and Drug Delivery	
	book/s*	System, Lippincott W	illiams and Walkins,	New Delhi.	
		Carter S.J., Cooper an	nd Gunn's-Dispensing	g for Pharmaceutical Students,	
		CBS publishers, New	Delhi.		
		M.E. Aulton, Pharm	naceutics, The Scien	nce& Dosage Form Design,	
		Churchill Livingstone	e, Edinburgh.		
1. Indian pharmacopoeia.					
		2. British pharmacopo	peia.		
		- ·			
	Other				
	Referenc				
	es				





School:		SOP	
Program	me:	B.Pharm	
Branch:		Semester: I	
1	Course Code	BP110P	
2	Course Title	Pharmaceutical inorganic chemistry- Practical	
3	Credits	2	
4	Contact Hours (L-T-P)	0-0-4	
	Course Type	Compulsory	
5	Course Objective	The students will know the sources of impurities and methods to determine the impurities in inorganic drugs and pharmaceuticals.	
6	Course Outcomes	CO1: Student shall be able to understand about the limits of impurities in a particular drug and to perform limit test to identify and determine the impurities in inorganic drugs and pharmaceuticals. CO2: Students shall be able to perform identification test and analyze given sample of drug or pharmaceuticals. CO3: Students shall be able to interpret the purity testing of inorganic drugs and pharmaceuticals. They can apply these purity tests to analyze and evaluate the sample. CO4: Students shall be able to apply methods of preparation of drugs and pharmaceuticals. CO5: Students shall be able to apply methods of preparation of various inorganic drugs. CO6: Students shall be able to evaluate the presence of inorganic compounds in biological fluids.	
7	Course Description	Limit test for non- toxic and toxic impurities, identification test for some Drugs, preparation of some drugs and purity test for some inorganic drugs and pharmaceuticals.	
8	Outline syllabus		
	1	UNIT-I a). Limit test for Chlorides and Sulphates b). Modified limit test for Chlorides and Sulphates c). Limit test for Iron d). Limit test for Heavy metals e). Limit test for Lead Limit test for Arsenic	





2 UNIT-II a). Magnesium hydroxide b). Ferrous sulphate c). Sodium bicarbonate d). Calcium gluconate e). Copper sulphate 3 UNIT-II a). Swelling power of Bentonite b). Neutralizing capacity of aluminum hydroxide gel c). Determination of potassium iodate and iodine in potassium Iodide 4 UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate 6 Practical/Viva 8 Sessional Exam Mode Assessment 05 Marks 10 35 Text book/s* Fextical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum Biology forum of Karnataka. ProfM.J.H.Shafi					www.siaida.ac.iii
b). Ferrous sulphate c). Sodium bicarbonate d). Calcium gluconate e). Copper sulphate 3 UNIT-III a). Swelling power of Bentonite b). Neutralizing capacity of aluminum hydroxide gel c). Determination of potassium iodate and iodine in potassium lodide 4 UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode examination Weightage Distribution Weightage Distribution Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi		2	UNIT-II		
c). Sodium bicarbonate d). Calcium gluconate e). Copper sulphate 3 UNIT-III a). Swelling power of Bentonite b). Neutralizing capacity of aluminum hydroxide gel c). Determination of potassium iodate and iodine in potassium Iodide 4 UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode examination Weightage Distribution Weightage Distribution Text book/s* Practical/Viva Sessional Exam ESE Mode Assessment 05 Marks 10 35 Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R. Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi			a). Magnesium hydr	oxide	
d). Calcium gluconate e). Copper sulphate UNIT-III a). Swelling power of Bentonite b). Neutralizing capacity of aluminum hydroxide gel c). Determination of potassium iodate and iodine in potassium Iodide UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode cxamination Weightage Distribution Weightage Distribution Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi			b). Ferrous sulphate		
e). Copper sulphate 3 UNIT-III a). Swelling power of Bentonite b). Neutralizing capacity of aluminum hydroxide gel c). Determination of potassium iodate and iodine in potassium Iodide 4 UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode camination Weightage Distribution Weightage Distribution Text book/s* Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi					
e). Copper sulphate 3 UNIT-III a). Swelling power of Bentonite b). Neutralizing capacity of aluminum hydroxide gel c). Determination of potassium iodate and iodine in potassium Iodide 4 UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode of examination Weightage Distribution Weightage Distribution Text book/s* Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi			d). Calcium glucona	ite	
UNIT-III a). Swelling power of Bentonite b). Neutralizing capacity of aluminum hydroxide gel c). Determination of potassium iodate and iodine in potassium Iodide 4			e). Copper sulphate		
b). Neutralizing capacity of aluminum hydroxide gel c). Determination of potassium iodate and iodine in potassium Iodide 4 UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode examination Weightage Distribution Wode Assessment 05 Marks 10 35 Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi		3			
c). Determination of potassium iodate and iodine in potassium Iodide 4 UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode of examination Weightage Distribution Mode Assessment 05 Marks 10 35 Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum . Biology forum of Karnataka. ProfM.J.H.Shafi			a). Swelling power of	of Bentonite	
4 UNIT-IV a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode examination Weightage Distribution Wode Assessment 05 Marks 10 35 Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum . Biology forum of Karnataka. ProfM.J.H.Shafi			b). Neutralizing cap	acity of aluminum	hydroxide gel
a). Boric acid b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode of examination Weightage Distribution Wode Assessment 05 Marks 10 35 Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi			c). Determination of	f potassium iodate	and iodine in potassium Iodide
b). Potash alum c). Ferrous sulphate 5 UNIT-V a). Ferrous sulphate b). Sodium bicarbonate Mode of examination Weightage Distribution Mode Assessment 05 Marks 10 35 Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi		4			
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a). Ferrous sulphate b). Sodium bicarbonate Mode of examination Weightage Distribution Mode Assessment 05 Marks 10 35 Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi					
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Examination Weightage Continuous Sessional Exam ESE			b). Sodium bicarbor	nate	
Examination Weightage Continuous Sessional Exam ESE					
Examination Weightage Continuous Sessional Exam ESE					
Weightage Distribution Continuous Mode Assessment O5 Marks Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi Other			Practical/Viva		
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Assessment 05 Marks 10 35 Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi				Sessional Exam	ESE
Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi Other		Distribution	Mode		
Text book/s* Practical human anatomy and physiology. by S.R.Kale and R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi Other					
R.R.Kale. A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi Other					35
A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi Other		Text book/s*		anatomy and p	hysiology. by S.R.Kale and
C.K.Kokate and S.P.Shriwastava. Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi Other				1 1 1	1 CDC111
Biology practical manual according to National core curriculum .Biology forum of Karnataka. ProfM.J.H.Shafi Other			-		ology practical by S.B.Gokhale,
Biology forum of Karnataka. ProfM.J.H.Shafi Other					
Other			0, 1		
			.Biology forum of	Karnataka. ProfM.	J.H.Shafi
References		Other			
		References			





School:		SOP	
Program	me:	B.Pharm	
Branch:		Semester: 1	
1	Course Code	BP112 RBP	
2	Course Title	Remedial biology Practical	
3	Credits	2	
4	Contact Hours (L-T-P)	0-0-2	
	Course Type	Compulsory	
5	Course	1. To understand how to handle the microscope in lab.	
	Objective	2. To identify axial and skeletal bones of Human skeleton	
		3. To learn and practice how to record Blood Pressure of given subject.	
		4. To Study morphology and microscopy of Stem, Root, Leaf, seed, fruit, flower and their modifications.5. Identification of blood group.	
6	Course Outcomes	CO1: Student will be able to understand how to handle the microscope in lab.	
		CO2: Student will be able to Identify axial and skeletal bones of Human skeleton	
		CO3: Student will be able to Record Blood Pressure of given subject.	
		CO4: Student will be able to understand study Morphological and histological characteristics of root, Stem, Leaf, Seed, Fruit and Flower.	
		CO5: Student will be able to interpret the blood group of subject. CO6: Students will be able to apply practical knowledge of identification of Bones.	
7	Course Description	Practical is complimentary to the theoretical discussions remedial biology and allow the verification of physiological processes discussed in theory classes through experiments on living tissue, intact animals or normal human beings and plants. This is helpful for developing an insight on the subject.	
8	Outline syllabus	· ·	
	1	UNIT-I a). Study of compound microscope b). Microscopic study of leaves and flowers c). Microscopic study of roots and stem	





	2	UNIT-II		
		a). Identification of	avial hones	
		b). Identification of		20
		b). Identification of	appendicular bone	
	3	UNIT-III		
		a). Determination of	f blood group	
		b). Estimation of her		
	4	UNIT-IV		
		Determination of he	art rate and pulse	rate
		I D IIT II		
	5	UNIT-V		
		Recording of blood	pressure	
	Mode of	Practical/Viva		
	examination	Tractical Viva		
	Weightage	Continuous	Sessional Exam	ESE
	Distribution	Mode		
		Assessment		
		05 Marks	05	15
	Text book/s*	Practical human	anatomy and p	physiology. by S.R.Kale and
		R.R.Kale.		
		A Manual of p	oharmaceutical b	piology practical by S.B.Gokhale,
		C.K.Kokate and S.	P.Shriwastava.	
		Biology practical	manual accordin	g to National core curriculum .
		Biology forum of k		-
		.M.J.H.Shafi		
	Other			
	References			
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School:		SOP
Progra		B.Pharm
Branch		Semester: 2
	Course Code	BP 201T
	Course Title	Human Antomy & Physiology-II
	Credits	4
1	Contact	3-1-0
1	Hours	
_ `	L-T-P)	
	Course Type	Compulsory
1	Course Objective	 Explain the gross morphology, structure and functions of various organs of the human body. Describe the various homeostatic mechanisms and their imbalances.
		 Identify the various tissues and organs of different systems of human body. Perform the hematological tests like blood cell counts, haemoglobin estimation, bleeding/clotting time etc and also record blood pressure, heart rate, pulse and respiratory volume. Appreciate coordinated working pattern of different organs of each system Appreciate the interlinked mechanisms in the maintenance of normal functioning (homeostasis) of human body.
	Course Outcomes	CO1: The students will understand the structure and functions of various systems and organs of the body. Also about increase the understanding about genes and genetics. CO2: The student will be able to summarize the functioning of various body systems and their homeostasis. CO3: The student will be able to apply the knowledge of the functioning of various body systems and the structures of the organs involved in it. CO4: The students will analyze the correlation of various body systems and how they result in particular kind of functions. CO5: The students would evaluate the processes like respiration, Excretion, digestion hormone release and reproduction by understand their mechanisms. CO6: The students would evaluate the processes and analyze their correlation with various body systems





7	Course Description	This subject is designed to impart fundamental knowledge on the structure and functions of the various systems of the human body. It also helps in understanding both homeostatic mechanisms. The subject provides the basic knowledge required to understand the various disciplines of pharmacy.		
8	Outline syllab	ıs		
UNIT-I A. Nervous system Organization of nervous system, neuron, neuroglia, classification and nerve fibre, electrophysiology, action potential, nerve impulse, receptors, synapse, neurotransmitters. B. Central nervous system: Meninges, ventricles ofbrain cerebrospinal fluid.structure and fu nctions of brain (cerebrospinal stem, cerebellum),		 A. Nervous system Organization of nervous system, neuron, neuroglia, classification and properties of nerve fibre, electrophysiology, action potential, nerve impulse, receptors, synapse, neurotransmitters. B. Central nervous system: Meninges, ventricles ofbrain and cerebrospinal fluid.structure and fu nctions of brain (cerebrum, brain stem, cerebellum), C. spinal cord (gross structure, functions of afferent and efferent 		
	2	UNIT-II		
		Anatomy of GI Tract with special reference to anatomy and functions of stomach, (Acid production in the stomach, regulation of acid production through parasympathetic nervous system, pepsinrole in protein digestion) small intestine and large intestine, anatomy and functions of salivary glands, pancreas and liver, movements of GIT, digestion and absorption of nutrients and disorders of GIT. B. Energetics Formation and role of ATP, CreatininePhosphateandBMR. C. Joints Structural and functional classification, types of joints movements and its articulation		





3	UNIT-III		
	 A. Respiratory system 10 hours Anatomy of respiratory system with special reference to anatomy of lungs, mechanism of respiration, regulation of respiration Lung Volumes and capacities transport of respiratory gases, artificial respiration, and resuscitation methods. B. Urinary system Anatomy of urinary tract with special reference to anatomy of kidney and nephrons, functions of kidney and urinary tract, physiology of urine formation, C. micturition reflex and role of kidneys in acid base balance, role of RAS in kidneyand disorders of kidney. 		
4	UNIT-IV		
	A. Endocrine system		
	Classification of hormones, mechanism of hormone action, B.structure and functions of pituitary gland, thyroid gland, parathyroid gland, adrenalgland,		
5	C.pancreas, pineal gland, thymus and their disorders. UNIT-V		
	A. Reproductive system Anatomy of male and female reproductive system, Functions of male and femalereproductive system, sex hormones, physiology of menstruation, fertilization, spermatogenesis, oogenesis, pregnancy and parturition B. Introduction to genetics C. Chromosomes, genes and DNA, protein synthesis, genetic pattern of inheritance		
Mode of	Theory		
examination Weightage	Continuous Mode	Sessional Exam	ESE
Distribution	Assessment	Sessional Exam	LUL
	10 Marks	15	75
Text book/s*	Practical human anator	ny and physiology.	by S.R.Kale and R.R.Kale.
	A Manual of pharmaceut	ical biology practical b	y S.B.Gokhale, C.K.Kokate and
	S.P.Shriwastava.		
	Biology practical manual according to National core curriculum .Biology		
	forum of Karnataka. Prof	M.J.H.Shafi	



School:		SOP
Program	me:	B.Pharma
Branch:		Semester: 2
1	Course Code	BP202T
2	Course Title	Pharmaceutical organic chemistry-I Theory
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Compulsory
5	Course Objective	Upon completion of the course the student shall be able to 1. Write the structure, name and the type of isomerism of the organic compound
		2. Write the reaction, name the reaction and orientation of reactions.
		3. Account for reactivity/stability of compounds.
		4. Identify/ confirm the identification of organic compound.
6	Course Outcomes	CO1: The students will have the knowledge to identify, name, and write the structure of different aliphatic compounds and their derivatives. CO2: The students will be able to understand and explain the mechanism behind the naming reactions of different aliphatic compounds and their derivatives.
		CO3: The students can apply the knowledge to prepare the derivatives of aliphatic compounds with different functional groups.
		CO4: Students will analyze the chemical reactions, stabilities of organic compounds and properties of the compounds prepared by them in the laboratory.
		CO5: Students would evaluate by comparing compounds prepared by them with standard compounds by chemical and physical properties
		CO6: Students will analyze the chemical reactions and stabilities of organic compounds
7	Course Description	This subject deals with classification and nomenclature of simple organic compounds, structural isomerism, intermediates forming in reactions, important physical properties, reactions and methods of preparation of these compounds. The syllabus also emphasizes on mechanisms and orientation of reactions.





8	Outline syl	labus
	1	UNIT-I A. Classification, nomenclature and isomerism Classification of Organic Compounds B.Common and IUPAC systems of nomenclature of organic compounds C.Structural isomerisms in organic compounds
	2	UNIT-II A. Alkanes*, Alkenes* and Conjugated dienes*
		 SP³ hybridization in alkanes, Halogenation of alkanes, uses of paraffins. B. Stabilization of alkenes, SP2 hybridization of alkenes. E1 and E2 reactions – kinetics, order of reactivity of alkyl halides, rearrangement of carbocations, Saytzeffs orientation and evidences. E1 verses E2 reactions, Factors affecting E1 and E2 reactions. Ozonolysis, electrophilic addition C. reactions of alkenes, Markownikoff's orientation, free radical addition reactions of alkenes, Anti Markownikoff's orientation. Stability of conjugated dienes, Diel-Alder, electrophilic addition, free radical addition reactions of conjugated dienes, allylic rearrangement
	3	UNIT-III
		A. Alkyl halides* SN ₁ and SN ₂ reactions - kinetics, order of reactivity of alkyl halides, stereochemistry and rearrangement of carbocations.
		SN1 versus SN2 reactions, Factors affecting SN1 and SN2 reactions
		Structure and uses of ethylchloride, Chloroform, trichloroethylene, tetrachloroethylene, dichloromethane,
		tetrachloromethane and iodoform. B. Alcohols*- Qualitative tests, Structure and uses of Ethyl alcohol,
		C.Methyl alcohol, chlorobutanol, Cetosteryl alcohol, Benzyl alcohol, Glycerol, Propylene glycol





4			
	UNIT-IV		
	A. Carbonyl compounds* (Aldehydes and		
	ketones)	,	
	1	on, Electromeric et	ffect, aldol condensation, Crossed
	Aldolcondensation,		-
		ro reaction, Ben	zoin condensation, Perkin
	condensation,	0 1 1	1
	B. Qualitative tests	• •	
		•	Paraldehyde, Acetone,
	C.Chloral hydra	te, Hexamine,	Benzaldehyde, Vanilin,
	Cinnamaldehyde.		
5	UNIT-V		
	A. Carboxylic acid		
		ic acids, effect of	substituents on acidity, inductive
	effect and		
	qualitative tests for	•	
		of Acetic acid, Lac	etic acid, Tartaric acid, Citric acid,
	Succinic acid.	ardia asid Danmaia	and Donard houseasts Discothed
	phthalate, Methyl sa	•	c acid, Benzyl benzoate, Dimethyl
	Aliphatic amines.	negrate and Accept	a same yile acid
	1 *	of substituents	on basicity, identification test,
			thylenediamine, amphetamine.
Mode of		•	*
examination	•		
Weightage	Continuous	Sessional Exam	ESE
Distribution	Mode		
	Assessment		
	10 Marks	15	75
Text book/s*	Practical human	anatomy and p	physiology. by S.R.Kale and
	R.R.Kale.		
	A Manual of p	harmaceutical bi	ology practical by S.B.Gokhale,
	C.K.Kokate and S.	P.Shriwastava.	
	Biology practical	manual accordin	g to National core curriculum
	.Biology forum of	Karnataka. Prof	
	.M.J.H.Shafi		
Other			
References			





	SOP	
nme:	B.Pharm	
	Semester: 2	
Course Code	BP203 T	
Course Title	Biochemistry- Theory	
Credits	4	
Contact Hours (L-T-P)	3-1-0	
Course Type	Compulsory	
Course	Upon completion of the course, the student shall be able to	
Objective	-Know the classification and salient features of five kingdoms of life	
	-Understand the basic components of anatomy & physiology of plant -Know understand the basic components of anatomy & physiology animal with special reference to human	
Course		
Outcomes	 CO1: The students will understand the structure and functions of carbohydrate, lipids, nucleic acids, amino acids and proteins. Concept of free energy, endergonic and exergonic reaction, Relationship, between free energy. CO2: The student will be able to summarize the Citric acid cycle-Pathway, energetics and significance, HMP shunt and its significance; Glucose-6-Phosphate dehydrogenase (G6PD) deficiency 	
	CO3: The student will be able to apply the knowledge of the Amino acid and lipid metabolism.	
	CO4: The students will analyze the correlation of Nucleic acid metabolism and genetic information transfer.	
	CO5: The students would Introduction, properties, nomenclature and IUB classification of enzymes, Enzyme kinetics (Michaelis plot, Line Weaver Burke plot).	
	CO6: The student will understand the importance of biochemistry.	
Course Description	Biochemistry deals with complete understanding of the molecular levels of the chemical process associated with living cells. The scope of the subject is providing biochemical facts and the principles to understand metabolism of nutrient molecules in physiological and pathological conditions. It is also emphasizing on genetic organization of mammalian genome and hetero & autocatalytic functions of DNA.	
	Course Code Course Title Credits Contact Hours (L-T-P) Course Type Course Objective Course Outcomes	





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8	Outline sylla	abus
	1	UNIT-I A. Biomolecules and Bioenergetics
		 Topic1- Introduction, classification, chemical nature and biological role of carbohydrate. Topic2- Introduction, classification, chemical nature and biological rolelipids, nucleic acids, amino acids and proteins. B. Topic3-Concept of free energy, endergonic and exergonic reaction, Relationship between free energy, enthalpy and entropy; Redox potential.
		C. Energy rich compounds; classification; biological significances of ATP and cyclic AMP
	2	UNIT-II
		A. Carbohydrate metabolism and Biological oxidation
		Topic1- Glycolysis – Pathway, energetics and significance Citric acid cycle- Pathway, energetics and significance HMP shunt and its significance; Glucose-6-Phosphate dehydrogenase(G6PD) deficiency Glycogen metabolism Pathways and glycogen storage diseases (GSD). B. Topic2- Gluconeogenesis- Pathway and its significance Hormonal regulation of blood glucose level and Diabetes mellitus C. Topic3- Electron transport chain (ETC) and its mechanism. Oxidative phosphorylation & its mechanism and
		substrate Phosphorylation, Inhibitors ETC and oxidative
	3	phosphorylation/Uncouplrslevel UNIT-III
	3	A. Lipid metabolism and Amino acid metabolism
		β-Oxidation of saturated fatty acid (Palmitic acid) 61Formation and utilization of ketone bodies; ketoacidosis De novo synthesis of fatty acids (Palmitic acid) Biological significance of cholesterol and conversion of cholesterol into bile acids, steroid hormone and vitamin D Disorders of lipid metabolism: Hypercholesterolemia, atherosclerosis, fatty liver and obesity.
		B. General reactions of amino acid metabolism: (Phenyketonuria, Albinism, alkeptonuria, tyrosinemia) Synthesis and significance of biological substances; 5-C.HT,melatonin,dopamine,noradrenaline,adrenaline Catabolism of heme; hyperbilirubinemia and jaundice
	4	
	•	·





5	UNIT-V			
	A. Enzymes	A. Enzymes		
	Topic1 - Introduction, properties, nomenclature and IUB classification of enzymes			
	Enzyme kinetics (Enzyme kinetics (Michaelis plot, Line Weaver Burke plot) Enzyme inhibitors with examples		
		•	es: enzyme induction and	
	repression, allosteric	•	•	
			stic applications of enzymes and	
	isoenzymes, Coenzy	ymes –Structure an	d biochemical functions	
Mode of	Theory			
examination				
Weightage	Continuous	Sessional Exam	ESE	
Distribution	Mode			
	Assessment			
	10 Marks	15	75	
Text book/s*	Practical human R.R.Kale.	anatomy and p	physiology. by S.R.Kale and	
	A Manual of p	harmaceutical bi	ology practical by S.B.Gokhale,	
	C.K.Kokate and S.	P.Shriwastava.		
	Biology practical	manual accordin	g to National core curriculum	
	.Biology forum of	Karnataka. ProfM.	J.H.Shafi	





School:		SOP		
Progra	mme:	B.Pharm		
Branch:		Semester: 2		
1	Course Code	BP204T		
2	Course Title	Pathophysiology- Theory		
3	Credits	4		
4	Contact Hours (L-T-P)	3-1-0		
	Course Type	Compulsory		
5	Course Objective	 To distinguish between environmental factors, physical, psychosocial, and cognitive characteristics of various diseases and conditions. To understand basic concepts of inflammatory diseases To Demonstrate and understand mechanisms of diseases, the diagnosis of diseases, and the treatment of diseases To understand how the various organ systems are interrelated, and use this understanding to promote a holistic approach towards the evaluation and treatment of patients 		
6	Course Outcomes	CO1: Student will be able to Distinguish environmental factors, physical, psychosocial, and cognitive characteristics of various diseases and conditions. CO2: Student will be able to apply concepts and elements of inflammatory diseases CO3: Student will be able to Demonstrate an understanding of the mechanisms of diseases, the diagnosis of diseases, and the treatment of diseases CO4: Students will be able to understand how the various organ systems are interrelated, and use this understanding to promote a holistic approach towards the evaluation and treatment of patients CO5: Students will be able to compare and discriminate between the Infectious and sexually transmitted diseases. CO6: Students will be able to understand about infectious diseases.		
7	Course Description	Pathophysiology is the study of causes of diseases and reactions of the body to such disease producing causes. This course is designed to impart a thorough knowledge of the relevant aspects of pathology of various conditions with reference to its pharmacological applications, and understanding of basic pathophysiological mechanisms. Hence it will not only help to study the syllabus of pathology, but also to get baseline knowledge required to practice medicine safely, confidently, rationally and effectively.		
8	Outline syllabus	3		





	 UNIT-I A.Basic principles of Cell injury and Adaptation & Basic mechanism involved in the process of inflammation and repair Causes of cellular injury, Pathogenesis (Cell membrane damage, Mitochondrial damage, Ribosome damage, Nuclear damage). A. Morphology of cell injury – Adaptive changes (Atrophy, Hypertrophy, hyperplasia, Metaplasia, Dysplasia), Cell swelling, Intra cellular accumulation, Calcification, Enzyme leakage and Cell Death Acidosis& Alkalosis, Electrolyte imbalance. B. Introduction, Clinical signs of inflammation, Different types of Inflammation, Mechanism of Inflammation – Alteration in vascular permeability and blood flow, migration of WBC's, C. Mediators of inflammation, Basic principles of wound healing in the skin, Pathophysiology of Atherosclerosis
2	UNIT-II A. Cardiovascular, Respiratory and Renal Diseases Hypertension, congestive heart failure, ischemic heart disease (angina,myocardial infarction, B.atherosclerosis and arteriosclerosis)Asthma,Chronic obstructive airways diseases. C. Acute and chronic renal failure
3	UNIT-III A.Hematological, Endocrine, Nervous and GIT diseases Iron deficiency, megaloblastic anemia (Vit B12 and folic acid), sickle cell anemia, thalasemia, hereditary acquired anemia, B.haemophilia Diabetes, thyroid diseases, disorders of sex hormones & Peptic ulcer C. Epilepsy, Parkinson's disease, stroke, psychiatric disorders: depression, schizophrenia and Alzheimer's disease.
4	UNIT-IV A. Cancer and inflammatory diseases Classification, etiology and pathogenesis of cancer Inflammatory bowel diseases, jaundice, hepatitis (A,B,C,D,E,F) alcoholic liver disease. B. Rheumatoid arthritis, C. osteoporosis and gout





5	UNIT-V		
	A. Infectious &Sexually transmitted diseases (STDs)		
	Meningitis, Typhoid, Leprosy,		
	B.Tuberculosis		
	Urinary tract infection		
	C.AIDS, Syphilis & C	Gonorrhea	
Mode of	Theory		
examination			
Weightage	Continuous	Sessional Exam	ESE
Distribution	Mode		
	Assessment		
	10 Marks	15	75
Text book/s*	Practical human	anatomy and p	hysiology. by S.R.Kale and
	R.R.Kale.		
	A Manual of pharma	aceutical biology p	ractical by S.B.Gokhale,
	C.K.Kokate and S.P	.Shriwastava.	
	Biology practical ma	anual according to N	ational core curriculum .Biology
	forum of Karnataka. F	Prof	
	.M.J.H.Shafi		
Other			
References			





School	:	SOP	
Progra	mme:	B.Pharm Semester: II	
Branch	ı:		
1	Course Code	BP205T	
2	Course Title	Computer applications in Pharmacy- Theory	
3	Credits	4	
4	Contact Hours (L-T-P)	3-1-0	
	Course Type	Compulsory	
5	Course Objective	Upon completion of the course the student shall be able to	
		know the various types of application of computers in pharmacy	
		know the various types of databases	
		know the various applications of databases in pharmacy	
6	Course	Upon completion of the course, the student shall be able to	
	Outcomes	CO1: understand the Binary number system	
		CO2: interpret the web technologies	
		CO3: apply about application of computers in Pharmacy	
		CO4: the Bioinformatics Databases, Concept of Bioinformatics	
		CO5: Computers as data analysis in Preclinical development:	
		CO6: Students will understand use of Computers in Preclinical	
		development:	
7	Course Description	This subject deals with the introduction Database, Database Management system, computer application in clinical studies and use of databases.	
8	Outline syllabus	S	
	1	YINYE Y	
		UNIT-I	
		A. Number system: Binary number system, Decimal number system, Octal number system, Hexadecimal number systems, conversion decimal to binary, binary to decimal, octal to binary etc, binary addition, binary subtraction — One's complement ,Two's complement method, binary multiplication, binary division	



2		UNIT-II		
		A. Webtechnologies: XML,CSS and Prointroduction to w	grammeming languates servers and Se	ages,
3		UNIT-III		
		and retrieval, Phan Hospital and Clini (EP) systems, barco of drugs, mobile	rmacokinetics, Math ical Pharmacy, Electode medicine identi technology and a nostic System, Pati	hematical model in Drug design, ctronic Prescribing and discharge fication and automated dispensing adherence monitoring Diagnostic ient Monitoring System, Pharma
4		UNIT-IV		
		A. Bioinformatics: Bioinformatics De Bioinformatics in V	atabases, Concept	Objective of Bioinformatics, of Bioinformatics, Impact of
5		UNIT-V		
		- -	dada analysis(C	in Preclinical development: CDS), Laboratory Information Text Information Management
	ode of amination	Theory/Jury/Practical/	Viva	
	eightage istribution	Continuous Mode Assessment	Sessional Exam	ESE
		10	15	75





Text book/s*	
1 CAL GOOM'S	1. Computer Application in Pharmacy – William E.Fassett –Lea and
	Febiger, 600 South Washington Square, USA, (215) 922-1330.
	2. Computer Application in Pharmaceutical Research and Development –
	Sean Ekins –
	Wiley-Interscience, A John Willey and Sons, INC., Publication, USA
	3. Bioinformatics (Concept, Skills and Applications) – S.C.Rastogi-CBS
	Publishers and Distributors, 4596/1- A, 11 Darya Gani, New Delhi – 110
	002(INDIA)
	4. Microsoft office Access - 2003,
	Application Development Using VBA, SQL Server, DAP and Infopath -
	Cary N.Prague – Wiley Dreamtech India
	(P) Ltd., 4435/7, Ansari Road, Daryagani, New Delhi - 110002
Other	
References	





School:		SOP		
Programme:		B. Pharm		
Bran		Semester: II		
1	Course Code	BP206 T		
	Course Title	Environmental Sciences (Theory)		
3	Credits	4		
4	Contact Hours (L-T-P)	3-1-0		
	Course Type	Compulsory		
5	Course Objective	Upon completion of the course the student shall be able to:		
		Create the awareness about environmental problems among learners. Impart basic knowledge about the environment and its allied problems.		
		Develop an attitude of concern for the environment.		
		Motivate learner to participate in environment protection and environment improvement. Acquire skills to help the concerned individuals in identifying and solving environmental problems.		
-		Strive to attain harmony with Nature.		
6	Course Outcomes	CO1: Student shall be able to understand the Multidisciplinary nature of environmental studies CO2: Student shall be able to apply Concept of an ecosystem. CO3: Student shall be able to analyze Structure and function of an ecosystem. CO4: Student shall be able to explain Structure and function of anenvironment CO5: Student shall be able to analyze computer data in Preclinical development: CO6: Students will be able to understand about water and air pollution		
7	Course Description	Environmental Sciences is the scientific study of the environmental system and the status of its inherent or induced changes on organisms. It includes not only the study of physical and biological characters of the environment but also the social and cultural factors and the impact of man on environment.		
8	Outline syllabus	S .		
	1	 UNIT-I A. The Multidisciplinary nature of environmental studies Natura lResources Renewable and non-renewable resources: B. Natural resources and associated problems (a)Forest resources; b) Water resources; c) Mineral resources; C.Food resources; e) Energy resources; f) Land resources: Role of an individual in conservation of natural resources. 		





	www.snaroa.ac.in		
2	UNIT-II		
	Ecosystems		
	A. Concept of an ecosystem.		
	B. Structure and function	on of an ecosystem.	
		•	res, structure and function of the
	ecosystems:		
			stem; Desert ecosystem; Aquatic
	ecosystems (ponds, stre	eams, lakes, rivers, o	oceans, estuaries
3	UNIT-III		
		on: Air pollution; Wa	ater pollution; Soil pollution
Mode of examination	Theory		
Weightage	Continuous Mode	Sessional Exam	ESE
Distribution	Assessment		
	10	15	75
Text book/s*			nce, New Age International
		hers, Bangalore	
	•	C.C. 2001 Environme	ental Biology, Nidi Publ. Ltd.
	Bikaner.		
			sity of India, Mapin Pu
		rt. Ltd., Ahmedabad	
		C., 1989, Hazardous	s Waste Incineration, McGraw Hill
	Inc. 480p 5. Clark R.S.,	Marina Dallytian C	Slandongen Duese Ovrfond
	5. Clark R.S., 6. Cunninghan		Clanderson Press Oxford T.H. Gorhani, E &
	Hepworth,		nvironmental Encyclopedia,
	•	House, Mumbai, 11	• 1
			istry, Wiley Eastern Ltd.
			nce and Environment



School:		SOP
Progra	mme:	B. Pharm
Branch	:	Semester: 2
1	Course Code	BP207 P
2	Course Title	Human Anatomy & Physiology-II Practical
3	Credits	2
4	Contact Hours (L-T-P)	0-0-4
	Course Type	Compulsory
5	Course Objective	Upon completion of the course, the student shall be able to -Know the classification and salient features of five kingdoms of life -Understand the basic components of anatomy & physiology of plant
		-Know understand the basic components of anatomy & physiology animal with special reference to human
6	Course Outcomes	CO1 : The students will understand the structure and functions of various tissues and organs of the body. Also correlate their relevance with each other.
		CO2: The student will be able to summarize the functioning of various body systems and their homeostasis.
		CO3: The student will be able to apply the knowledge of the anatomy and physiology of different body parts in explaining the working patterns of different body systems.
		CO4: The students will analyze the structures of various tissues and their origin to evaluate their damage and repair process.
		CO5: The students would evaluate the mechanisms of various processes on which the functioning of the various body organs depend. Moreover, will observe the anatomical differentiation of different body parts.
		CO6: The students would evaluate the molecular mechanisms of various body pathways.
7	Course Description	Practical physiology is complimentary to the theoretical discussions in physiology. Practicals allow the verification of
		physiological processes discussed in theory classes through
		experiments on living tissue, intact animals or normal human beings. This is helpful for developing an insight on the subject.
8	Outline syllabus	





			www.sharda.ac.in
	UNIT-I To study the integur system using specim		l senses, nervous system, endocrine
2	1 *	ction of olfactory r y.	egical examination. b). To nerve, different types of taste, visual
3	UNIT-III	oositive and negati	ve feed back mechanism. vital Capacity.
4	UNIT-IV a). Study of digestive	e, respiratory, card	liovascular systems, urinary and nodels, charts and specimens.
5	UNIT-V a). Study of family p	planning devices and f total blood count	nd pregnancy diagnosistest. by cell analyser Permanent slides
Mode of examination	Practical/Viva	2	
Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE
	05	10	35
Text book/s*	 Essentials of P. Sembulings Delhi. Anatomy and Kathleen J.W Physiological 	Medical Physiolog am. Jaypee brother Physiology in Hea V. Wilson, Churchi	y by K. Sembulingam and se medical publishers, New alth and Illness by ill Livingstone, New York Practice-Best and Tailor.
Other References			





School:		SOP
Program	me:	B. Pharm
Branch:		Semester: 2
1	Course Code	BP208 P
2	Course Title	Pharmaceutical Organic Chemistry-I Practical
3	Credits	2
4	Contact Hours (L-T-P)	4
	Course Type	Compulsory
5	Course Objective	This subject deals with classification and nomenclature of simple organic compounds, structural isomerism, intermediates forming in reactions, important physical properties, reactions and methods of preparation of these compounds. The syllabus also emphasizes on mechanisms and orientation of reactions.
6	Course Outcomes	CO1 - Students will be able to understand practical laboratory skills and get hands-on training of systematic qualitative analysis of organic compounds.
		CO2 - Students will be able to apply knowledge and understanding of systematic qualitative analysis of organic compounds and will be able to apply this knowledge in identification of organic compounds. CO3 - Students will be able to prepare the solid derivatives of organic compounds and can apply this knowledge for the identification of drugs and pharmaceuticals also use these skills to modify various characteristics of drugs and Pharmaceuticals. CO4 - Students will be able to analyze professional transferable skills as exemplified by problem solving and teamwork.
		CO5 - Students will be able to generalize skills for the predicting the atomic structure of drugs and chemicals.
		CO6 - Students will be able to analyze models of different organ system that will elaborate the learning



		www.snarda.ac.in
7	Course Description	 Systematic qualitative analysis of unknown organic compounds like Preliminary test: Color, odour, aliphatic/aromatic compounds, saturation and unsaturation, etc. Detection of elements like Nitrogen, Sulphur and Halogen by Lassaigne's test. Functional group test like Phenols, Amides/ Urea, Carbohydrates, Amines, Carboxylic acids, Aldehydes and Ketones, Alcohols, Esters, Aromatic and Halogenated Hydrocarbons, Nitro compounds and Anilides. Melting point/Boiling point of organic compounds Identification of the unknown compound from the literature Using melting point/ boiling point Preparation of the derivatives and confirmation of the unknown compound by melting point/ boiling point. Minimum 5 unknown organic compounds to be analyzed systematically. Preparation of suitable solid derivatives from organic compounds Construction of molecular models
8	Outline syllabus	
	1	UNIT-I I. Experiments involving preliminary test: Color, odour, aliphatic/aromatic compounds, saturation and unsaturation, etc. Physical characteristics Flame Test Bromine Test
	2	UNIT-II Element Detection (Lassaigne's test)
	3	UNIT-III Solubility test
	4	UNIT-IV Functional group test like Phenols, Amides/ Urea, Carbohydrates, Amines, Carboxylic acids, Aldehydes and Ketones, Alcohols, Esters, Aromatic and Halogenated Hydrocarbons, Nitro compounds and Anilides.





5	UNIT-V Melting point/Boilir Preparation of suital Construction of mol	ole solid derivative	compounds es from organic compounds
Mode of examination	Practical/Viva		
Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE
	05	10	35
Text book/s*	 Organic Chen Textbook of C Organic Chen Practical Organic Vogel's text b Advanced Prantal Chen Introduction than Kriz. 	nistry by P.L.Soni anic Chemistry by book of Practical O actical organic cher o Organic Laborat	by B.S. Bahl & Arun Bahl. Mann and Saunders.
Other References			





School:		SOP
Progran	ıme:	B. Pharm
Branch:		Semester: 2
1	Course Code	BP 209 0P
2	Course Title	Biochemistry Practical
3	Credits	2
4	Contact Hours (L-T-P)	0-0-4
	Course Type	Compulsory
5	Course Objective	Upon completion of course student shall able to
		Understand the catalytic role of enzymes, importance of enzyme inhibitors in design of new drugs, therapeutic and diagnostic applications of enzymes. Understand the metabolism of nutrient molecules in physiological and pathological conditions. Understand the genetic organization of mammalian genome and functions of DNA in the synthesis of RNAs and proteins.
6	Course Outcomes	CO1: Students will be able to understand the Qualitative analysis of carbohydrates CO2: Students will be able to understand the Quantitatative analysis of reducing sugars. CO3: Students will be able to analyze how to determine creatinine CO4: Students will be able to determine serum cholesterol CO5: Students will be able to compare amino acids by Paper Chromatographic Technique. CO6: Students will be able to apply the practical aspect and use of biochemistry.
7	Course Description	 Qualitative analysis of carbohydrates (Glucose, Fructose, Lactose, Maltose, Sucrose and starch) Identification tests for Proteins (albumin and Casein) Quantitative analysis of reducing sugars (DNSA method) and Proteins (Biuret method) Qualitative analysis of urine for abnormal constituents Determination of blood creatinine Determination of blood sugar Determination of serum total cholesterol Preparation of buffer solution and measurement of pH Study of enzymatic hydrolysis of starch Determination of Salivary amylase activity Study the effect of Temperature on Salivary amylase activity. Study the effect of substrate concentration on salivary amylase activity.





	0 11 11 1			
8	Outline syllabus			
	1	UNIT-I		
		a). Qualitative analy	sis of carbohydrat	es (Glucose, Fructose, Lactose,
		Maltose, Sucrose an	d starch)	
		b). Identification te	sts for Proteins (all	bumin and Casein)
	2	UNIT-II		
		a). Quantitative ana	lysis of reducing s	ugars (DNSA method) and Proteins
		(Biuret method)		
		b). Qualitative analy	sis of urine for ab	normal constituents
	3	UNIT-III		
		a). Determination of	f blood creatinine	
		b). Determination of	f blood sugar	
			_	
	4	IDIT IV		
		UNIT-IV	C 4 4 1 1 1	. 1
		a). Determination of		
	7	b). Preparation of bu	affer solution and i	neasurement of pH
	5	UNIT-V	. 1 1 1	1
		a). Study of enzyma		
			f amino acids by P	aper Chromatographic Technique.
) A 1 C	Practical/Viva		
	Mode of			
	examination		G : 15	DOE
	Weightage	Continuous	Sessional Exam	ESE
	Distribution	Mode		
		Assessment		
		05	10	35
	Text book/s*	Practical Biochemis	stry by R.C. Gupta	a and S. Bhargavan. Introduction of
		Practical Biochemis	try by David T. Pl	ummer. (3rd Edition)
		Practical Biochem	istry for Medic	cal students by Rajagopal and
		Ramakrishna.	•	, , , ,
		Practical Biochemis	try by Harold Varl	ey
	Other		- · ·	
	References			





information about a particular disease. Create a HTML web page to show personal information. Retrieve the information of a drug and its adverse effects using online tools Creating mailing labels Using Label Wizard, generating label in MS WORD Create a database in MS Access to store the patient information with the required fields Using access	School	•	SOP
Course Code BP210 P	Progra	mme:	B. Pharm
2 Course Title Computer applications in Pharmacy-Practical 3 Credits 2 4 Contact Hours (L-T-P) Course Type Course Objective Dipective Upon completion of the course the student shall be able to know the various types of application of computers in pharmacy know the various applications of databases in pharmacy Not course Outcomes Course Outcomes Course Outcomes Course Upon completion of the course, the student shall be able to Col: understand the Binary number system Co2: the web technologies Co3: the application of computers in Pharmacy Co4: the Bioinformatics Databases, Concept of Bioinformatics Co5: Computers as data analysis in Preclinical development Co6: Students will overall understand the use of computers in pharmacy Course Description Design a questionnaire using a word processing package to gather information about a particular disease. Create a HTML web page to show personal information. Retrieve the information of a drug and its adverse effects using online tools Creating mailing labels Using Label Wizard, generating label in MS WORD Create a database in MS Access to store the patient information with the required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access	Branch	1:	Semester: 2
Credits Contact Hours (L-T-P)	1	Course Code	BP210 P
Course Objective Upon completion of the course the student shall be able to know the various types of application of computers in pharmacy know the various applications of databases in pharmacy Upon completion of the course, the student shall be able to Course Outcomes Upon completion of the course, the student shall be able to CO1: understand the Binary number system CO2: the web technologies CO3: the application of computers in Pharmacy CO4: the Bioinformatics Databases, Concept of Bioinformatics CO5: Computers as data analysis in Preclinical development CO6: Students will overall understand the use of computers in pharmacy Course Description Design a questionnaire using a word processing package to gather information about a particular disease. Create a HTML web page to show personal information. Retrieve the information of a drug and its adverse effects using online tools Creating mailing labels Using Label Wizard, generating label in MS WORD Create a database in MS Access to store the patient information with the required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access	2	Course Title	Computer applications in Pharmacy- Practical
Course Type Compulsory	3	Credits	2
Course Objective Upon completion of the course the student shall be able to know the various types of application of computers in pharmacy know the various applications of databases in pharmacy	4	(L-T-P)	
Objective Upon completion of the course the student shall be able to know the various types of application of computers in pharmacy know the various applications of databases in pharmacy Upon completion of the course, the student shall be able to CO1: understand the Binary number system CO2: the web technologies CO3: the application of computers in Pharmacy CO4: the Bioinformatics Databases, Concept of Bioinformatics CO5: Computers as data analysis in Preclinical development CO6: Students will overall understand the use of computers in pharmacy Design a questionnaire using a word processing package to gather information about a particular disease. Create a HTML web page to show personal information. Retrieve the information of a drug and its adverse effects using online tools Creating mailing labels Using Label Wizard, generating label in MS WORD Create a database in MS Access to store the patient information with the required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access		Course Type	Compulsory
Course Outcomes Upon completion of the course, the student shall be able to CO1: understand the Binary number system CO2: the web technologies CO3: the application of computers in Pharmacy CO4: the Bioinformatics Databases, Concept of Bioinformatics CO5: Computers as data analysis in Preclinical development CO6: Students will overall understand the use of computers in pharmacy Design a questionnaire using a word processing package to gather information about a particular disease. Create a HTML web page to show personal information. Retrieve the information of a drug and its adverse effects using online tools Creating mailing labels Using Label Wizard, generating label in MS WORD Create a database in MS Access to store the patient information with the required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access	5		know the various types of application of computers in pharmacy know the various types of databases
Outcomes CO1: understand the Binary number system CO2: the web technologies CO3: the application of computers in Pharmacy CO4: the Bioinformatics Databases, Concept of Bioinformatics CO5: Computers as data analysis in Preclinical development CO6: Students will overall understand the use of computers in pharmacy Design a questionnaire using a word processing package to gather information about a particular disease. Create a HTML web page to show personal information. Retrieve the information of a drug and its adverse effects using online tools Creating mailing labels Using Label Wizard, generating label in MS WORD Create a database in MS Access to store the patient information with the required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access	6	Course	
Course Description Design a questionnaire using a word processing package to gather information about a particular disease. Create a HTML web page to show personal information. Retrieve the information of a drug and its adverse effects using online tools Creating mailing labels Using Label Wizard, generating label in MS WORD Create a database in MS Access to store the patient information with the required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access	O .		CO1: understand the Binary number system CO2: the web technologies CO3: the application of computers in Pharmacy CO4: the Bioinformatics Databases, Concept of Bioinformatics CO5: Computers as data analysis in Preclinical development
Description Design a questionnaire using a word processing package to gather information about a particular disease. Create a HTML web page to show personal information. Retrieve the information of a drug and its adverse effects using online tools Creating mailing labels Using Label Wizard, generating label in MS WORD Create a database in MS Access to store the patient information with the required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access	7	Course	Students will overall understand the use of computers in pharmacy
Create a database in MS Access to store the patient information with the required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access			Create a HTML web page to show personal information.
required fields Using access Design a form in MS Access to view, add, delete and modify the patient record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access			Creating mailing labels Using Label Wizard, generating label in MS WORD
record in the database Generating report and printing the report from patient database Creating invoice table using – MS Access			Create a database in MS Access to store the patient information with the required fields Using access
Creating invoice table using – MS Access			Design a form in MS Access to view, add, delete and modify the patient record in the database
			Generating report and printing the report from patient database
Exporting Tables, Queries, Forms and Reports to XML pages			Creating invoice table using – MS Access
			Exporting Tables, Queries, Forms and Reports to XML pages





8 Outline sylla	Outline syllabus		
	Octal number of decimal to binar addition, binar	system, Hexadecima ary, binary to decima	ystem, Decimal number system, I number systems, conversion al, octal to binary etc, binary s complement ,Two's complement ary division
2	Programmemin Products	ng languages, introdu	HTML, XML,CSS and action to web servers and Server CCESS, Pharmacy Drug database
3	and retrieval, I Hospital and C (EP) systems dispensing of C	Pharmacokinetics, M Clinical Pharmacy, El , barcode medicin drugs, mobile techno ab-diagnostic System	rmacy – Drug information storage fathematical model in Drug design, lectronic Prescribing and discharge the identification and automated logy and adherence monitoring the patient Monitoring System,
4	l l		Objective of Bioinformatics, ept of Bioinformatics, Impact of ry
5	Chromatograph management S System(TIMS)	System (LIMS) and	in Preclinical development: (CDS), Laboratory Information d Text Information Management
examination			
Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE
	05	05	15





Text book/s*		
	1.	Computer Application in Pharmacy - William E.Fassett -Lea and
		Febiger, 600 South Washington Square, USA, (215) 922-1330.
	2.	Computer Application in Pharmaceutical Research and Development –
		Sean Ekins –
		Wiley-Interscience, A John Willey and Sons, INC., Publication, USA
	3.	Bioinformatics (Concept, Skills and Applications) - S.C.Rastogi-CBS
		Publishers and Distributors, 4596/1- A, 11 Darya Gani, New Delhi –
		110 002(INDIA)
	4.	Microsoft office Access - 2003,
		Application Development Using VBA, SQL Server, DAP and Infopath
		- Cary N.Prague - Wiley Dreamtech India
		(P) Ltd., 4435/7, Ansari Road, Daryagani, New Delhi - 110002





School:		SOP
Programme:		B. Pharm
Branch		Semester: 3
1	Course Code	BP301 T
	Course Title	Pharmaceutical organic chemistry-II- Theory
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Compulsory
5	Course Objective	Upon completion of the course the student shall be able to
	objective	write the structure, name and the type of isomerism of the organic compound
		write the reaction, name the reaction and orientation of reactions
		account for reactivity/stability of compounds,
		prepare organic compounds
6	Course Outcomes	CO1: The students will have the knowledge to identify, name, and write the structure of different aromatic compounds and their derivatives.
		CO2: The students will be able to understand and explain the mechanism behind the naming reactions of different aromatic compounds and their derivatives.
		CO3: The students can apply the knowledge to prepare the derivatives of aromaic compounds with different fuctional groups.
		CO4: Students will analyze the chemical reactions, stabilities of organic compounds and properties of the compounds prepared by them in the laboratory.
		CO5: Students would evaluate bycomparing compounds prepared by them with standard compounds by chemical and physical properties.
		CO6: Students would be able to evaluate Poly nuclear hydrocarbons. by comparing structures.
7	Course Description	This subject deals with general methods of preparation and reactions of some organic compounds. Reactivity of organic compounds are also studied here. The syllabus emphasizes on mechanisms and orientation of reactions. Chemistry of fats and oils are also included in the syllabus.
8	Outline syllabu	





	TTT TIDING WARDIN
1	UNIT-I
	A. Benzene and its derivatives
	Analytical, synthetic and other evidences in the derivation of structure of
	benzene, Orbital picture, resonance in benzene, aromatic characters,
	Huckel's rule
	B. Reactions of benzene - nitration, sulphonation, halogenation-
	reactivity, Friedelcrafts alkylation- reactivity, limitations, Friedelcrafts
	acylation.
	C. Substituents, effect of substituents on reactivity and orientation
	of mono substituted benzene compounds towards electrophilic
	substitution reaction
	Structure and uses of DDT, Saccharin, BHC and Chloramine
2	UNIT-II
	A. Phenols* - Acidity of phenols, effect of substituents on acidity,
	qualitative tests, Structure and uses of phenol, cresols, resorcinol,
	naphthols
	B. Aromatic Amines* - Basicity of amines, effect of substituents on
	basicity, and synthetic uses of aryl diazonium salts
	C. Aromatic Acids* -Acidity, effect of substituents on acidity and
	important reactions of benzoic acid.
3	UNIT-III
	A. Fatty acids – reactions.
	Hydrolysis, Hydrogenation, Saponification and Rancidity of oils, Drying
	oils.
	B. Analytical constants – Acid value, Saponification value, Ester
	value, Iodine value, Acetyl value, Reichert Meissl (RM) value –
	significance and principle involved in their determination.
	UNIT-IV
	A. Polynuclear hydrocarbons:
	Synthesis, reactions
	B. Structure and medicinal uses of Naphthalene, Phenanthrene,
	Anthracene,
	C. Diphenylmethane, Triphenylmethane and their derivatives
	UNIT-V
	A. Cyclo alkanes*
	Stabilities – Baeyer's strain theory, limitation of Baeyer's strain theory, Coulson
	and Moffitt's modification, Sachse Mohr's theory (Theory of strainless rings),
	reactions of cyclopropane and cyclobutane only





Mode of	Theory		
examination			
Weightage	Continuous Mode	Sessional Exam	ESE
Distribution	Assessment		
	10	15	75
Text book/s*	 Organic Che Textbook of Organic Che Practical Org Vogel's text 	mistry by P.L. Soni ganic Chemistry by book of Practical O	, Volume-I by B.S. Bahl & Arun Bahl. Mann and Saunders.





School:		SOP		
Program	me:	B. Pharm		
Branch:		Semester: 3		
1	Course Code	BP302T		
2	Course Title	Physical Pharmaceutics I- Theory		
3	Credits	4		
4	Contact Hours (L-T-P)	3-1-0		
	Course Type	Compulsory		
5	Course Objective	Upon the completion of the course student shall be able to 1. Understand various physicochemical properties of drug molecules in the designing the dosage forms 2. Demonstrate use of physicochemical properties in the formulation development and evaluation of dosage forms. 3. Apply the concept of surface tension and surfactants in formulation and development.		
6	Course Outcomes	CO1: Students would be able to understand the concept of solubility, solutions, diffusion, CST, distribution and apply them in formulation, development and biological systems. CO2: Students would be able to explain the basics of states of matter and physical properties of drugs and use them in pharmaceutical field. CO3: Students would be able to apply the basics of surface and interfacial tension, surface active agents, HLB and adsorption in formulation and development of pharmaceutical systems. CO4: Students would be able to describe Complexation, protein binding and relate it with drug action. CO5: Students would be able to compare the methods of determination of pH and demonstrate the applications of buffered isotonic solutions in pharmaceutical and biological systems. CO6: Students would be able to evaluate the Isoonicity of solutions.		
7	Course Description	The course deals with the various physica and physicochemical properties, and principles involved in dosage forms/formulations. Theory and practical components of the subject help the student to get a better insight into various areas of formulation research and development, and stability studies of pharmaceutical dosage forms.		





8	Outline syllabus	
0	,	
		 UNIT-I A. Solubility of drugs: Solubility expressions, mechanisms of solute solvent interactions, ideal solubility parameters, solvation & association, quantitative approach to the factors influencing solubility of drugs, diffusion principles in biological systems. B. Solubility of gas in liquids, solubility of liquids in liquids, (Binary solutions, ideal solutions) Raoult's law, real solutions. C. Partially miscible liquids, Critical solution temperature and applications. Distribution law, its limitations and applications
	2	UNIT-II A. States of Matter and properties of matter: State of matter, changes in the state of matter, latent heats, vapour pressure, sublimation critical point, eutectic mixtures, gases, aerosols inhalers, relative humidity, liquid complexes, liquid crystals, glassy states, solid-crystalline, amorphous & polymorphism.
		 B. Physicochemical properties of drug molecules: Refractive index, optical rotation, dielectric constant, dipole moment, C. dissociation constant, determinations and applications
	3	 UNIT-III A. Surface and interfacial phenomenon: Liquid interface, surface & interfacial tensions, surface free energy, measurement of surface & interfacial tensions, spreading coefficient, adsorption at liquid interfaces, surface active agents, HLB Scale, solubilisation, detergency, adsorption at solid interface.
	4	 UNIT-IV A. Complexation and protein binding: Introduction, Classification of Complexation, Applications, methods of analysis, B. protein binding, C. Complexation and drug action, crystalline structures of complexes and thermodynamic treatment of stability constants.
	5	 UNIT-V A. buffers and Isotonic solutions: Sorensen's pH scale, pH determination (electrometric and calorimetric), B. B. applications of buffers, buffer equation, buffer capacity, C. buffers in pharmaceutical and biological systems, buffered isotonic solutions.
	Mode of examination	Theory





Weightage	Continuous	Sessional Exam	ESE
Distribution	Mode		
	Assessment		
	10 Marks	15	75
Text book/s*	Physical Pharmacy 1	by Alfred Martin	
	Experimental Pharm	naceutics by Eugen	e, Parott.
	Tutorial Pharmacy b	y Cooper and Gur	ın.
	Stocklosam J. Pharmaceutical Calculations, Lea &Febiger, Philadelphia.		
	Liberman H.A, Lachman C., Pharmaceutical Dosage forms,		
	Tablets, Volume-1 to 3, MarcelDekkar Inc.		
	Liberman H.A,	Lachman C, Pl	narmaceutical
	Dosage forms. Disp	perse systems, vol	lume 1, 2, 3.
	Marcel Dekkar Inc.		





Sc	hool:	SOP
	ogramme:	B. Pharm
	anch:	Semester: 3
1	Course	BP303 T
	Code	
2	Course	Pharmaceutical Microbiology- Theory
	Title	
3	Credits	4
4	Contact	3-1-0
	Hours	
	(L-T-P)	
	Course	Compulsory
5	Type Course Objective	Upon completion of the course the student shall be able to tell about the history, scope of microbiology and describe the structure, morphology and cultivation of microorganism. Student shall identify the bacteria on the basis of various staining technique and importance of sterilization in microbiology. Upon completion of the course the student shall understand the various methods for assessment of antibiotic, test for sterility for preparation. Student shall analyze the source of contamination and their prevention in aseptic areas and importance of cell culture technique.
Outcomes and application of various kind of microscopy. CO2: Students shall be able the differentiate the types of bacteria on the basis of statechnique and biochemical test and with different type of microscopic technique their method of validation CO3: Students shall acquire complete knowledge of microorganism (viruses, fung classification reproduction pattern, disinfection and antiseptic their evaluation metand about sterility testing of various pharmaceutical products. CO4: Students can apply their knowledge to design the aseptic area and standardized of antibiotic, biomolecules. CO5: Students will be able to analyze the sources of contamination and their preventions.		CO2: Students shall be able the differentiate the types of bacteria on the basis of staining technique and biochemical test and with different type of microscopic technique and their method of validation CO3: Students shall acquire complete knowledge of microorganism (viruses, fungi) like classification reproduction pattern, disinfection and antiseptic their evaluation methods and about sterility testing of various pharmaceutical products. CO4: Students can apply their knowledge to design the aseptic area and standardization of antibiotic, biomolecules. CO5: Students will be able to analyze the sources of contamination and their preventions in pharmaceutical products, and their application in pharmaceutical industry and





7	Course Description	antihiotics vaccines vitamins enzymes etc	
8	Outline syllab	ous	
	1	UNIT-I	
		 A. Introduction, history of microbiology, its branches, scope and its importance. B. Introduction to Prokaryotes and Eukaryotes Study of ultra-structure and morphological classification of bacteria, nutritional requirements, raw materials used for culture media and physical parameters for growth, growth curve, isolation and preservation methods for pure cultures, cultivation of anaerobes, quantitative measurement of bacterial growth (total & viable count). C. Study of different types of phase constrast microscopy, dark field microscopy and electron microscopy. 	
	2		
		UNIT-II A. Identification of bacteria using staining techniques (simple, Gram's & Acid fast staining) and biochemical tests (IMViC).	
		B. Study of principle, procedure, merits, demerits and applications of physical, chemical gaseous, radiation and mechanical method of sterilization.	
	2	C. Evaluation of the efficiency of sterilization methods.	
	3	UNIT-III A. Study of morphology, classification, reproduction/replicationand cultivation of Fungi and Viruses.	
		Classification and mode of action of disinfectants	
		Factors influencing disinfection, antiseptics and their evaluation. For bacteriostatic and bactericidal actions	
		B. Evaluation of bactericidal & Bacteriostatic.	
		C. Sterility testing of products (solids, liquids, ophthalmic and other sterile products) according to IP, BP and USP.	





4	UNIT-IV		
	 A. Designing of aseptic area, laminar flow equipments; study of different sources of contamination in an aseptic area and methods of prevention, clean area classification. B. Principles and methods of different microbiological assay. Methods for standardization of antibiotics, vitamins and amino acids. Assessment of a new antibiotic. 		
5	UNIT-VA. Types of spoilage, factors products.B. sources and types of microbiand spoilage.C. Preservation of pharmaceuti microbial stability of formula	ial contaminants, assessm	nent of microbial contamination
Mode of examination	Theory		
Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE
	10 Marks	15	75
Text book/s*	 W.B. Hugo and A.D. Russel: Pharmaceutical Microbiology, Blackwell Scientific publications, Oxford London. Prescott and Dunn., Industrial Microbiology, 4th edition, CBS Publishers & Distributors, Delhi. 		
	 Pelczar, Chan Kreig, Microbiology, Tata McGraw Hill edn. Malcolm Harris, Balliere Tindall and Cox: Pharmaceutical Microbiology. Rose: Industrial Microbiology. Probisher, Hinsdill et al: Fundamentals of Microbiology, 9th ed. Japan Cooper and Gunn's: Tutorial Pharmacy, CBS Publisher and Distribution. Peppler: Microbial Technology. I.P., B.P., U.S.P latest editions. Ananthnarayan: Text Book of Microbiology, Orient-Longman, Chennai Edward: Fundamentals of Microbiology. N.K.Jain: Pharmaceutical Microbiology, Vallabh Prakashan, Delhi Bergeys manual of systematic bacteriology, Williams and Wilkins- A Waverly company 		





School:		SOP		
Program	nme:	B. Pharm Semester: 3		
Branch:				
1	Course Code	BP 304 T		
2	Course Title	Pharmaceutical Engineering - Theory		
3	Credits	4		
4	Contact Hours (L-T-P)	3-1-0		
	Course Type	Compulsory		
5	Course Objective	Upon completion of the course student shall be able:		
		To know various unit operations used in Pharmaceutical industries.		
		To understand the material handling techniques.		
		To perform various processes involved in pharmaceutical manufacturing process.		
		To carry out various test to prevent environmental pollution.		
		To appreciate and comprehend significance of plant lay out design for optimum use of resources.		
		To appreciate the various preventive methods used for corrosion control in Pharmaceutical industries.		
6 Course Outcomes		CO1: Students will be able to describe about various unit operations used in pharmaceutical industries. and size separation and their applications in pharmaceutical field.		
		CO2: Students will be able to understand about basic concepts and importance of various heat transfer methods involved in pharmaceutical filed. CO3: Students will be able to illustrate about the concepts, equipments and pharmaceutical applications of drying and mixing.		
		CO4: Students will be able to distinguish between different types of equipments used in various unit operations such as filtration and centrifugation. CO5: Students will be able to predict about various materials used in pharmaceutical plant construction, types of corrosion and its prevention methods and basics of material handling system CO6: Students will be able to evaluate about various mechanisms of unit operations.		





	1	www.snarwa.ac.iii
7	Course Description	This course is designed to impart a fundamental knowledge on the art and science of various unit operations used in pharmaceutical industry.
8	Outline syllabı	ıs
	1	UNIT-I
		A. Flow of fluids: Types of manometers, Reynolds number and its significance, Bernoulli's theorem and its applications, Energy losses, Orifice meter, Venturimeter, Pitot tube and Rotometer.
		B. Size Reduction: Objectives, Mechanisms & Laws governing size reduction, factors affecting size reduction, principles, construction, working, uses, merits and demerits of Hammer mill, ball mill, fluid energy mill, Edge runner mill & end runner mill.
		C. Size Separation: Objectives, applications & mechanism of size separation, official standards of powders, sieves, size separation Principles, construction, working, uses, merits and demerits of Sieve shaker, cyclone separator, Air separator, Bag filter & elutriation tank.
	2	UNIT-II
		A. Heat Transfer: Objectives, applications & Heat transfer mechanisms. Fourier's law, Heat transfer by conduction, convection & radiation. Heat interchangers & heat exchangers.
		 B. Evaporation: Objectives, applications and factors influencing evaporation, differences between evaporation and other heat process. principles, construction, working, uses, merits and demerits of Steam jacketed kettle, horizontal tube evaporator, climbing film evaporator, forced circulation evaporator, multiple effect evaporator& Economy of multiple effect evaporator. C. Distillation: Basic Principles and methodology of simple distillation, flash distillation, fractional distillation, distillation under reduced pressure, steam distillation & molecular distillation





3	UNIT-III		
	A. Drying: Objectives, applications & mechanism of process, measurements & applications of Equilibrium content, rate of drying curve. principles, construction, working merits and demerits of Tray dryer, drum dryer spray dryer, bed dryer, vacuum dryer, freeze dryer. B. Mixing: Objectives, applications & factors affecting Difference between solid and liquid mixing, mechanism mixing, liquids mixing and semisolids mixing. Principles, C. Construction, Working, uses, Merits and Demerits of cone blender, twin shell blender, ribbon blender, Sigma blad planetary mixers, Propellers, Turbines, Paddles & Semulsifier,		
4	UNIT-IV		
	A.Filtration: Objectives, applications, Theories & Factors influe filtration, filter aids, filter medias. Principle, Construction, Working, Merits and demerits of plate & frame filter, filter leaf, rotary drum Meta filter & Cartridge filter, membrane filters and Seidtz filter.		
	B. Centrifugation: principles,	Objectives, princip	ole & applications of Centrifugation,
		_	s and demerits of Perforated basket strifuge, semi continuous centrifuge
5	Corrosion and selected for Phan Theori prevention. Ferro	its prevention: rmaceutical plant codes of corrosion, ous rrous metals, inorg	harmaceutical plant construction, Factors affecting during materials construction, types of corrosion and there ganic and organic non metals, basic
Mode of examination	Theory		
Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE
	10 Marks	15	75





Text book/s*	Introduction to chemical engineering – Walter L Badger & Julius Banchero, Latest edition.
	2. Solid phase extraction, Principles, techniques and applications by Nigel J.K. Simpson- Latest edition.
	3. Unit operation of chemical engineering – Mcabe Smith, Latest edition.
	4. Pharmaceutical engineering principles and practices – C.V.S Subrahmanyam et al., Latest edition.
	5. Remington practice of pharmacy- Martin, Latest edition.
	6. Theory and practice of industrial pharmacy by Lachmann., Latest edition.
	7. Physical pharmaceutics- C.V.S Subrahmanyam et al., Latest edition.
	8. Cooper and Gunn's Tutorial pharmacy, S.J. Carter, Latest edition.





School:		SOP		
Prograi	nme:	B. Pharm		
Branch		Semester: 3		
1	Course Code	BP305 P		
	Course Title	Pharmaceutical organic chemistry II – Practical		
3	Credits	2		
4	Contact Hours (L-T-P)	0-0-4		
	Course Type	Compulsory		
5	Course Objective	Upon completion of the course the student shall be able to:		
		Create the awareness about environmental problems among learners. Impart basic knowledge about the environment and its allied problems.		
		Develop an attitude of concern for the environment. Motivate learner to participate in environment protection and environment improvement. Acquire skills to help the concerned individuals in identifying and solving environmental problems.		
		Strive to attain harmony with Nature.		
6	Course Outcomes	CO1: The students will have the knowledge to identify, name, and write the structure of different aromatic compounds and their derivatives.		
		CO2: The students will be able to understand and explain the mechanism behind the naming reactions of different aromatic compounds and their derivatives.		
		CO3: The students can apply the knowledge to prepare the derivatives of aromaic compounds with different fuctional groups.		
		CO4: Students will analyze the chemical reactions, stabilities of organic compounds and properties of the compounds prepared by them in the laboratory.		
		CO5: Students would evaluate bycomparing compounds prepared by them with standard compounds by chemical and physical properties.		
		CO6: The students will be able to plan about new derivatives based on the above knowledge.		





7	Course				
	Description	1. Experiments involving laboratory techniques			
		a. Recrystallization			
		b. Steam distillation			
		2. Determination of following oil values (including standardization of reagents)			
		a. Acid value			
		b. Saponification value			
		c. Iodine value 3. Preparation Of Compounds			
		Benzanilide/Phenyl benzoate/Acetanilide from Aniline/Phenol/Aniline by acylation reaction.			
		• 2,4,6-Tribromo aniline/Para bromo acetanilide from Aniline/			
		Acetanilide by halogenation (Bromination) reaction.			
		• 5-Nitro salicylic acid/Meta di nitro benzene from Salicylic acid / Nitro benzene by nitration reaction.			
		Benzoic acid from Benzyl chloride by oxidation reaction.			
		Benzoic acid/ Salicylic acid from alkyl benzoate/ alkyl salicylate by hydrolysis reaction.			
		• 1-Phenyl azo-2-napthol from Aniline by diazotization and coupling reactions.			
		Benzil from Benzoin by oxidation reaction.			
		Dibenzal acetone from Benzaldehyde by Claison Schmidt reaction			
		Cinnammic acid from Benzaldehyde by Perkin reaction			
		P-Iodo benzoic acid from P-amino benzoic acid			
8	Outline syllabus				
		Experiments involving laboratory techniques a). Recrystallization b). Steam distillation c). Derivatives of benzene			





	Deterination of follow	ing oil values	
	a). Acid value	C	
	b). Saponification valu	ie	
	c). Iodine value		
3	III Preparation of com	pound	
	a). Benzil	1	
	b. Phenyl benzoate		
	Benzoic acid		
	Oxalic acid		
	c. Rancidity of oils, D	rying oils.	
	Analytical constants –	Acid value, Saponif	fication value, Ester value, Iodine
	value, Acetyl value, R	eichert Meissl (RM)	value – significance and principle
	involved in their deter	mination.	
Mode	of Theory		
examination		1	
Weightage	Continuous Mode	Sessional Exam	ESE
Distribution	Assessment		
	0.5	10	35
Text book/s*	1. Organic Che	emistry by Morrison	and Boyd
		emistry by I.L. Finar	•
	_	• •	by B.S. Bahl & Arun Bahl.
		emistry by P.L.Soni	o, z.e. zam w man zam.
			Mann and Saunders.
		book of Practical O	
	•		nistry by N.K.Vishnoi.





School:		SOP
Progran	nme:	B. Pharm
Branch		Semester: 3
1	Course Code	BP306 P
2	Course Title	Physical pharmaceutics I-Practical
3	Credits	4
4	Contact Hours (L-T-P)	0-0-4
	Course Type	Compulsory
5	Course Objective	Upon the completion of the course student shall be able to Understand various physicochemical properties of drug molecules in the designing the dosage forms Demonstrate use of physicochemical properties in the formulation development and evaluation of dosage forms.
6	Course Outcomes	CO1: The students would be able to describe the various methods of determination of physicochemical properties of drugs and pharmaceuticals. CO2: The students would be able to demonstrate methods for determination of HLB value and Critical Micelle concentration of surfactants. CO3: The students would be able to calculate the value of stability constants in complexation by various methods. CO4: The students would be able to compare various methods of determ ination of stability constants CO5: The students would be able to determine the effect of addition of salt CST and to determine adsorption constants. CO6: The students would be able to determine of Freundlich and Langmuir constants.
7	Course Description	Determination of physicochemical properties of drugs and pharmaceuticals and determination of stability constants, adsorption constants, HLB and CMC values.
8	Outline syllabus	





1	Pharmaceuticals a). Determination of Determination of pk Hasselbalch equatio c). Determination of water d). Determination of	Solubility of drug Xa value by Half Nn. Partition co- effic Partition co- effic Surface tensionof	at room temperature b). eutralization/ Henderson ient of benzoic acid in benzene and eient of Iodine in CCl4 and water given liquids by drop count and	
2	To Determine importants parameter of Surfactants a). Determination of HLB number of a surfactant by saponification method b). Determination of critical micellar concentration of surfactants			
3	To determine stability constants of complexation by various methods a). Determination of stability constant and donor acceptor ratio of PABA- Caffeine complex by solubility method. b). Determination of stability constant and donor acceptor ratio of Cupric- Glycine complex by pH titration method			
4	To study the effect of addition of salt CST and to determine adsorption constants a). Determination of % composition of NaCl in a solution using phenolwater system by CST method. b). Determination of Freundlich and Langmuir constants using activated char coal			
Mode of examination	Practical/Viva			
Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE	
	0 5 Marks	10	35	





,	Text book/s*	Physical Pharmacy by Alfred Martin	
		Experimental Pharmaceutics by Eugene, Parott.	
		Tutorial Pharmacy by Cooper and Gunn.	
		Stocklosam J. Pharmaceutical Calculations, Lea &Febiger, Philadelphia.	
		Liberman H.A, Lachman C., Pharmaceutical Dosage forms, Tablets,	
		Volume-1 to 3, MarcelDekkar Inc.	
		Liberman H.A, Lachman C, Pharmaceutical Dosage forms. Disperse	
		systems, volume 1, 2, 3. Marcel Dekkar Inc.	
		Physical Pharmaceutics by Ramasamy C and ManavalanR.	
		Laboratory Manual of Physical Pharmaceutics, C.V.S. Subramanyam, J.	
		Thimma settee	
		Physical Pharmaceutics by C.V.S. Subramanyam	
		Test book of Physical Phramacy, by Gaurav Jain & Roop K. Khar.	
		•	
	Other		
	References		





School:		SOP
Pr	ogramme:	B. Pharm
Br	anch:	Semester: 3
1	Course Code	BP307 P
2	Course Title	Pharmaceutical microbiology Practical
3	Credits	2
4	Contact Hours (L-T-P)	0-0-4
	Course Type	Compulsory
5	Course Objective	Upon completion of the course the student shall be able to tell about the history, scope of microbiology and describe the structure, morphology and cultivation of microorganism. Student shall identify the bacteria on the basis of various staining technique and importance of sterilization in microbiology. Upon completion of the course the student shall understand the various methods for assessment of antibiotic, test for sterility for preparation. Student shall analyze the source of contamination and their prevention in aseptic areas and importance of cell culture technique.
6	Course Outcomes	CO1: Students shall have knowledge about the various equipment used in experimental microbiology and understand the principle and working of these instruments. CO2: Students shall be able to understand the importance of sterilization in microbiology and apply this knowledge for the preparation of various media. CO3: Students shall acquire complete knowledge of isolation procedure of microorganism (viruses, fungi) and will be able to differentiate microorganism on the basis of various staining technique
		CO4: Students can apply their knowledge for the standardization of antibiotics.
		CO5: Students can apply their knowledge for identification of bacteria.
7	C	CO6: Students can apply their knowledge for nutrient and slab culture.
7	Course Description	To emphasize on definition, types, classification, principles/mechanisms, applications, examples and differences
8	Outline syllal	bus
	1	a). To study various equipment used in microbiology b). To perform the sterilization of glassware by moist heat and dry heat





2	a). Preparation of sterile nutr b). Preparation of sterile nutr			
3 4	a). Study of environmental microflora of various region a). Standardization of antibiotic by cup and plate method Identification of bacteria by gram staining technique identification of bacteria by acid fast staining technique Preparation of nutrient slant and stab culture			
Mode of examination	Practical/Viva			
Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE	
	05	10	35	
Text book/s*	W.B. Hugo and A.D. Russel: Pharmaceutical Microbiology, Blackwell Scientific publications, Oxford London. Prescott and Dunn., Industrial Microbiology, 4 th edition, CBS Publishers & Distributors, Delhi.			
	Pelczar, Chan Kreig, Microbiology, Tata McGraw Hill edn.			
	Rose: Industrial Micro Probisher, Hinsdill et a Cooper and Gunn's: To Peppler: Microbial Tec I.P., B.P., U.S.P lates Ananthnarayan: Text I Edward: Fundamentals N.K.Jain: Pharmaceuti	biology. l: Fundamentals of Micrutorial Pharmacy, CBS Fehnology. t editions. Book of Microbiology, Cas of Microbiology, Cas of Microbiology, Cal Microbiology, Vallal	Publisher and Distribution. Drient-Longman, Chennai	
Other References				
Kelelences				





School:		SOP		
Program	me:	B. Pharm		
Branch:		Semester: 3		
1	Course Code	BP 308 P		
2	Course Title	Pharmaceutical Engineering Practical		
3	Credits	4		
4	Contact Hours (L-T-P)	0-0-4		
	Course Type	Compulsory		
5	Course Objective	Upon completion of the course student shall be able:		
	,	To know various unit operations used in Pharmaceutical industries.		
		To understand the material handling techniques.		
		To perform various processes involved in pharmaceutical manufacturing process.		
		To carry out various test to prevent environmental pollution.		
		To appreciate and comprehend significance of plant lay out design for optimum use of resources.		
		To appreciate the various preventive methods used for corrosion control in Pharmaceutical industries.		
6	Course Outcomes	CO1: Student shall be able to Understand the different factors effecting rate of filtration, evaporation and overall heat transfer coefficient etc.		
		CO2: Students shall be able to predict humidity of air, effect of time on crystallization rate and laws of size reduction.		
		CO3: Students shall be able to calculate uniformity index of given sample, efficiency of steam distillation and construct various size frequency curves, drying curves etc. CO4: Students shall be able to evaluate size distribution of tablet granulations.		
		CO5: Students shall be able to calculate time of crystallization		
		CO6: Students shall be able to study the effect of time on rate of crystallization.		





7				
7	Course Description	science of various un	nit operations used	ndamental knowledge on the art and in pharmaceutical industryand their pharmaceutical unit is emphasized to
8	Outline syllabus	1		
	1	coefficient by heat edistillation. b). Students would by	exchanger and calcone able to construct	e the overall heat transfer ulate the efficiency of steam t drying curves (for calcium noisture content and loss on drying.
a). Students would be able to determine humidry bulb temperatures —use of Dew point method. b). Students would be able to evaluate size disgranulations by using sieving method.			e humidity of air – i) From wet and int method.	
	3	mill using Ball Mill b). Students would b and Filtration. c). Students would b	oe able to relate fac	e laws of size reduction using ball ctors affecting Rate of Evaporation and the working of major equipment
	4	a). Students would be able to study the effect of time on the Rate Crystallization b). Students would be able to calculate the uniformity Index for g sample by using Double Cone Blender		
	Mode of examination	Practical/Viva		
	Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE
		0 5 Marks	10	35
_				





Text book/s	1. Introduction to chemical engineering – Walter L Badger & Julius Banchero, Latest edition.
	2. Solid phase extraction, Principles, techniques and applications by Nigel J.K. Simpson- Latest edition.
	3. Unit operation of chemical engineering – Mcabe Smith, Latest edition.
	4. Pharmaceutical engineering principles and practices – C.V.S Subrahmanyam et al., Latest edition.
	5. Remington practice of pharmacy- Martin, Latest edition.
	6. Theory and practice of industrial pharmacy by Lachmann., Latest edition.
	7. Physical pharmaceutics- C.V.S Subrahmanyam et al., Latest edition.
	8. Cooper and Gunn's Tutorial pharmacy, S.J. Carter, Latest edition.





Programme: Branch: Semester: IV Course Code	School:		SOP
Branch: BP401T Code			B.Pharm
Code Course Title			Semester: IV
Course Title	1	Course	BP401T
Title 3 Credits 4 Contact Hours (L-T-P) Course Type 5 Course Objective 1		Code	
3 Credits 4	2	Course	Pharmaceutical Organic Chemistry III - Theory
4 Contact Hours (L-T-P) Course Objective 5 Course Objective 6 Course Objective 6 Course Objective 7 Course Objective 7 Course Objective 6 Course Outcomes 7 Course Outcomes 7 Course Outcomes 8 CO1: Students shall be able to understand the configuration to optical and geometrical isomers and diasteriomers. CO2: Students shall acquire the knowledge of properties of enantiomers and on the basis of this knowledge students can separate the desired isomeric form. CO3: Students shall be able to do nomenclature of heterocyclic compounds and draw the structure of heterocyclic compounds. CO4: students shall gain the knowledge of various heterocyclic compounds and draw the structure of heterocyclic and acquire the knowledge of their synthesis, chemical reactions and their applications in medicines. CO5: The students will be able to understand and explain the mechanism behind various naming reactions and acquire the knowledge of their applications in preparation of various drugs and intermediates. CO6: Students shall be able to study the SAR of compounds. This subject imparts knowledge on stereo-chemical aspects of organic compounds and organic reactions, important named reactions, chemistry of important hetero cyclic compounds. It also emphasizes on medicinal and other uses of organic compounds.			
Hours (L-T-P) Course Type 5 Course Objective Upon completion of the course the student shall be able to 1. 1. understand the methods of preparation and properties of organic compounds 2. explain the stereo chemical aspects of organic compounds and stereo chemical reactions 3. 3. know the medicinal uses and other applications of organic compounds COurse Outcomes CO1: Students shall be able to understand the configuration to optical and geometrical isomers. They also get the knowledge of properties of enantiomers and geometrical isomers and diasteriomers. CO2: Students shall acquire the knowledge of separation of different isomers and on the basis of this knowledge students can separate the desired isomeric form. CO3: Students shall be able to do nomenclature of heterocyclic compounds and draw the structure of heterocyclic compounds. CO4: students shall gain the knowledge of various heterocyclic compounds in terms of their synthesis, chemical reactions and their applications in medicines. CO5: The students will be able to understand and explain the mechanism behind various naming reactions and acquire the knowledge of their applications in preparation of various drugs and intermediates. CO6: Students shall be able to study the SAR of compounds. This subject imparts knowledge on stereo-chemical aspects of organic compounds and organic reactions, important named reactions, chemistry of important hetero cyclic compounds. It also emphasizes on medicinal and other uses of organic compounds.	-		
Course Type	4		3-1-0
Course Objective Upon completion of the course the student shall be able to 1. 1. understand the methods of preparation and properties of organic compounds 2. explain the stereo chemical aspects of organic compounds and stereo chemical reactions 3. 3. know the medicinal uses and other applications of organic compounds Course Outcomes CO1: Students shall be able to understand the configuration to optical and geometrical isomers. They also get the knowledge of properties of enantiomers and geometrical isomers and diasteriomers. CO2: Students shall acquire the knowledge of separation of different isomers and on the basis of this knowledge students can separate the desired isomers form. CO3: Students shall be able to do nomenclature of heterocyclic compounds and draw the structure of heterocyclic compounds. CO4: students shall gain the knowledge of various heterocyclic compounds in terms of their synthesis, chemical reactions and their applications in medicines. CO5: The students will be able to understand and explain the mechanism behind various naming reactions and acquire the knowledge of their applications in preparation of various drugs and intermediates. CO6: Students shall be able to study the SAR of compounds. This subject imparts knowledge on stereo-chemical aspects of organic compounds and organic reactions, important named reactions, chemistry of important hetero cyclic compounds. It also emphasizes on medicinal and other uses of organic compounds.			
Type 5 Course Objective 1. 1. understand the methods of preparation and properties of organic compounds 2. explain the stereo chemical aspects of organic compounds and stereo chemical reactions 3. 3. know the medicinal uses and other applications of organic compounds 6 Course Outcomes CO1: Students shall be able to understand the configuration to optical and geometrical isomers. They also get the knowledge of properties of enantiomers and geometrical isomers and diasteriomers. CO2: Students shall acquire the knowledge of separation of different isomers and on the basis of this knowledge students can separate the desired isomeric form. CO3: Students shall be able to do nomenclature of heterocyclic compounds and draw the structure of heterocyclic compounds. CO4: students shall gain the knowledge of various heterocyclic compounds in terms of their synthesis, chemical reactions and their applications in medicines. CO5: The students will be able to understand and explain the mechanism behind various naming reactions and acquire the knowledge of their applications in preparation of various drugs and intermediates. CO6: Students shall be able to study the SAR of compounds. 7 Course Description This subject imparts knowledge on stereo-chemical aspects of organic compounds and organic reactions, important named reactions, chemistry of important hetero cyclic compounds. It also emphasizes on medicinal and other uses of organic compounds.			
Description Upon completion of the course the student shall be able to			Compulsory
Objective 1. 1. understand the methods of preparation and properties of organic compounds 2. explain the stereo chemical aspects of organic compounds and stereo chemical reactions 3. 3. know the medicinal uses and other applications of organic compounds CO1: Students shall be able to understand the configuration to optical and geometrical isomers. They also get the knowledge of properties of enantiomers and geometrical isomers and diasteriomers. CO2: Students shall acquire the knowledge of separation of different isomers and on the basis of this knowledge students can separate the desired isomeric form. CO3: Students shall be able to do nomenclature of heterocyclic compounds and draw the structure of heterocyclic compounds. CO4: students shall gain the knowledge of various heterocyclic compounds in terms of their synthesis, chemical reactions and their applications in medicines. CO5: The students will be able to understand and explain the mechanism behind various naming reactions and acquire the knowledge of their applications in preparation of various drugs and intermediates. CO6: Students shall be able to study the SAR of compounds. This subject imparts knowledge on stereo-chemical aspects of organic compounds and organic reactions, important named reactions, chemistry of important hetero cyclic compounds. It also emphasizes on medicinal and other uses of organic compounds.	5		Unan completion of the course the student shall be able to
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8 Outline syllabus			uses of organic compounds.
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1	UNIT-I 10
	Hours
	A. Stereo isomerism
	Optical isomerism –Optical activity, enantiomerism, diastereoisomerism, meso compounds Elements of symmetry, chiral and achiral molecules B. DL system of nomenclature of optical isomers, sequence rules, RS system of nomenclature of optical isomersReactions of chiral molecules
	Racemic modification and resolution of racemic mixture.
	C. Asymmetric synthesis: partial and absolute
2	
	UNIT-II Hours
	A. Geometrical isomerism
	Nomenclature of geometrical isomers (Cis Trans, EZ, Syn Anti systems)
	Methods of determination of configuration of geometrical isomers.
	B. Conformational isomerism in Ethane, n-Butane and Cyclohexane.
	Stereo isomerism in biphenyl compounds (Atropisomerism) and conditions for optical activity. Stereospecific and stereoselective reactions
3	Unit III
	A. Heterocyclic compounds:
	Nomenclature and classification
	B. Synthesis, reactions and medicinal uses of following compounds/derivatives
	Pyrrole, Furan, and Thiophene
	C. Relative aromaticity and reactivity of Pyrrole, Furan and Thiophene





4	UNIT-IV		www.sharda.ac.in		
	A. Synthesis, react	ions and medicina	l uses of following compounds/derivatives		
	Pyrazole,				
	B. Imidazole, Oxaz	zole and Thiazole.	Pyridine, Quinoline, Isoquinoline, Acridine		
	and Indole. Basi	city of pyridine Sy	nthesis		
			ne, azepines and their derivatives		
5					
	UNIT-V				
	A. Reactions of sy	nthetic importanc	e		
	Metal hydride reduction, Wolff Kis		d LiAlH ₄), Clemmensen reduction, Birch		
	B. Oppenauer-oxid	ation and Dakin rea	action.		
	C. Beckmanns rea	rrangement and	Schmidt rearrangement. Claisen-Schmidt		
	condensation				
Mode of examinat ion	Theory				
Weighta	Continuous Mode	Sessional Exam	ESE		
ge	Assessment				
Distribut ion	10 Marks	15	75		
Text book/s*		ic chemistry by I.L	. Finar, Volume-I & II.		
OOOK 5	2. A text book of organic chemistry – Arun Bahl, B.S. Bahl.				
	3. Heterocyclic Chemistry by Raj K. Bansal				
	4. Organ	ic Chemistry by M	orrison and Boyd		
	5. Hetero	ocyclic Chemistry b	y T.L. Gilchrist		
Other					
Referenc					
es					





Course Code BP402T	School:		SOP		
Course Code BP402T	Programme:		B.Pharm		
Course Title Medicinal chemistry I - Theory Cordits 4 Contact Hours (L-T-P) Course Type Objective Upon completion of the course the student shall be able to 1. understand the chemistry of drugs with respect to their pharmacologica activity 2. understand the drug metabolic pathways, adverse effect and therapeutic value of drugs know the Structural Activity Relationship (SAR) of different class of drugs Write the chemical synthesis of some drugs CO1: The students will be able to apply knowledge to identify, name and classify the different categories of drugs with respect to their pharmacological activities. CO2: The students will be able to understand and explain the structure activity relationship, drug metabolic pathways, adverse effects and their therapeutic activity of different categories of drugs. CO3: The students will be able to apply the knowledge to construct the chemical synthesis of some drugs. CO4: The students will be able to analyze chemical reactions, stabilities of compounds and properties of the compounds prepared by them in the laboratory. CO5: The students would be able to illustrate different drugs acting of Nervous system. CO6: The students would be able to evaluate various transmitters. This subject is designed to impart fundamental knowledge on the structure chemistry and therapeutic value of drugs. The subject emphasizes of structure activity relationships of drugs, importance of physicochemical properties and metabolism of drugs. The syllabus also emphasizes of chemical synthesis of important drugs under each class.	Branch:		Semester: IV		
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	8	Outline syllabu	ls		





1	UNIT- I		
	A. Introduction to Medicinal Chemistry History and development of medicinal chemistry Physicochemical properties in relation to biological action Ionization, Solubility, Partition Coefficient, Hydrogen bonding, Protein binding, Chelation, Bioisosterism, Optical and Geometrical isomerism.		
	B. Drug metabolism		
	a). Drug metabolism principles- Phase I and Phase II.		
	b). Factors affecting drug metabolism including stereo chemical aspects.		
2	UNIT- II A. Drugs acting on Autonomic Nervous System Adrenergic Neurotransmitters:		
	a). Biosynthesis and catabolism of catecholamine.		
	b). Adrenergic receptors (Alpha & Beta) and their distribution.		
	B. Sympathomimetic agents: SAR of Sympathomimetic agents		
	a). Direct acting: Nor-epinephrine, Epinephrine, Phenylephrine*, Dopamin Methyldopa, Clonidine, Dobutamine, Isoproterenol, Terbutaline, Salbutamol Bitolterol, Naphazoline, Oxymetazoline and Xylometazoline.		
	Indirect acting agents: Hydroxyamphetamine, Pseudoephedrine, Propylhexedrine.		
	Agents with mixed mechanism: Ephedrine, Metaraminol.		
	C. Adrenergic Antagonists:		
	Alpha adrenergic blockers: Tolazoline*, Phentolamine, Phenoxybenzamine, Prazosin, Dihydroergotamine, Methysergide.		
	D. Beta adrenergic blockers: SAR of beta blockers, Propranolol*, Metibranolol, Atenolol, Betazolol, Bisoprolol, Esmolol, Metoprolol, Labetolol, Carvedilol.		





3 UNIT-III

A. Cholinergic neurotransmitters:

- a). Biosynthesis and catabolism of acetylcholine.
- b). Cholinergic receptors (Muscarinic & Nicotinic) and their distribution.
- c). Parasympathomimetic agents: SAR of Parasympathomimetic agents
- B. **Direct acting agents:** Acetylcholine, Carbachol*, Bethanechol, Methacholine, Pilocarpine.
- C. Indirect acting/ Cholinesterase inhibitors (Reversible & Irreversible): Physostigmine, Neostigmine*, Pyridostigmine, Edrophonium chloride, Tacrine hydrochloride, Ambenonium chloride, Isofluorphate, Echothiophate iodide, Parathione, Malathion.

Cholinesterase reactivator: Pralidoxime chloride.

Cholinergic Blocking agents: SAR of cholinolytic agents

Solanaceous alkaloids and analogues: Atropine sulphate, Hyoscyamine sulphate, Scopolamine hydrobromide, Homatropine hydrobromide, Ipratropium bromide*.

Synthetic cholinergic blocking agents: Tropicamide, Cyclopentolate bromide, Dicyclomine hydrochloride*, hydrochloride, Clidinium Glycopyrrolate, Methantheline bromide, Propantheline bromide, Benztropine Biperidine hydrochloride, Orphenadrine citrate, Procyclidine hydrochloride*, Tridihexethyl chloride, Isopropamide iodide, Ethopropazine hydrochloride.





4

UNIT- IV

Drugs acting on Central Nervous System

A. Sedatives and Hypnotics:

Benzodiazepines: SAR of Benzodiazepines, Phenobarbital, Mephobarbital, Amobarbital, Butabarbital, Pentobarbital, Secobarbital

B. Miscelleneous:

Amides & imides: Glutethmide.

Alcohol & their carbamate derivatives: Meprobomate, Ethchlorvynol. Aldehyde & their derivatives: Triclofos sodium, Paraldehyde.

C. Antipsychotics

Phenothiazeines: SAR of Phenothiazeines - Promazine hydrochloride, Chlorpromazine hydrochloride*, Triflupromazine, Thioridazine hydrochloride, Piperacetazine hydrochloride, Prochlorperazine maleate, Trifluoperazine hydrochloride.

Ring Analogues of Phenothiazeines: Chlorprothixene, Thiothixene, Loxapine succinate, Clozapine.

Fluro buterophenones: Haloperidol, Droperidol, Risperidone.

Beta amino ketones: Molindone hydrochloride.

Benzamides: Sulpieride.

Chlordiazepoxide, Diazepam*, Oxazepam, Chlorazepate, Lorazepam, Alprazolam, Zolpidem

Barbiturtes: SAR of barbiturates, Barbital*,





5	UNIT - V Drugs acting on Central Nervous System A. General anesthetics:		
	a). Inhalation anesthe Sevoflurane, Isoflurane, De		*, Methoxyflurane, Enflurane,
	b). Ultra short acting sodium, Thiopental sodium.		Methohexital sodium*, Thiamylal
	c). Dissociative anesthetics	s: Ketamine hydro	chloride.*
	B. Narcotic and non-narc	cotic analgesics	
	sulphate, Codeine, Mepe Diphenoxylate hydrochlori	eridine hydrochl ide, Loperamide	of Morphine analogues, Morphine oride, Anilerdine hydrochloride, hydrochloride, Fentanyl citrate*, ne hydrochloride, Pentazocine,
	b). Narcotic antagonists: Naloxone hydrochloride.	Nalorphine hyd	rochloride, Levallorphan tartarate,
	c). Anti-inflammatory agents: Sodium salicylate, Aspirin, Mefenamic acid*, Meclofenamate, Indomethacin, Sulindac, Tolmetin, Zomepriac, Diclofenac, Ketorolac, Ibuprofen*, Naproxen, Piroxicam, Phenacetin, Acetaminophen, Antipyrine, Phenylbutazone.		
	C.Actions of synthetic importance		
	a). Metal hydride reduction (NaBH ₄ and LiAlH ₄), Clemmensen reduction, Birch reduction, Wolff Kishner reduction.		
	b). Oppenauer-oxidation and	d Dakin reaction.	
	c). Beckmanns rearrangen	nent and Schmid	t rearrangement. Claisen-Schmidt
	condensation		
Mode of examinati on	Theory		
Weightag	Continuous Mode	Sessional	ESE
e Distributi	Assessment	Exam 15	75
on	10 Marks		





Text book/s*	Chemistry.	Wilson and Giswold's Organic medicinal and Pharmaceutical
		Foye's Principles of Medicinal Chemistry.
		Burger's Medicinal Chemistry, Vol I to IV.
		Introduction to principles of drug design- Smith and Williams.
	•	Remington's Pharmaceutical Sciences.
		Martindale's extra pharmacopoeia.
	•	Organic Chemistry by I.L. Finar, Vol. II.
	•	The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1-5.
	•	Indian Pharmacopoeia.
	•	Text book of practical organic chemistry- A.I.Vogel.
Other		
Reference		
S		





		www.sharda.ac.in
School:		SOP
Programme:		B. Pharm
Br	anch:	Semester: IV
1	Course Code	BP403T
2	Course Title	Physical Pharmaceutics II - Theory
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Compulsory
5	Course Objective	Upon the completion of the course student shall be able to 1. Understand various physicochemical properties of drug molecules in the designing the dosage form 2. Know the principles of chemical kinetics & to use them in assigning expiry date for Formulation 3. Demonstrate use of physicochemical properties in evaluation of dosage forms. Appreciate physicochemical properties of drug molecules in formulation research and Development
6	Course Outcomes	CO1: Students would be able to understand the concept of reaction kinetics, degradation pathways, factor effects stability of drugs. CO2: Students would be able to understand flow of liquid, law of flow, determination of viscosity of liquid by viscometer. CO3: Students would be able to apply the basics of surface and interfacial tension, surface active agents. CO4: Students would be able to describe properties of powder like particle size and distribution, determining particle size by different methods. CO5: Students would be able to evaluate about the colloidal dispersion, role of particle size and shape in colloidal dispersion. CO6: Students would be able to illustrate the Stabilization of medicinal agents.
7	Course Description	The course deals with the various physica and physicochemical properties, and principles involved in dosage forms/formulations. Theory and practical components of the subject help the student to get a better insight into various areas of formulation research and development, and stability studies of pharmaceutical dosage forms.
8	Outline syllabu	as





1	UNIT-I 07 Hours
	A. Colloidal dispersions: Classification of dispersed systems & their general characteristics, size & shapes of colloidal particles, classification of colloids & comparative account of their general properties. Optical, kinetic
	B. B. electrical properties.
	C. Effect of electrolytes, coacervation, peptization& protective action.
2	UNIT-II 10 Hours
	A. Rheology: Newtonian systems, law of flow, kinematic viscosity, effect of temperature, non-Newtonian systems, pseudoplastic, dilatant, plastic, thixotropy, thixotropy in formulation, determination of viscosity, capillary, falling Sphere, rotational viscometers
	B. Deformation of solids: Plastic and elastic deformation,
	C. Heckel equation, Stress, Strain, Elastic Modulus
3	UNIT-III 10 Hours
	A. Coarse dispersion: Suspension, interfacial properties of suspended particles, settling in suspensions, formulation of flocculated and deflocculated suspensions.
	B. B. Emulsions and theories of emulsification, microemulsion and multiple emulsions;
	C. Stability of emulsions, preservation of emulsions, rheological properties of emulsions and emulsion formulation by HLB method.
4	UNIT IV
	 A. Micromeretics: Particle size and distribution, mean particle size, number and weight distribution, B. particle number, methods for determining particle size by different methods,
	counting and separation method, particle shape, specific surface, methods for determining surface area, permeability,
	C. adsorption, derived properties of powders, porosity, packing arrangement, densities, bulkiness & flow properties.





_				
5	UNIT-V		10 Hours	
	A. Drug stability: Reaction kinetics: zero, pseudo-zero, first & second order units of basic rate constants, determination of reaction order. Physical chemical factors influencing the chemical degradation of pharmaceut product: temperature, solvent, ionic strength, dielectric constant, specific general acid base catalysis, Simple numerical problems.			
	B. Stabilizati oxidation.	6 6	inst common reactions like hydrolysis &	
		ed stability testing in expi otolytic degradation and its	ration dating of pharmaceutical dosage prevention	
Mode of examination	Theory			
Weightage Distribution	Continuous Mode	Sessional Exam	ESE	
	Assessment			
	10 Marks	15	75	
Text book/s*	1. Physical Pharmacy by Alfred Martin, Sixth edition			
	2. Experimental	pharmaceutics by Eugene,	Parott.	
	3. Tutorial phar	macy by Cooper and Gunn.		
	4. Stocklosam J	. Pharmaceutical calculation	ns, Lea & Febiger, Philadelphia.	
	5. Liberman H to 3, Marcel l		atical Dosage forms, Tablets, Volume-1	
		A, Lachman C, Pharmace 3. Marcel Dekkar Inc.	utical dosage forms. Disperse systems,	
	7. Physical Phan	maceutics by Ramasamy C	c, and Manavalan R.	



School:		SOP
Pı	ogramme:	B. Pharm
Branch:		Semester: IV
1	Course	BP404T
	Code	
2	Course	Pharmacology I - Theory
	Title	
3	Credits	4
4	Contact	3-1-0
	Hours	
	(L-T-P)	
	Course	Compulsory
_	Type	
5	Course Objective	Upon completion of this course the student should be able to
		Understand the pharmacological actions of different categories of drugs Explain the mechanism of drug action at organ system/sub cellular/ macromolecular levels. Apply the basic pharmacological knowledge in the prevention and treatment of various diseases. Observe the effect of drugs on animals by simulated experiments
		Appreciate correlation of pharmacology with other bio medical sciences
6	Course Outcomes	CO1: Students will able to understand the pharmacological actions of different categories of drugs. CO2: Students will able to explain the mechanism of drug action at organ system/sub cellular/macromolecular levels. CO3: Students will able to apply the basic pharmacological knowledge in the prevention and treatment of various diseases. CO4: Students will able to illustrate the effect of drugs on animals by simulated experiments. CO5: Students will able to apply the correlation of pharmacology with other biomedical Sciences. CO5: Students will able to evaluate the mechanisms and action of drugs under CNS CO6: Students will able to evaluate the mechanisms and action of drugs.





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7	Course Descriptio n	The main purpose of the subject is to understand what drugs do to the living organisms and how their effects can be applied to therapeutics. The subject covers the information about the drugs like, mechanism of action, physiological and biochemical effects (pharmacodynamics) as well as absorption, distribution, metabolism and excretion (pharmacokinetics) along with the adverse effects, clinical uses, interactions, doses, contraindications and routes of administration of different classes of drugs.	
8	Outline syll	abus	
	1	UNIT-I A. General Pharmacology	
		 a. Introduction to Pharmacology- Definition, historical landmarks and scope of pharmacology, nature and source of drugs, essential drugs concept and routes of drug administration, receptors, addiction, tolerance, dependence, tachyphylaxis, idiosyncrasy, allergy. b. Pharmacokinetics- Membrane transport, absorption, distribution, metabolism and excretion of drugs .Enzyme induction, enzyme inhibition, kinetics of elimination 	
		c.Agonists, antagonists(competitive and non competitive), spare receptors,	
addiction, tolerance, dependence, tachyphylaxis,		addiction, tolerance, dependence, tachyphylaxis,	
2 UNIT-II Hours			
		A. General Pharmacology Pharmacodynamics- Principles and mechanisms of drug action. Receptor theories and classification of receptors, regulation of receptors. drug receptors interactions signal transduction mechanisms, G-protein—coupled receptors, ion channel receptor, transmembrane enzyme linked receptors, transmembrane JAK-STAT binding receptor and receptors that regulate transcription factors, dose response relationship, therapeutic index, combined effects of drugs and factors modifying drug action. a. Adverse drug reactions. b. Drug interactions (pharmacokinetic and pharmacodynamic) Drug discovery and clinical evaluation of new drugs -Drug discovery phase, preclinical evaluation phase, c. clinical trial phase, phases of clinical trials and pharmacovigilance.	





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3	UNIT-III		10	
	Hours			
	Pharmacology of drugs acting on peripheral nervous system			
	NS.			
	Neurohumoral tran	smission, co-transmission	n and classification of neurotransmitters.	
	Parasympathomim	etics, Parasympatholytics	s, Sympathomimetics, sympatholytics.	
	Neuromuscular blo	ocking agents and skeleta	l muscle relaxants (peripheral).	
	B. Local a	nesthetic agents.		
	C. Drugs u	ised in myasthenia gravis	and glaucoma	
4	UNIT-IV		08	
	Hours			
	Pharmacology of	drugs acting on central	nervous system	
			C.N.S.special emphasis on importance of ABA, Glutamate, Glycine, serotonin,	
	b. General an	nesthetics and pre-anesth	etics.	
	c. Sedatives, Alcohols and disu	• •	acting muscle relaxants. Anti-epileptics,	
5	UNIT-V		07	
	Hours		,	
		drugs acting on central	•	
		armacological agents: ti-manics and hallucinoge	Antipsychotics, antidepressants, antiens.	
	b. Drugs use	d in Parkinsons disease a	nd Alzheimer's disease.	
	c. CNS stime	ulants and nootropics.		
	Opioid analgesics and antagonists			
	Drug addiction, dru	ug abuse, tolerance and d	ependence.	
			•	
Mode of	Theory			
examination	Cantinus	Cassianal Every	ECE	
Weightage Distribution	Continuous Mode	Sessional Exam	ESE	
DIBUIOUUOII	Assessment			
	10 Marks	15	75	
	i	i	1	



Text book/s*	1. Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology, Churchil Livingstone Elsevier
	2. Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata Mc Graw-Hill
	Goodman and Gilman's, The Pharmacological Basis of Therapeutics
	4. Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A. K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins
	5. Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews- Pharmacology
Other	
References	





School:		SOP		
Pr	ogramme:	B.Pharm		
Branch:		Semester: IV		
1	Course Code	BP405T		
2	Course Title	Pharmacognosy and Phytochemistry I - Theory		
3	Credits	4		
4	Contact	3-1-0		
	Hours			
	(L-T-P)			
	Course Type	Compulsory		
5	Course	Upon the completion of the course student shall be able to		
	Objective	Understand the techniques in the cultivation and production of crude drugs.		
		Identify the crude drugs, their uses and chemical nature.		
		Understand the evaluation techniques for the herbal drugs.		
		Carry out the microscopic and morphological evaluation of crude drugs		
6	Course	CO1: Students shall be able to define pharmacognosy, identify the sources of crude		
		drugs, and describe type of adulteration, evaluation of crude drugs, cultivation		
techniques, various medicine systems and plant tissue culture.				
		CO2: Students will be able to classify the crude drugs, understand their properties,		
		chemical nature and uses and are able to distinguish drugs with the help of chemical		
		tests and describe various cultivation techniques.		
		CO3: Students will be able to apply their knowledge in identification, cultivation, evaluation of drugs, and prescribing the crude drug for various health issues.		
		CO4: Students will analyze the crude drugs and its chemical nature and their		
		activities.		
		CO5: Students would be able to compare two drugs with the help of chemical and		
		physical properties, and evaluate them for their quality.		
		CO6: Students will be able to study plant tissue culture techniques.		
7	Course	The subject involves the fundamentals of Pharmacognosy like scope, classification		
	Description	of crude drugs, their identification and evaluation, phytochemicals present in them		
		and their medicinal properties.		
8	Outline syllab	1 1		





1	UNIT-I 10
	Hours
	A. Introduction to Pharmacognosy:
	a. Definition, history, scope and development of Pharmacognosy
	b. Sources of Drugs – Plants, Animals, Marine & Tissue culture
	c. Organized drugs, unorganized drugs (dried latex, dried juices, dried extracts, gums and mucilages, oleoresins and oleo- gum -resins).
	B. Classification of drugs:
	Alphabetical, morphological, taxonomical, chemical, pharmacological, chemo and sero taxonomical classification of drugs
	C. Quality control of Drugs of Natural Origin:
	a. Adulteration of drugs of natural origin. Evaluation by organoleptic, microscopic, physical, chemical and biological methods and properties.
	b. Quantitative microscopy of crude drugs including lycopodium spore method, leafconstants, camera lucida and diagrams of microscopic objects to scale with camera lucida.
2	UNIT-II 10
	 Hours A. Cultivation, Collection, Processing and storage of drugs of natural origin: a. Cultivation and Collection of drugs of natural origin Factors influencing cultivation of medicinal plants. Plant hormones and their applications. b. Polyploidy, mutation and hybridization with reference to medicinal plants
	B. Conservation of medicinal plants
3	UNIT-III 07
	Hours
	A. Plant tissue culture:
	Historical development of plant tissue culture,
	B. types of cultures, Nutritional requirements, growth and their maintenance.
	C. Applications of plant tissue culture in pharmacognosy. Edible vaccines
4	UNIT V
	Hours
	A. Study of biological source, chemical nature and uses of drugs of natural origin containing following drugs
	B. Plant Products:
	Fibers - Cotton, Jute, Hemp
	Hallucinogens, Teratogens, Natural allergens





	5	UNIT V			08	
		Hours				
		Study of biological source, chemical nature and uses of drugs of natural origin				
		containing following	drugs			
		A. Plant Products:				
		Fibers - Cotton, Jute	_			
		Hallucinogens, Terat	togens, Natural allerg	ens		
		B. Primary metabo	olites:			
		General introduction	, detailed study with	respect to chemistry,		
		evaluation, preserva	ation, storage, there	respect to chemistry, sources, preparageutic used and commercial utility the following Primary metabolites:		
		C. Carbohydrates:	Acacia, Agar, Tragac	eanth, Honey		
		Proteins and Enzy	mes : Gelatin, casei	n, proteolytic enzymes (Papain, bro	melain.	
			okinase, streptokinase	, 1	moram,	
		1 1	· •	il, Chaulmoogra oil, Wool Fat, Bees V	Vax	
		Marine Drugs:				
		Novel medicinal age	nts from marine source	ces		
	Mode of	Theory				
	examinatio					
	n Waishasa	C C M	C 1 E	ECE		
	Weightage Distributio	Continuous Mode	Sessional Exam	ESE		
	n	Assessment 10 Marks	15	75		
-	Text			ognosy, 16th edition, W.B. Sounders	& Co	
	book/s*	London, 2009.	and Evans Fhaimacc	ignosy, four edition, w.b. Sounders	& Co.,	
		Tyler, V.E., Brady, I	L.R. and Robbers, J.E	., Pharmacognosy, 9th Edn., Lea and I	Febiger,	
		Philadelphia, 1988.				
		Text Book of Pharm	acognosy by T.E. Wa	llis		
		Mohammad Ali. Pharmacognosy and Phytochemistry, CBS Publishers & Distribution, New Delhi.				
		Text book of Pharm	acognosy by C.K. Ko	okate, Purohit, Gokhlae (2007), 37th	Edition,	
		Nirali Prakashan, New Delhi.				
		Herbal drug industry by R.D. Choudhary (1996), Ist Edn, Eastern Publisher, New Delhi.				
		Essentials of Pharmacognosy, Dr.SH.Ansari, IInd edition, Birla publications, New				
		Delhi, 2007	nosvi C V Valzata D	urahit Gakhlaa		
			nosy: C.K. Kokate, P Prugs by M.A. Iyenga			
L		1 Mawiny of Clude L	riugo by ivi.A. Tyciiga	1		





Other	
References	





School:		SOP			
Program	me:	B. Pharm			
Branch:		Semester: 4			
1 Course Code		BP406 P			
2	Course Title	Medicinal chemistry-i (practial)			
3	Credits	2			
4	Contact Hours (L-T-P)	0-0-4			
	Course Type	Compulsory			
5	Course Objective	Upon completion of the course, the student shall be able to -Know the classification and salient features of five kingdoms of life			
	Ü	-Understand the basic components of anatomy & physiology of plant -Know understand the basic components of anatomy & physiology animal with special reference to human			
6	Course				
	Outcomes	CO1: Students will be able to understand about practical laboratory skills and get hands-on experience of modern scientific instrumentation and methodology, particularly in relation to the chemistry of pharmaceuticals			
		CO2: Students will be able to apply knowledge and understanding of the fundamental principles of chemistry and their applications to pharmaceuticals.			
		CO3: Students will be able to use and apply their skills and methodology to a range of techniques used in pharmaceutical chemistry.			
		CO4: Students will be able to generalize professional transferable skills as exemplified by problem solving and teamwork.			
		CO5: Students will be able to predict the skills to make synthetic scheme for certain reactions involved in synthesis of drugs.			
		CO6: Students will be able to apply knowledge about different analytical methods to establish qualitative as well as quantitative reports about the chemical entity.			
Describition 1		Preparation of drugs/ intermediates, assay of drugs and determination of Partition coefficient.			
8	Outline syllabus				
	1	I Preparation of drugs/ intermediates			
		1,3-pyrazole 1,3-oxazole Benzimidazole			





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2	Benztriazole			
	2,3- diphenyl quin	oxaline		
	Benzocaine			
	Phenytoin			
	Phenothiazine			
3	Assay of drugs			
	Barbiturate			
	II Assay of drugs			
	Chlorpromazine			
	Phenobarbitone			
	Atropine			
	Ibuprofen			
4	Determination of	Partition coeffi	cient for any two drugs	
) / 1 C	D .: 1/77			
Mode of examination	Practical/Viva			
Weightage	Continuous	Sessional	ESE	
Distribution	Mode	Exam		
	Assessment			
	05	10	35	
Text book/s*				
	5. Remington's Pharmaceutical Sciences. Martindale's extra pharmacopoeia			
0.1	iviartingale's extra	i pnarmacopoeia		
Other				
References				





School:		SOP		
Prograi	nme:	B. Pharm		
Branch:		Semester: 4		
1	Course Code	BP407P		
2	Course Title	Physical pharmaceutics II (Practical)		
3	Credits	2		
4	Contact Hours (L-T-P)	0-0-4		
	Course Type	Compulsory		
5	Course Objective	Upon completion of the course, the student shall be able to -Know the classification and salient features of five kingdoms of life		
		-Understand the basic components of anatomy & physiology of plant -Know understand the basic components of anatomy & physiology animal with special reference to human		
6	Course Outcomes	CO1: Thestudents would be able to describe the derived properties of powder like angle of repose, bulk density, true density and porosity.		
		CO2: Thestudents would be able to demonstrate methods for determination of HLB value and Critical Micelle concentration of surfactants.		
		CO3: The students would be able to analyze the particle size, particle size distribution by using methods like Sieving and Microscopic.		
		CO4: The students would be able to describe the viscosity, effect of sedimentation on suspension		
		CO5: The students would be able to plan about rate of reaction and accelerated stability studies according to ICH guidelines.		
		CO6: The students would be able to compare different types of stability parameters.		
7	Course Description	Determination of particle size, particle size distribution, derived properties of powder, viscosity, effect of suspending agent on sedimentation volume, factors affecting viscosity, viscosity determination and various stability studies as per ICH guidelines		
8	Outline syllabus			





		TOTAL MELACITI
	1	Determination of particle size, particle size distribution using
		sieving method.
		2. Determination of particle size, particle size distribution using microscopic method.
		3. Determination of bulk density, true density and porosity.
		4. Determine angle of repose and influence of lubricant on angle of
		repose
		5. Determination of viscosity of liquid using Ostwald's viscometer
		6. Determination sedimentation volume with effect of different suspending agent
		7. Determination of sedimentation volume with effect of different
		concentration of single suspending agent
		8. Determination of viscosity of semisolid by using Brookfield
		viscometer
		9. Determination of reaction rate constant first order.
		10. Determination of reaction rate constant first order.
		Accelerated stability studies
	2	Determination of particle size, particle size distribution using sieving
		method& microscopic method
		I. Determination of particle size
		II. Determination of particle size distribution
		III. Using the sieving method
	3	To determine the derived properties of powder
		Determine the bulk density, true density and porosity of the
		powder
		Determine the flow properties of powder
		Determine the effect of glidants on flow properties of powder
	4	Determination of viscosity of liquids & semi solids
		 Determination of viscosity of liquid by using Ostwald's
		viscometer
		Determination of viscosity of different concentration of
		glycerineby using Ostwald's viscometer
	5	Determination of sedimentation volume of suspension
		I. Determination sedimentation volume with effect of different
		suspending agent
		II. Determination of sedimentation volume with
	Mode of	
	examination	
L	1	1





Weightage	Continuous	Sessional Exam	ESE
Distribution	Mode		
	Assessment		
	05	10	35
Text book/s*	1. Physical Phar	macy by Alfred M	artin, Sixth edition
	2. Experimental	pharmaceutics by	Eugene, Parott.
	3. Tutorial pharr	nacy by Cooper an	d Gunn.
	4. Stocklosam Philadelphia.	J. Pharmaceutica	l calculations, Lea & Febiger,
		A, Lachman C., Pha B, Marcel Dekkar I	armaceutical Dosage forms, Tablets, nc.
Other			
References			





School:		SOP		
Programme:		B. Pharm		
Branch:		Semester: 4		
1	Course Code	BP408 P		
2	Course Title	Pharmacology I Practical		
3	Credits	2		
4	Contact Hours (L-T-P)	0-0-4		
	Course Type	Compulsory		
5	Course Objective	Objectives: 1. Upon the completion of the course student shall be able to 2. Understand the pharmacological actions of different categories of		
		drugs. 3. Observe the effect of drugs on animals by simulated experiments 4. Appreciate correlation of pharmacology with other bio medical sciences		
6	Course Outcomes	CO1: Thestudents would be able to explain the pharmacological aspects of drugs. CO2: Thestudents would be able to understand and carry out the animal experiments		
		CO3: The students would be able to appreciate the importance of Pharmacology subject as a basis of therapeutics. CO4: The students would be able to Correlate and apply the knowledge therapeutically. CO5: The students would be able to apply experimental concepts. CO6: The students will be able to explore CNS action on experimental models.		





7	Course Description	Introduction to experimental pharmacology.	
	1	Commonly used instruments in experimental pharmacology. Study of	
		common laboratory animals.	
		Maintenance of laboratory animals as per CPCSEA guidelines.	
		Common laboratory techniques. Blood withdrawal, serum and plasma separation, anesthetics and euthanasia used for animal studies.	
		Study of different routes of drugs administration in mice/rats.	
		Study of effect of hepatic microsomal enzyme inducers on the phenobarbitone sleeping time in mice.	
		Effect of drugs on ciliary motility of frog oesophagus	
		Effect of drugs on rabbit eye.	
		Effects of skeletal muscle relaxants using rota-rod apparatus.	
		Effect of drugs on locomotor activity using actophotometer.	
		Anticonvulsant effect of drugs by MES and PTZ method.	
		Study of stereotype and anti-catatonic activity	
8	Outline syllabus		
	1	Basic Pharmacology Experiment	
		a). Introduction to experimental pharmacology.	
		b). Commonly used instruments in experimental pharmacology.	
		c). Study of common laboratory animals.d). Maintenance of laboratory animals as per CPCSEA guidelines.	
		a). Manifemence of taboratory animals as per or Colly guidelines.	





				www.snarda.a	Can
2		To Study common lab techniques and study the effects of Drugs			
		a. Common	lasma separation,	techniques.	Blo withdrawal,
		inducers on the pher in mice. d. Effect of dru oesophagus e. Effect of dru f. Effects of sk rod apparatus. g. Effect of dru actophotometer.	ce/rats. ect of hepatic micr	ng time lity of frog xants using ro	
	lode of kamination	Practical/Viva			
W	/eightage istribution	Continuous Mode Assessment	Sessional Exam	ESE	
		05	10	35	
Te	ext book/s*	M., Flower Pharmacology Elsevier	P., Dale M. M., IR. J., Rang and y., Churchil Livi	Dale's ngstone	A. J., Basic
		2	harmacology, Tata		*
		3. Goodman as Therapeutics	nd Gilman's, T	he Pharmace	ological Basis of
		4. Marry Anne K L.C., Joseph G. B. Therapeutics, The Williams & Wilkins	Clinical use of Dr	Bradley R.V	
		5. Mycek M.J, Lippincott's Illustra	Gelnet S.B and ted Reviews- Pha	-	1.





	www.sharda.ac.in
Other	
References	





School:		SOP		
Programme:		B. Pharm		
Branch		Semester: 4		
1 Course Code		BP409 P		
2	Course Title	Pharmacognosy and Phytochemistry Practical		
3	Credits	2		
4	Contact Hours (L-T-P)	0-0-4		
	Course Type	Compulsory		
5	Course Objective	Upon completion of the course the student shall be able to understand different methods for analysis of crude drugs by various evaluation parameters i.e. Physical, chemical and organoleptic and anatomical parameters.		
6	Course Outcomes	CO1: Students will discover practical laboratory skills and get hands- on experience of modern scientific instrumentation in relation to the Pharmacognosy.		
		CO2: Students will receive knowledge and understanding of the fundamental principles of pharmacognosy and their applications to pharmaceuticals.		
		CO3: Students will be able to use and apply their skills to a range of techniques used in pharmacognosy.		
		CO4: Students will analyze various crude drugs by various methods.		
		CO5: Students would be able to evaluate various crude drugs for their quality.		
		CO6: Students would be able to study about chemical and quantitative analysis.		
7	Course Description	Analysis of crude drugs by chemical tests: (i)Tragaccanth (ii) Acacia (iii)Agar (iv) Gelatin (v) starch (vi) Honey (vii) Castor oil Determination of stomatal number and index		
		Determination of vein islet number, vein islet termination and palisade ratio. Determination of size of starch grains, calcium oxalate crystals by eye piece micrometer		
		Determination of Fiber length and width Determination of number of starch grains by Lycopodium spore method		
		Determination of Ash value Determination of Extractive values of crude drugs Determination of moisture content of crude drugs		
0	0 41' 11 1	Determination of swelling index and foaming index.		
8	Outline syllabus	·		





2	 Chemica Macrosc Use of m II Determination Ash value Extractive Moisture 	on of physical evalues ve values	pical analysis acida, eye piece micrometer etc.			
3	StomataVein isleFiber ler	of crude drugs by a number and Stomat et, vein termination a gth and width tarch grains and calc	nd Palisade ratio			
4	• Analysis	Chemical and Quantitative analysis • Analysis of crude drugs by chemical tests • Quantitative analysis by lycopodium spore method				
Mode examin	of Practical/Viva					
Weigh Distrib	tage Continuous Mode Assessment	Sessional Exam				
Text book/s* 1. W.C.Evans, Trease and Evans Pharmacognosy, 16th edition, W.B. & Co., London, 2009. 2. Tyler, V.E., Brady, L.R. and Robbers, J.E., Pharmacognosy, 9th and Febiger, Philadelphia, 1988. 3. Text Book of Pharmacognosy by T.E. Wallis 4. Mohammad Ali. Pharmacognosy and Phytochemistry, CBS Pu Distribution, New Delhi. 5. Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhl 37th Edition, Nirali Prakashan, New Delhi. 6. Herbal drug industry by R.D. Choudhary (1996), Ist Edn, Eastern New Delhi			macognosy, 16th edition, W.B. Sounders ers, J.E., Pharmacognosy, 9th Edn., Lea E. Wallis and Phytochemistry, CBS Publishers & C.K. Kokate, Purohit, Gokhlae (2007), elhi.			
Other Refere	ences					





S.	haal	www.sharda.ac.in		
School:		SOP		
	ogramme:	B.Pharm		
\vdash	ranch:	Semester: V		
1	Course	BP501T		
	Code			
2	Course	Medicinal Chemistry-II - Theory		
	Title			
3	Credits	4		
4	Contact	3-1-0		
	Hours			
	(L-T-P)			
	Course	Compulsory		
	Type			
5	Course	Upon completion of the course the student shall be able to		
	Objective	1. understand the chemistry of drugs with respect to their pharmacological		
		activity		
		2. understand the drug metabolic pathways, adverse effect and therapeutic		
		value of drugs		
		3. know the Structural Activity Relationship (SAR) of different class of drugs		
		write the chemical synthesis of some drugs		
6	Course	CO1: The students will be able to understand different categories of drugs		
	Outcomes	with respect to their pharmacological activities.		
		CO2: The students will apply and explain the structure activity relationship,		
		drug metabolic pathways, adverse effects and their therapeutic activity of		
		different categories of drugs.		
		CO3: The student's will be able to plan the knowledge to construct the		
		chemical synthesis of some drugs.		
		CO4: The students will analyse chemical reactions, stabilities of compounds		
		and properties of the compounds prepared by them in the laboratory.		
		CO5: The students will be able to modify and design new chemical		
		compounds with therapeutic activity.		
		CO6: Students will be able to anlyze chemical compounds with therapeutic		
		activity in different diseases.		
7	Course	This subject is designed to impart fundamental knowledge on the structure,		
	Description	chemistry and therapeutic value of drugs. The subject emphasizes on structure		
		activity relationships of drugs, importance of physicochemical properties and		
		metabolism of drugs. The syllabus also emphasizes on chemical synthesis of		
		important drugs under each class.		
8	Outline syllabus			





Study of the development of the following classes of drugs, Classification, mechanism of action, uses of drugs mentioned in the course, Structure activity relationship of selective class of drugs as specified in the course and synthesis of drugs superscripted (*)

Unit I

- A. **Antihistaminic agents:** Histamine, receptors and their distribution in the human body.
- B. **H1–antagonists:** Diphenhydramine hydrochloride*, Dimenhydrinate, Doxylamines cuccinate, Clemastine fumarate, Diphenylphyraline hydrochloride, Tripelenamine hydrochloride, Chlorcyclizine hydrochloride, Meclizine hydrochloride, Buclizine hydrochloride, Chlorpheniramine maleate, Triprolidine hydrochloride*, Phenidamine tartarate, Promethazine hydrochloride*, Trimeprazine tartrate, Cyproheptadine hydrochloride, Azatidine maleate, Astemizole, Loratadine, Cetirizine, Levocetrazine Cromolyn sodium
 - a. **H₂-antagonists:** Cimetidine*, Famotidine, Ranitidin.
 - b. **Gastric Proton pump inhibitors**: Omeprazole, Lansoprazole, Rabeprazole, Pantoprazole
- C. Anti-neoplastic agents:
- a). Alkylating agents: Meclorethamine*, Cyclophosphamide, Melphalan, Chlorambucil, Busulfan, Thiotepa
- **b).** Antimetabolites: Mercaptopurine*, Thioguanine, Fluorouracil, Floxuridine, Cytarabine, Methotrexate*, Azathioprine
- c). Antibiotics: Dactinomycin, Daunorubicin, Doxorubicin, Bleomycin

Plant products: Etoposide, Vinblastin sulphate, Vincristin sulphate Miscellaneous: Cisplatin, Mitotane.





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2	Unit II
	 A. Anti-anginal: B. Vasodilators: Amyl nitrite, Nitroglycerin*, Pentaerythritol tetranitrate, Isosorbide dinitrite*, Dipyridamole. C. Calciumchannel blockers: Verapamil, Bepridil hydrochloride, Diltiazem hydrochloride, Nifedipine, Amlodipine, Felodipine, Nicardipine, Nimodipine. Diuretics:
	Carbonicanhydrase inhibitors: Acetazolamide*, Methazolamide, Dichlorphenamide.
	Thiazides:Chlorthiazide*,Hydrochlorothiazide,
3	 Unit III A. Anti-arrhythmic Drugs: Quinidine sulphate, Procainamide hydrochloride, Disopyramide phosphate*, Phenytoin sodium, Lidocaine hydrochloride, Tocainide hydrochloride, Mexiletine hydrochloride, Lorcainide hydrochloride, Amiodarone, Sotalol.
	B. Anti-hyperlipidemicagents: Clofibrate, Lovastatin, Cholesteramine and Cholestipol
	C. Coagulant&Anticoagulants:Menadione,Acetomenadione,Warfarin*, Anisindione, clopidogrel
	Drugs used in Congestive Heart Failure: Digoxin, Digitoxin, Nesiritide, Bosentan, Tezosentan.





	4	UNIT- IV Hours				
		A. Drugs acting on Endocrine system				
		Nomenclature, Stereochemistry and metabolism of steroids				
		 B. Sex hormones: Testosterone, Nandralone, Progestrones, Oestriol, Oestradiol, Oestrione, Diethyl stilbestrol. C. Drugs for erectile dysfunction: Sildenafil, Tadalafil. Oral contraceptives: Mifepristone, Norgestril, Levonorgestrol Corticosteroids: Cortisone, Hydrocortisone Prednisolone, Betamethasone, Dexamethasone Thyroid and antithyroid drugs: L-Thyroxine, L-Thyronine, Propylthiouracil, Methimazole. 				
	5	UNIT - V Hours				
		A. Antidiabetic agents:				
		Insulin and its preparations				
		Sulfonyl ureas Tolbutamide*, Chlorpropamide, Glipizide, Glimepiride. Biguanides:				
		Metformin.				
		ThiazolidinedionPioglitazone, Rosiglitazone. Meglitinides: Repaglinide,				
	Nateglinide.					
		Glucosidase inhibitors: Acrabose, Voglibose.				
		B. Local Anesthetics: SAR of Local anesthetics				
	C. Benzoic Acid derivatives ; Cocaine, Hexylcaine, Meprylcaine, Cyclomethycaine, Piperocaine.					
	Amino Benzoic acid derivatives: Benzocaine*, Butamben, Procaine*, Butacaine, Propoxycaine, Tetracaine, Benoxinate.					
		Lidocaine/Anilide derivatives : Lignocaine, Mepivacaine, Prilocaine Etidocaine.				
		Miscellaneous: Phenacaine, Diperodon, Dibucaine.*				
	Mode of examinat ion	Theory				
	Weighta	Continuous Mode Sessional Exam ESE				
	ge	Assessment				





Distribut	10 Marks	15	75
ion			
Text	1. Wilson and	Giswold's Organi	c medicinal and Pharmaceutical Chemistry.
book/s*	2. Foye's Prin	ciples of Medicina	l Chemistry.
	3. Burger's M	edicinal Chemistry	y, Vol I to IV.
	4. Introduction to principles of drug design- Smith and Williams.5. Remington's Pharmaceutical Sciences.		
	6. Martindale's extra pharmacopoeia.7. Organic Chemistry by I.L. Finar, Vol. II.		
	8. The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1to 5. 9. Indian Pharmacopoeia.		
	10. Text book of practical organic chemistry- A.I.Vogel.		
Other			
Referenc			
es			





Scho	ool:	SOP		
Programme:		B. Pharm		
Branch:		Semester: V		
1	Course Code	BP502 T		
2	Course Title	Industrial Pharmacy I - Theory		
3	Credits	4		
4	Contact	3-1-0		
	Hours			
	(L-T-P)	C1		
5	Course Type	Compulsory		
3	Course Objective	Upon completion of the course the student shall be able to 1. Know about the various pharmaceutical dosage forms and their manufacturing techniques (large scale equipment's etc)		
		2. Understand the various considerations in development of pharmaceutical dosage forms		
		3. Develop solid, liquid dosage forms and evaluate them for their quality Know about containers, closures and packaging material used for different type of dosage forms.		
6	Course Outcomes	CO1: Students would be able to understand the concept of preformulation studies for the development of safe and effective dosage form.		
		CO2: Students would be able to apply the knowledge various types of dosage form. (tablet, capsule, parenteral, liquid orals, pellets cosmetic preparations etc)		
		CO3: Students would be able to understand the formulation component and manufacturing procedures for different dosage form on Laboratory scale.		
		CO4: Students would be able to formulate ophthalmic Preparations.		
		CO5: Students would be able to formulate and evaluate the formulations for their quality.		
		CO6: Students shall be able to interpret regarding various packaging material for pharmaceutical products and evaluate them for quality.		
7	Course Description	Course enables the student to understand and appreciate the influence of pharmaceutical additives and various pharmaceutical dosage forms on the performance of the drug product.		
8	Outline syllabu	IS .		





1	Unit I
	Preformulation studies-I Introduction to Preformulation, Goals and Objective, study of physicochemical characteristics of drug substances. A. Physical properties: Physical form (crystal & amorphous), particle size, shape, flow properties, solubility profile (pKa, pH, partition coefficient), polymorphism B. Chemical Properties: Hydrolysis, oxidation, reduction, racemisation, polymerization BCS classification of drugs & its significant Application of preformulation considerations in the development of solid, liquid oral and parenteral dosage forms and its impact on stability of dosage forms.
2	Unit II
	A. Tablets: a. Introduction, ideal characteristics of tablets, classification of tablets. Excipients, Formulation of tablets, granulation methods, compression and processing problems. Equipments and tablet tooling. B. Tablet coating: Types of coating, coating materials, formulation of coating composition, methods of coating, equipment employed and defects in coating.
	C. Quality control tests: In process and finished product tests Liquid orals: Formulation and manufacturing consideration of syrups and elixirs suspensions and emulsions; Filling and packaging; evaluation of liquid orals official in pharmacopoeia
3	Unit III Capsules: A. Hard gelatin capsules: Introduction, Production of hard gelatin capsule shells. size of capsules, Filling, finishing and special techniques of formulation of hard gelatin capsules, manufacturing defects. In process and final product quality control tests for capsules. B. Soft gelatin capsules: Nature of shell and capsule content, size of capsules, importance of base adsorption and minim/gram factors, production, in process and final product quality control tests. Packing, storage and stability testing of soft gelatin capsules and their applications. C. Pellets: Introduction, formulation requirements, pelletization process, equipments for manufacture of pellets





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5		essential re B. Production processing parenterals C. Containers vials and in Ophthalmic formulation preparation; Unit V A. Cosmetics preparatio tooth paste B. Pharmacet types of a Evaluation C. Packaging pharmacet and officie	quirements, very procedure, procedure, procedure, procedure, procedure, procedure, procedure and lyophilizer and closures and closures of eye drops, labeling, contact and contact are solved and contact are solved are sol	ages and limitations. Preformulation factors and chicles, additives, importance of isotonicity oduction facilities and controls, aseptic of injections, sterile powders, large volume of products. selection, filling and sealing of ampoules, Quality control tests of parenteral products. Introduction, formulation considerations; eye ointments and eye lotions; methods of tiners; evaluation of ophthalmic preparations and preparation of the following cosmetic shampoos, cold cream and vanishing cream, and sunscreens s: Definition, propellants, containers, valves, as; formulation and manufacture of aerosols; Quality control and stability studies. Science: Materials used for packaging of a factors influencing choice of containers, legal as for containers, stability aspects of packaging
Mode examin	of nation	Theory	Гheory	
Weigh		Continuous	Sessional	ESE
Distrib	_	Mode	Exam	
Distric	ounon		Lam	
		Assessment		
		10 Marks	15	75





	Text book/s*	Pharmaceutical dosage forms - Tablets, volume 1 -3 by H.A. Liberman, Leon Lachman &J. B.Schwartz
Pharmaceutical dosage form - Parenteral n Liberman & Lachman		Pharmaceutical dosage form - Parenteral medication vol- 1&2 by Liberman & Lachman
		Pharmaceutical dosage form disperse system VOL-1 by Liberman & Lachman
		Modern Pharmaceutics by Gilbert S. Banker & C.T. Rhodes, 3rd Edition
Remington: The Science and Practice of Pharmacy, 20th edition Pharmaceutical Science (RPS)		, ,
	Theory and Practice of Industrial Pharmacy by Liberman & Lachman	
		Pharmaceutics- The science of dosage form design by M.E.Aulton, Churchill livingstone, Latest edition
		Introduction to Pharmaceutical Dosage Forms by H. C.Ansel, Lea & Febiger, Philadelphia, 5 th edition, 2005
		Drug stability - Principles and practice by Cartensen & C.J. Rhodes, 3rd Edition, Marcel Dekker Series, Vol 107.





School:		SOP				
Programme:		B. Pharm				
Bra	nch:	Semester: V				
1	Course Code	BP503 T				
2	Course Title	Pharmacology II-Theory				
3	Credits	4				
4	Contact Hours (L-T-P)	3-1-0				
	Course Type	Compulsory				
5	Course Objective	Upon completion of this course the student should be able to 1. Understand the mechanism of drug action and its relevance in the treatment of different diseases 2. Demonstrate isolation of different organs/tissues from the laboratory animals by simulated experiments 3. Demonstrate the various receptor actions using isolated tissue preparation 4. Appreciate correlation of pharmacology with related medical sciences				
6	Course Outcomes	CO1: Students would be able to define and describe various categories of drugs to be used in the treatment of cardiovascular, haematological, endocrine and inflammatory disorders. CO2: Students would be able to understand and explain the mechanisms, pharmacokinetic profile, adverse effects and uses of various drugs. CO3: Students would be able to demonstrate the use of various categories of drugs and their bioassays. CO4: Students would be able to analyze and explain the pathology of the cardiovascular, blood related and endocrine disorders. CO5: Students would be able to evaluate and discriminate amongst the normal and abnormal physiological processes, and various drugs that can e employed for different treatment protocols. CO6: Students would be able to plan about mechanism and therapeutic actions of drugs related to CVS, hormones.				
7	Course Description	This subject is intended to impart the fundamental knowledge on various aspects (classification, mechanism of action, therapeutic effects, clinical uses, side effects and contraindications) of drugs acting on different systems of body and in addition, emphasis on the basic concepts of bioassay.				
8	Outline syllabı	ıs				





1	Unit-1 A. Pharmacology of drugs acting on cardio vascular system a. Introduction to hemodynamic and electrophysiology of heart. b. Drugs used in congestive heart failure c. Anti-hypertensive drugs. d. Anti-anginal drugs. e. Anti-arrhythmic drugs. f. Anti-hyperlipidemic drugs.
2	Unit-2 A. Pharmacology of drugs acting on cardio vascular system a. Drug used in the therapy of shock. b. Hematinics, coagulants and anticoagulants. c. Fibrinolytics and anti-platelet drugs d. Plasma volume expanders B. Pharmacology of drugs acting on urinary system a. Diuretics b. Anti-diuretics.
3	Unit-3 A. Autocoids and related drugs a. Introduction to autacoids and classification b. Histamine, 5-HT and their antagonists. c. Prostaglandins, Thromboxanes and Leukotrienes. d. Angiotensin, Bradykinin and Substance P. e. Non-steroidal anti-inflammatory agents f. Anti-gout drugs g. Antirheumatic drugs.
4	Unit-4 A. Pharmacology of drugs acting on endocrine system B. Basic concepts in endocrine pharmacology. Anterior Pituitary hormones- analogues and their inhibitors. C.Thyroid hormones- analogues and their inhibitors. Hormones regulating plasma calcium level- Parathormone, Calcitonin and Vitamin-D. d. Insulin, Oral Hypoglycemic agents and glucagon. ACTH and corticosteroids.





				TT TI T	
	5	Unit-5			
		A. Pharmacology of drugs acting on endocrine system			
		steroids.			
	d oral contraceptives.				
		c. Drugs actin	g on the uterus	S.	
		B. Bioassay			
		a. Principles a	nd application	s of bioassay.	
		b.Types of bio			
		c. Bioassa	y of ins	ulin, oxytocin, vasopressin, ACTH,d-	
		tubocurarine,	ligitalis, histan	nine	
		and 5-HT			
	Mode of	Theory			
	examination	·			
	Weightage	Continuous	Sessional	ESE	
	Distribution	Mode	Exam		
		Assessment			
		10 Marks	15	75	
	Text book/s*	1. Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras			
	medical publisher				
		2. Modern Pharmacology with clinical Applications, by Charles			
		R.Craig& Robert.			
		3. Ghosh MN. Fundamentals of Experimental			
		Pharmacology. Hilton & Company, Kolkata.			
		4. Kulkarni SK. Handbook of experimental pharmacology. Vallable			
		Prakashan.			





Sc	hool:	SOP		
Programme:		B. Pharm		
	anch:	Semester: V		
1	Course	BP504T		
-	Code			
2	Course	Pharmacognosy – II Theory		
-	Title	That macognosy if Theory		
3	Credits	4		
4	Contact	3-1-0		
	Hours			
	(L-T-P)			
	Course	Compulsory		
	Type			
5	Course	Upon completion of the course, the student shall be able		
	Objective	1. to know the modern extraction techniques, characterization and		
		identification of the herbal drugs and phytoconstituents		
		2. to understand the preparation and development of herbal formulation.		
		3. to understand the herbal drug interactions		
		to carryout isolation and identification of phytoconstituents		
6	Course	CO1: Students would be able to define and describe various metabolic		
	Outcomes	pathways, various secondary metabolites like alkaloids glycosides by		
		spectroscopic techniques and chromatography and various extraction		
		methods		
		CO2: Students would be able to explain applications of phytoconstituents		
		and their industrial production, isolation process and extraction methods		
		was mon manufacture production product and constraint months and		
		CO3: Students would be able to apply and demonstrate various identification		
		process and latest technique of phytoconstituents		
		CO4: Students would be able to separate and analyse various		
		phytoconstituents		
		phytoconstituents		
		CO5: Students would be able to evaluate various phytoconstituent		
		Cos. Students would be able to evaluate various phytoconstituent		
		CO6: Students would be able to understand about metabolic pathways.		
		p		
7	Course	The main purpose of subject is to impart the students the knowledge		
	Description	of how the secondary metabolites are produced in the crude drugs,		
	1	how to isolate and identify and produce them industrially. Also this		
		subject involves the study of producing the plants and		
		phytochemicals through plant tissue culture, drug interactions and		
		basic principles of traditional system of medicine		
8	Outline Sylla	bus		





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1	 UNIT-I Metabolic pathways in higher plants and their determination a. Brief study of basic metabolic pathways and formation of different secondary metabolites through these pathways- Shikimic acid pathway, b. Acetate pathways and Amino acid pathway. c. Study of utilization of radioactive isotopes in the investigation of Biogenetic studies.
2	UNIT-II
	14 Hours General introduction, composition, chemistry & chemical classes, biosources, therapeutic uses and commercial applications of following a. secondary metabolites:
	Alkaloids: Vinca, Rauwolfia, Belladonna, Opium, Phenylpropanoids and Flavonoids: Lignans, Tea, Ruta Steroids, Cardiac Glycosides & Triterpenoids: Liquorice, Dioscorea, Digitalis Volatile oils: Mentha, Clove, Cinnamon, Fennel, Coriander, Tannins: Catechu, Pterocarpus b. Resins: Benzoin, Guggul, Ginger, Asafoetida, Myrrh, Colophony c. Glycosides: Senna, Aloes, Bitter Almond Iridoids, Other terpenoids & Naphthaquinones: Gentian, Artemisia, taxus, carotenoids
3	UNIT-III
	06 Hours Isolation, Identification and Analysis of Phytoconstituents ATerpenoids: Menthol, Citral, Artemisin B. Glycosides: Glycyrhetinic acid & Rutin C. Alkaloids: Atropine, Quinine, Reserpine, Caffeine Resins: Podophyllotoxin, Curcumin
4	UNIT-IV
	 10 Hours A. Industrial production, estimation and utilization of the following phytoconstituents: B. Forskolin, Sennoside, Artemisinin, Diosgenin, Digoxin, Atropine, Podophyllotoxin, C. Caffeine, Taxol, Vincristine and Vinblastine





			WWW. State Control of the Control of
5	Hours a. Basics of Phytochemistry b. Modern methods of extraction, application of latest techniques like Spectroscopy, c. chromatography and electrophoresis in the isolation, purification and identification of crude drugs.		
Mode of examinati	Theory		
Weightag e	Continuous Mode Assessment	Sessional Exam	ESE
Distributi on	10 Marks	15	75
Text book/s*	W.C. Evans, Trease and Evans Pharmacognosy, 16th edition, W.B. Sounders & Co., London, 2009. Mohammad Ali. Pharmacognosy and Phytochemistry, CBS Publishers & Distribution, New Delhi. Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhlae (2007), 37th Edition, Nirali Prakashan, New Delhi. Herbal drug industry by R.D. Choudhary (1996), Ist Edn, Eastern Publisher, New Delhi. Essentials of Pharmacognosy, Dr.SH.Ansari, IInd edition, Birla publications, New Delhi, 2007 Herbal Cosmetics by H.Pande, Asia Pacific Business press, Inc, New Delhi. A.N. Kalia, Textbook of Industrial Pharmacognosy, CBS Publishers, New Delhi, 2005. R Endress, Plant cell Biotechnology, Springer-Verlag, Berlin, 1994. Pharmacognosy & Pharmacobiotechnology. James Bobbers, Marilyn KS, VE Tylor.		
Other Reference s			





School: SOP		SOP		
Programme:		B.Pharm		
	ranch:	Semester: V		
1	Course	BP505T		
	Code			
2	Course	Pharmaceutical Jurisprudence - Theory		
	Title			
3	Credits	4		
4	Contact	3-1-0		
	Hours			
	(L-T-P)			
	Course	Compulsory		
	Type			
5	Course	Upon completion of the course, the student shall be able to understand:		
	Objective	1. The Pharmaceutical legislations and their implications in the development		
		and marketing of pharmaceuticals.		
		Various Indian pharmaceutical Acts and Laws		
		3. The regulatory authorities and agencies governing the manufacture and sale		
	C	of PharmaceuticalsCode of ethics		
6	Course	CO1: Students would be able to identify and understand theknowledge of Legal		
	Outcomes	definitions of schedules to the Act and Rules, license for manufacturing and		
		sale of drugs.		
		CO2: Students would be able to explain various schedules labelling and		
		CO2: Students would be able to explain various schedules labelling and packaging of drugs and various acts and rules.		
		packaging of drugs and various acts and rules.		
		CO3: Students would be able to differentiate various acts and rules Schedules		
		sale of drugs and various Acts and Rules and apply Rules.		
		sale of drags and various from and reales and appry reales.		
		CO4: Students would be able to interpret various Acts and Rules and how to		
		apply various acts and Rules.		
		CO5: Students would be able to summarize the acts and code and conduct and		
		also would explain Intellectual Proprietary Rights		
		CO6: Students would be able to plan about various schedules of D&C Act.		
7	Course	This course is designed to impose having beginning to the state of the		
	Description	This course is designed to impart basic knowledge on important		
		legislations related to the profession of pharmacy in India		
0	Outline syllabus			
8	8			





Act and Rules whibited from import, es.
nd sale of certain
t XII B, Sch F & DMR
license. Offences and
rements and specimen colors. Offences and
nical Advisory Board, tee, Government drug rugs Inspectors
Council of India; its Joint state pharmacy cists, Offences and
bjectives, Definitions, Export of alcoholic Patent & Proprietary
tion and Functions of Fund for Controlling poppy cultivation and f opium, Offences and





UNIT-IV					
rules: Objectives, 1	gs and Magic Remedies Act and its ition of certain advertisements, Classes of d Penalties				
B. Prevention of Cruelty to animals Act-1960: Objective Institutional Animal Ethics Committee, CPCSEA guidelines for Stocking of Animals, Performance of Experiments, Transfer and animals for experiment, Records, Power to suspend or revolutional Conference and Penalties					
C. National Pharmaceutical Pricing Authority: Drugs Price Co (DPCO)- 2013. Objectives, Definitions, Sale prices of bulk drugs, Reformulations, Retail price and ceiling price of scheduled formulation. List of Essential Medicines (NLEM)					
UNIT-V					
A. Pharmaceutical Legislations – A brief review, Introduction, Study of drug enquiry committee, Health survey and development committee, Hathi committee and Mudaliar committee					
B. Code of Pharmaceutical ethics D efinition, Pharmacist in relation to his job, trade, medical profession and his profession, Pharmacist's oath					
C.Medical Termina	tion of Pregnancy	Act			
Right to I	Information Act				
Introduction to Intellectual Property Rights (IPR)					
Theory					
Continuous Mode	Sessional Exam	ESE			
Assessment	1.5	95			
10 Marks	15	75			
	A. Study of Salient rules: Objectives, Exempted advertise B. Prevention of Institutional Animal Stocking of Animal animals for experion Offences and Penalt C. National Pharm (DPCO)- 2013. Object formulations, Retail List of Essential Medical Enquiry committee, and Mudaliar committee, and Mudal	A. Study of Salient Features of Drug rules: Objectives, Definitions, Prohib Exempted advertisements, Offences and B. Prevention of Cruelty to ani Institutional Animal Ethics Committe Stocking of Animals, Performance of animals for experiment, Records, P Offences and Penalties C. National Pharmaceutical Pricin (DPCO)- 2013. Objectives, Definitions formulations, Retail price and ceiling List of Essential Medicines (NLEM) UNIT-V A. Pharmaceutical Legislations – A enquiry committee, Health survey and and Mudaliar committee B. Code of Pharmaceutical ethics D trade, medical profession and his profes C.Medical Termination of Pregnancy Right to Information Act Introduction to Intellectual Theory Continuous Mode Assessment Sessional Exam Assessment			





Text	1. Text book of Forensic Pharmacy by B.M. Mithal
book/s*	2. Hand book of drug law-by M.L. Mehra
	3. A text book of Forensic Pharmacy by N.K. Jain
	4. Drugs and Cosmetics Act/Rules by Govt. of India publications.
	5. Medicinal and Toilet preparations act 1955 by Govt. of India publications.
	6. Narcotic drugs and psychotropic substances act by Govt. of India publications
	7. Drugs and Magic Remedies act by Govt. of India publication
	8.Bare Acts of the said laws published by Government. Reference books (Theory)





Sc	hool:	SOP
Pr	ogramme:	B.Pharm
_	anch:	Semester: V
1	Course	BP506P
	Code	
2	Course	Industrial Pharmacy- I Practical
	Title	
3	Credits	2
4	Contact	0-0-4
	Hours	
	(L-T-P)	
	Course	Compulsory
	Type	
5	Course	Upon completion of the course the student shall be able to
	Objective	
		1. Know about the various pharmaceutical dosage forms and their
		manufacturing techniques(large scale equipment's etc)
		2. Understand the various considerations in development of pharmaceutical
		dosage forms
		3. Develop solid, liquid dosage forms and evaluate them for their quality
		Know about containers, closures and packaging material used for different type
		of dosage forms.
6	Course	CO1: Students would be able to understand various types of dosage
	Outcomes	form. (tablet, capsule, Parenterals, creams etc)
		CO2: Students would be able to plan about the manufacturing procedures for
		different dosage form on laboratory scale. (tablet, capsule, Parenterals,
		creams etc)
		CO3: Students would be able to evaluate the Aerosols.
		COA. Students would be able to also regions and region westerial for
		CO4: Students would be able to plan various packaging material for
		pharmaceutical products and evaluate them for quality
CO5: Students shall be able to		CO5: Students shall be able to formulate Cosmetics.
Cos. Students shall be able to formulate Cosmetics.		Cos. Students shall be able to formulate Cosmeties.
		CO6: Students shall be able to evaluate the different types of Dosage forms.
		!





	G			www.sharda.ac.in
'/	Course	To emphasize on	definition types	classification, principles/mechanisms,
	Description	applications, examples and differences		
		Formulation and evaluation of the following dosage forms containing drugs		
		mentioned in pharmacopoeia.		
		1. Capsules.		
		2. Microcapsules/microspheres3. Tablets by dry and wet granulation methods		
		4. Film coate	d tablets/ Enteric co	oated tablets, Ear drops
8	Outline Syllal			
	1	To study the variou	is instruments used	in evaluation of
		tablet.		
	2	Evaluation of table	t as per IP.	
			1	
	3	Prepare and evaluat	te oranules of Calci	um lactate 50 Tablets.
	4			f Acetyl salicylic acid by using Tablet
	7			Disintegration Time and hardness of
			ina determine men	Disintegration Time and naturess of
		prepared tablets.	C .: 1: :	
	5	To study the effect		
	6	To prepare efferves	scent granules by he	ot and wet method.
	7	To prepare microcapsules by using phase separation & coacervation technique		
		brought about by polymer polymer interaction.		
	8	To prepare and submit cold cream		
	9	Preparation of injection		
	10	To prepare and submit vanishing cream		
		i i	\mathcal{E}	
	Mode of	Practical/Viva		
	examination	Tractical/Viva		
	Weightage	Continuous	Sessional Exam	ESE
			Sessional Exaill	ESE
	Distribution	Mode		
		Assessment		
		05Marks	10	35
	Text	Pharmaceutical do	osage forms - Ta	hlets volume 1 -3 by H A
	book/s*	Pharmaceutical dosage forms - Tablets, volume 1 -3 by H.A.		
		Liberman, Leon Lachman & J.B.Schwartz.		
		Pharmaceutical dosage form - Parenteral medication vol- 1&2 by		
		Liberman & Lachman		
		Pharmaceutical dosage form disperse system VOL-1 by Liberman & Lachman		
		rnamaceuncai dos	sage form disperse s	system vol-1 by Liberman & Lachman





			www.silaida.ac.iii
Other			
References			





Sc	hool:	SOP
Pr	ogramme:	B. Pharm
	anch:	Semester: V
1	Course	BP507P
	Code	
2	Course	Pharmacology- II Practical
	Title	
3	Credits	2
4	Contact	0-0-4
	Hours	
	(L-T-P)	
	Course	Compulsory
	Type	
5	Course	Upon completion of this course the student should be able to
	Objective	1. Understand the mechanism of drug action and its relevance in the
		treatment of different diseases
		2. Demonstrate isolation of different organs/tissues from the laboratory
		animals bysimulated experiments
		3. Demonstrate the various receptor actions using isolated tissue preparation
		4. Appreciate correlation of pharmacology with related medical
		sciences
6	Course Outcomes	CO1: Students would be able to define and describe various instruments and methods used in the evaluation of <i>in vitro</i> and <i>in vivo</i> evaluation of various drugs. CO2: Students would be able to understand and explain the working
		principles of the instruments used and actions of various drugs on biological systems. CO3: Students would be able to demonstrate the effects of various categories
		of drugs and bioassays of physiological substances.
		CO4: Students would be able to analyze and explain the outcomes of experiments through simulation studies.
		CO5: Students would be able to evaluate and discriminate amongst the normal and abnormal physiological processes, and various drugs that can be employed for different treatment protocols.
		CO6: Students would be able to plan about the effects of various bioassays of physiological substance and its application.





7				www.sharda.ac.in		
/	Course	1. Introduction to <i>in-vitro</i> pharmacology and physiological salt solutions.				
	Description	2. Effect of drugs on isolated frog heart.				
		3. Effect of drugs on blood pressure and heart rate of dog.				
		e .				
		4. Study of diuretic activity of drugs using rats/mice.5. DRC of acetylcholine using frog rectus abdominis muscle.				
		_				
				oine on DRC of acetylcholine using frog		
			muscle and rat ileur	<u> </u>		
		2		a pig ileum by matching method.		
		2 2		rine horn by interpolation method.		
		Bioassay of sero	otonin using rat fur	ndus strip by three point bioassay.		
		10. Bioassay of ac	etylcholine using r	rat ileum/colon by four point bioassay.		
		11. Determination	of PA2 value of p	razosin using rat anococcygeus muscle		
		(by Schilds plot m				
		` •	,	g guinea pig ileum.		
		13. Effect of spasr	nogens and spasmo	olytics using rabbit jejunum.		
		-		ugs using carrageenan induced paw-		
		edema model.	itely desiring of al	ags asing carrageonan maacca paw		
			of drug using cent	ral and peripheral methods		
8	Outline Syalla		of drug using cent	rai and peripheral methods		
8		<u>uus</u>				
	1	Introduction to i	<i>n-vitro</i> pharmaco	logy and physiological salt		
		solutions.	-			
	2	Effect of drugs on	isolated frog heart.			
	3	Effect of drugs on	blood pressure and	l heart rate of dog.		
	4	Study of diuretic activity of drugs using rats/mice.				
	5	,		tus abdominis muscle.		
	6		<u> </u>	ne on DRC of acetylcholine using frog		
			muscle and rat ileur			
	7			big ileum by matching method.		
	8					
	9			horn by interpolation method.		
		•		us strip by three point bioassay.		
	10	Bioassay of acetyl	cnoline using rat il	eum/colon by four point bioassay.		
	Mode of	Practical/Viva				
	examination					
	Weightage	Continuous	Sessional Exam	ESE		
	Distribution	Mode				
		Assessment				
			10	25		
	1	05Marks	10	35		





Text	Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras medical
book/s*	publisher
	Modern Pharmacology with clinical Applications, by Charles R.Craig&
	Robert.
	Ghosh MN. Fundamentals of Experimental Pharmacology.
	Hilton & Company, Kolkata.
	Kulkarni SK. Handbook of experimental pharmacology. Vallabh Prakashan.
Other	
References	





Sc	hool:	SOP	
Pr	ogramme:	B. Pharm	
	anch:	Semester: V	
1	Course	BP508P	
	Code		
2	Course	Pharmacognosy - II Practical	
	Title		
3	Credits	2	
4	Contact	0-0-4	
	Hours		
	(L-T-P)		
	Course	Compulsory	
	Type		
5	Course Objective	1. Explain correct use of various equipments in Pharmacognosy laboratory.	
		2. Handle simple/compound microscope in technically correct way.	
		3. Expain and understand the Morphology, histology and powder characteristics	
		4. Demonstrate skill of plant material sectioning, staining, mounting & focusing.	
	5. Decide on staining reagents required for specific part of plant.		
		6. Demonstrate Isolation and detection methods	
		7. Separate phytoconstituents by TLC	
6	Course Outcomes	CO1: Students would able to identify and describe the morphology and chemical test of crude drugs	
		CO2: Students would be able to explain and compare the microscopy of crude drugs and powder CO3: Students would be able to calculate the Rf value of phytoconstituents CO4: Students would be able to separate and analyze the phytoconstituents CO5: Students would be able to apply knowledge to isolate the compounds. CO6: Students would be able to evaluate the Phytoconstituents.	





7	Course	Morphology histol	ogy and nowder ch	naracteristics & extraction & detection of:	
'	Description	Cinchona, Cinnamon, Senna, Clove, Ephedra, Fennel and Coriander			
	Bescription	Exercise involving isolation & detection of active principles			
			iosgenin from Dios		
			tropine from Bella		
			ennosides from Ser		
		2. Separ	ration of sugars by	Paper chromatography	
		3. TLC	of herbal extract		
		4. Distil	lation of volatile o	ils and detection of phytoconstitutents by	
		TLC			
		5. Analy	vsis of crude drugs	by chemical tests: (i)	
		_		(iii) Colophony (iv)	
			s (v) Myrrh		
			•		
8	Outline Syllal	bus			
	1	To study the morp	phological and mice	oscopy of Cinchona bark.	
	2	To study the morph	nological and micro	scopy of Fennel fruits.	
		_			
	3	To study the mor	phological charact	teristics of Senna leaves, Cinnamon	
		bark and Ephedra	bark and Ephedra stem		
	4	To study the power	der characteristics of	of clove buds and Cinnamon bark	
	5	To study the morphological, and histological characteristics of clove bud			
	6	To study the mo	rphological, micro	scopy and powder characteristics of	
		Ephedra stem			
	7		from tea powder a	and identify by Thin Layer	
		chromatography	1	3 3	
		8 1 7			
	8			per chromatography	
	9	To perform Thin La	ayer chromatograp	hy of the given herbal extract.	
	10	To isolate volat	ile oil by bydro	distillation method using Clavengers	
	10		nation of viscosity		
		liquid using Ostwa	-	01	
	Mode of	Practical/Viva	ara s visconicici		
	examination	Tractical/Viva			
	Weightage	Continuous	Sessional Exam	ESE	
	Distribution	Mode	Sessional Exam	LOL	
	Distribution				
		Assessment	10	25	
		05 Marks	10	35	





Text book/s*	 A.N. Kalia, Textbook of Industrial Pharmacognosy, CBS Publishers, New Delhi, 2005. R Endress, Plant cell Biotechnology, Springer-Verlag, Berlin, 1994. Pharmacognosy & Pharmacobiotechnology. James Bobbers, Marilyn KS, VE Tylor. The formulation and preparation of cosmetic, fragrances and flavours. Remington's Pharmaceutical sciences. Text Book of Biotechnology by Vyas and Dixit. Text Book of Biotechnology by R.C. Dubey.
Other References	





Sc	chool:	SOP
	ogramm	B.Pharm
e:	ogi amm	D.1 Hat III
	ranch:	Semester: VI
1	Course	BP601T
1	Code	DI 0011
2	Course	Medicinal Chemistry III – Theory
	Title	Niculcinal Chemistry III – Theory
3	Credits	4
4	Contact	3-1-0
•	Hours	
	(L-T-P)	
	Course	Compulsory
	Type	
5	Course	Upon completion of the course student shall be able to-
	Objectiv	1. Understand the importance of drug design and different techniques of drug
	e	design.
		2. Understand the chemistry of drugs with respect to their biological
		activity.
		3. Know the metabolism, adverse effects and therapeutic value of drugs.
		Know the importance of SAR of drugs.
6	Course	CO1: Student will be able to describe get fundamental knowledge of the
	Outcom	structure, chemistry and its correlation with the therapeutic value of drugs.
	es	
		CO2: Students will be able to apply conceptual knowledge and background of
		drugs and ensure their rational use.
		CO2. Students will be able to also about the countback and Standard Activity
		CO3: Students will be able to plan about the synthesis and Structure Activity Relationships (SAR) associated with the drugs structure.
		Relationships (SAR) associated with the drugs structure.
		CO4: Students will also conclude about the chemistry, mechanism of action,
		metabolism, adverse effects, and therapeutic uses of important drugs.
		,,
		CO5: Students will be able to the modern techniques of rational drug design like
		quantitative structure activity relationship
		•
		CO6: Students will be able to construct the Drug design
7	Course	This subject is designed to impart fundamental knowledge on the structure,
	Descript	chemistry and therapeutic value of drugs. The subject emphasis on modern
	ion	techniques of rational drug design like quantitative structure activity relationship
		(QSAR), Prodrug concept, combinatorial chemistry and Computer aided drug
		design (CADD). The subject also emphasizes on the chemistry, mechanism of
		action, metabolism, adverse effects, Structure Activity Relationships (SAR),
		therapeutic uses and synthesis of important drugs.





8	Outline syllabus		
	1	UNIT – I	
		A. Antibiotics	
		Historical background, Nomenclature, Stereochemistry, Structure activity	
		relationship, Chemical degradation classification and important products of the following classes.	
		β-Lactam antibiotics: Penicillin, Cepholosporins, β- Lactamase inhibitors, Monobactams	
		Aminoglycosides: Streptomycin, Neomycin, Kanamycin	
		Tetracyclines: Tetracycline,Oxytetracycline, Chlortetracycline	
	2	UNIT – II	
		A. Antibiotics	
		Historical background, Nomenclature, Stereochemistry, Structure activity relationship, Chemical degradation classification and important products of the following classes.	
		B. Macrolide: Erythromycin Clarithromycin, Azithromycin.	
i		C. Miscellaneous: Chloramphenicol*, Clindamycin.	
		Prodrugs: Basic concepts and application of prodrugs design.	
		Antimalarials: Etiology of malaria.	
		Quinolines: SAR, Quinine sulphate, Chloroquine*, Amodiaquine, Primaquine phosphate, Pamaquine*,	
		Quinacrine hydrochloride, Mefloquine.	
		Biguanides and dihydro triazines: Cycloguanil pamoate, Proguanil. Miscellaneous: Pyrimethamine, Artesunete, Artemether, Atovoquone	





3	
	UNIT – III

- A. Anti-tubercular Agents
- **B. Synthetic anti tubercular agents:** Isoniozid*, Ethionamide, Ethambutol, Pyrazinamide, Para amino salicylic acid.*
- C. Anti tubercular antibiotics: Rifampicin, Rifabutin, Cycloserine Streptomycine, Capreomycin sulphate. Urinary tract anti-infective agents

Quinolones: SAR of quinolones, Nalidixic Acid,Norfloxacin, Enoxacin, Ciprofloxacin*, Ofloxacin, Lomefloxacin, Sparfloxacin, Gati

floxacin, Moxifloxacin

Miscellaneous: Furazolidine, Nitrofurantoin*, Methanamine.

Antiviral agents:

Amantadine hydrochloride, Rimantadine hydrochloride, Idoxuridine trifluoride, Acyclovir*, Gancyclovir, Zidovudine, Didanosine, Zalcitabine, Lamivudine, Loviride, Delavirding, Ribavirin, Saquinavir, Indinavir,





4	UNIT - IV		
	8 Hours		
	A. Antifungal agents: B. Antifungal antibiotics: Amphotericin-B, Nystatin, Natamycin,		
	Griseofulvin. C. Synthetic Antifungal agents: Clotrimazole, Econazole, Butoconazole, Oxiconazole Tioconozole, Miconazole*, Ketoconazole, Terconazole, Itraconazole, Fluconazole, Naftifine hydrochloride, Tolnaftate*.		
	Anti-protozoal Agents: Metronidazole*, Tinidazole, Ornidazole, Diloxanide, Iodoquinol, Pentamidine Isethionate, Atovaquone, Eflornithine.		
	Anthelmintics: Diethylcarbamazine citrate*, Thiabendazole, Mebendazole*, Albendazole, Niclosamide, Oxamniquine, Praziquantal, Ivermectin.		
	Sulphonamides and Sulfones		
	Historical development, chemistry, classification and SAR of Sulfonamides: Sulphamethizole, Sulfisoxazole, Sulphamethizine, Sulfacetamide*, Sulphapyridine, Sulfamethoxaole*, Sulphadiazine, Mefenide acetate, Sulfasalazine.		
	Folate reductase inhibitors: Trimethoprim*, Cotrimoxazole.		
	Sulfones: Dapsone*.		
5	UNIT – V		
	A. Introduction to Drug Design		
	Various approaches used in drug design.		
	Physicochemical parameters used in quantitative structure activity relationship (QSAR) such as partition coefficient, Hammet's electronic parameter, Tafts steric parameter and Hansch analysis.		
	Pharmacophore modeling and docking techniques.		
	B. Combinatorial Chemistry: Concept and applications chemistry:		
	C. solid phase and solution phase synthesis.		





Mode of	Theory		
examina			
tion			
Weighta	Continuous Mode	Sessional Exam	ESE
ge	Assessment		
Distribut	10 Marks	15	75
ion			
Text	1. Wilson and Giswo	ld's Organic medicin	al and Pharmaceutical Chemistry.
book/s*	2. Foye's Principles of Medicinal Chemistry.		
	3. Burger's Medicina	l Chemistry, Vol I to	IV.
	Introduction to princip	les of drug design- S	mith and Williams
Other	Text book of practical organic chemistry- A.I.Vogel.		
Referenc			
es			





Sc	chool:	SOP		
Pr	ogramme:	B.Pharm		
Bı	ranch:	Semester: VI		
1	Course	BP602T		
	Code			
2	Course	Pharmacology - III – Theory		
	Title			
3	Credits	4		
4	Contact	3-1-0		
	Hours			
	(L-T-P)			
	Course	Compulsory		
	Type			
5	Course	Upon completion of this course the student should be able to:		
	Objective	1. understand the mechanism of drug action and its		
		relevance in the treatment of different infectious diseases		
		2. comprehend the principles of toxicology and treatment		
		of various poisonings		
		3. appreciate correlation of pharmacology with related medical sciences.		
6	Course	CO1: Students would be able to define and describe various categories of		
	Outcomes	drugs to be used in the treatment of respiratory, gastrointestinal, infectious		
		and malignant disorders.		
		CO2: Students would be able to understand and explain the mechanisms,		
pharmacokinetic profile, adverse effects and uses of various drugs.		pharmacokinetic profile, adverse effects and uses of various drugs.		
		CO3: Students would be able to demonstrate the use of various categories of		
drugs and their bioassays.		drugs and their bioassays.		
		CO4: Students would be able to analyze and explain the pathology of cancer,		
		infectious, respiratory and gastrointestinal diseases.		
		CO5: Students would be able to evaluate and discriminate amongst the normal		
		and abnormal physiological processes, and various drugs that can be employed		
		for different treatment protocols.		
		CO6: The student would be able to conclude the rational use of antibiotics.		
7	Course			
	Descriptio	This subject is intended to impart the fundamental knowledge on various		
	n	aspects (classification, mechanism of action, therapeutic effects, clinical uses,		
		side effects and contraindications) of drugs acting on respiratory and		
		gastrointestinal system, infectious diseases, immuno-pharmacology and in		
		addition, emphasis on the principles of toxicology and chronopharmacology.		
\vdash				
8	8 Outline syllabus			
U	Juline Sylle			





1				
1	UNIT-I			
	A. Pharmacology of drugs acting on Respiratory system			
	a. Anti -asthmatic drugs			
b. Drugs used in the management of COPD				
	c. Expectorants and antitussives			
	d. Nasal decongestants			
	B. Respiratory stimulants Pharmacology of drugs acting on the			
	Gastrointestinal Tract			
	a. Antiulcer agents.			
	b. Drugs for constipation and diarrhoea.			
	c. Appetite stimulants and suppressants.			
	d. Digestants and carminatives.			
	e. Emetics and anti-emetics.			
2	UNIT-II			
	A. Chemotherapy			
	a. General principles of chemotherapy.			
	b. Sulfonamides and cotrimoxazole.			
	c.Antibiotics- Penicillins, cephalosporins,			
	chloramphenicol, macrolides, quinolones and			
	fluoroquinolins, tetracycline and			
	aminoglycosides			
	ammogrycosides			
3	UNIT-III			
	A. Chemotherapy			
	a. Antitubercular agents			
	b. Antileprotic agents			
	c. Antifungal agents			
	d. Antiviral agents			
	e. Anthelmintic agents			
	f. Antimalarial drugs			
	g. Antiamoebic agents			
	1			





4	UNIT-IV		
	A. Chemotherapy		
	1. Urinary tract infections and sexually transmitted diseases.		
	m. Chemotherapy of m	alignancy.	
	B. Immunopharmacol	ogy	
	a. Immunostimulants		
	b. Immunosuppressant		
	Protein drugs, monoclo	onal antibodies, targe	et drugs to antigen, biosimilars
5	UNIT-V		
3	A. Principles of toxico	alogy	
	· -	0.0	subacute and chronictoxicity
	a. Definition and basic knowledge of acute, subacute and chronictoxicity. Definition and basic knowledge of genotoxicity, carcinogenicity, teratogenicity		
	and mutagenicity		
	General principles of treatment of poisoning		
	c. Clinical symptoms and management of barbiturates, morphine,		
	organophosphorus compound and lead, mercury and arsenic poisoning.		
	B. Chronopharmacology		
	a. Definition of rhythm and cycles.b. Biological clock and their significance leading to chronotherapy.		
	b. Biological clock	and their significance	reading to enfonomerapy.
Mode of	Theory		
examinat			
ion			
Weighta	Continuous Mode	Sessional Exam	ESE
ge	Assessment		
Distribut	10 Marks	15	75
10n			





-	4 7 77 7 7 1 1 1 1 1 7 7 7 7 7 7 7 7 7 7		
Text	1. Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and		
book/s*	Dale's Pharmacology, Churchil Livingstone Elsevier		
	2. Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical		
	pharmacology, Tata Mc Graw-Hill		
	β. Goodman and Gilman's, The Pharmacological Basis of Therapeutics		
	4. Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin		
	L.C., Joseph G. B., Wayne A. K., Bradley R.W., Applied		
	Therapeutics, The Clinical use of Drugs. The Point Lippincott		
	Williams & Wilkins		
	5. Mycek M.J, Gelnet S.B and Perper M.M.		
	Lippincott's Illustrated Reviews- Pharmacology		
	6. K.D.Tripathi. Essentials of Medical Pharmacology, ,		
	JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.		
	7. Sharma H. L., Sharma K. K., Principles of Pharmacology,		
	Paras medical publisher Modern Pharmacology with clinical		
	Applications, by Charles R.Craig& Robert,		
	8. Ghosh MN. Fundamentals of Experimental Pharmacology.		
	Hilton & Company, Kolkata,		
	9. Kulkarni SK. Handbook of experimental pharmacology.		
	VallabhPrakashan,		
	10. N.Udupa and P.D. Gupta, Concepts in Chronopharmacology.		
	and the super sure that the superior in the consequent that the superior is the superior in th		
Other			
Referenc			
es			
CS			





Sc	chool:	SOP	
_	ogramme:	B.Pharm	
	ranch:	Semester: VI	
1	Course Code	BP603T	
2	Course Title	Herbal Drug Technology – Theory	
3	Credits	4	
4	Contact	3-1-0	
	Hours		
	(L-T-P)	Communication	
_	Course Type	Compulsory	
5	Course	Upon completion of the course, the student shall be able to	
	Objective	1. Understand raw material as source of herbal drugs from cultivation to	
		herbal drug product. 2. Know the WHO and ICH guidelines for evaluation of herbal drugs.	
		know the herbal cosmetics, natural sweeteners, neutraceuticals.	
6	Course	CO1: Students would be able to define herbal medicine, identify and	
U	Outcomes	authentication of herbal materials, describe neutraceuticals and herbal drug	
	Outcomes	interactions	
		CO2: Students would be able to differentiate Indian system of medicine	
		and would be able to describe Stability testing of herbal drugs and	
		explain patenting	
		CO3: Students would be able apply and various identification process and	
		latest technique of phytoconstituents	
		CO4: Students would be able to demonstrate evaluation of drugs	
		according to W.H.O. guidelines.Patenting and regulatory requirements of natural and analyse various phytoconstituents	
		CO5: Students would be able to evaluate various phytoconstituents Herbal	
		drug Industry Schedule T-Good manufacturing practices of Indian system of medicine	
		CO6: Students would be able to formulate herbs or natural origin drugs as	
		raw materials for preparation of cosmetics, excipients, conventional herbal	
		formulation and novel dosage forms like Phytosomes	
		Termination and nover debuge forms like I hyposomes	
7	Course		
	Description	This subject gives the student the knowledge of basic understanding of herbal	
	_	drug industry, the quality of raw material, guidelines for quality of herbal	
		drugs, herbal cosmetics, natural sweeteners, nutraceutical etc. The subject also	
		emphasizes on Good Manufacturing Practices (GMP), patenting and	
		regulatory issues of herbal drugs	
8	Outline syllabi	us	
	Catillio Byllabl	nu .	





	WWW.3lidi Ma.de.ili		
1	UNIT-I		
	A. Herbs as raw materials		
	Definition of herb, herbal medicine, herbal medicinal product, herbal drug		
	preparation Source of Herbs		
	Selection, identification and authentication of herbal materials Processing of		
	herbal raw material		
	B. Biodynamic Agriculture		
	Good agricultural practices in cultivation of medicinal plants including Organic		
	farming. Pest and Pest management in medicinal plants:		
	Biopesticides/Bioinsecticides.		
	C. Indian Systems of Medicine		
	a) Basic principles involved in Ayurveda, Siddha, Unani and Homeopathy		
	a) Preparation and standardization of Ayurvedic formulations viz Aristas and		
	Asawas, Ghutika, Churna, Lehya and Bhasma.		
2	UNIT-II		
	A. Nutraceuticals		
	a. General aspects, Market, growth, scope and types of products available in		
	the market. Health benefits and role of Nutraceuticals in ailments like		
	Diabetes, CVS diseases, Cancer, Irritable bowel syndrome and various		
	Gastro intestinal diseases.		
	b. Study of following herbs as health food: Alfaalfa, Chicory, Ginger,		
	Fenugreek, Garlic, Honey, Amla, Ginseng,		
	Ashwagandha, Spirulina		
	B. Herbal-Drug and Herb-Food Interactions: General introduction to		
	interaction and classification. Study of following drugs and their possible side		
	effects and interactions:		
2	C. Hypercium, kava-kava, Ginkobiloba, Ginseng, Garlic, Pepper & Ephedra.		
3	Unit III		
	A. Herbal Cosmetics		
	Sources and description of raw materials of herbal origin used via, fixed oils,		
	waxes, gums colours, perfumes, protective agents, bleaching agents,		
	antioxidants in products such as skin care, hair care and oral hygiene products.		
	B. Herbal excipients:		
	Herbal Excipients – Significance of substances of natural origin as excipients –		
	colorants, sweeteners, binders, diluents, viscosity builders, disintegrants,		
	flavors & perfumes.		
	C. Harbal farmulations		
	C. Herbal formulations:		
	Conventional herbal formulations like syrups, mixtures and tablets and Novel		
	dosage forms like phytosomes		





4	 UNIT- IV A. Evaluation of Drugs WHO & ICH guidelines for the assessment of herbal drugs Stability testing of herbal drugs. 		
	Bioprospecting	the terms: Patent, and Biopiracy ects of Traditional I	ts of natural products: IPR, Farmers right, Breeder's right, Knowledge and Natural Products. Case
	C. Regulatory Issues - Regulations in India (ASU DTAB, ASU DCC), Regulation of manufacture of ASU drugs - Schedule Z of Drugs & Cosmetics Act for ASU drugs.		
5	UNIT-V A. General Introduction to Herbal Industry Herbal drugs industry: Present scope and future prospects. A brief account of plant based industries and institutions involved in work on medicinal and aromatic plants in India.		
	B. Schedule T – Good Manufacturing Practice of Indian systems of medicine a. Components of GMP (Schedule – T) and its objectives Infrastructural requirements, working space, storage area, machinery and equipments, standard operating procedures, health and hygiene, documentation and records.		
Mode of examinat ion			
Weighta ge	Continuous Mode Assessment	Sessional Exam	ESE
Distribut ion	10 Marks	15	75





Text	1.	Textbook of Pharmacognosy by Trease & Evans.	
book/s*	2.	Textbook of Pharmacognosy by Tyler, Brady & Robber.	
	3.	Pharmacognosy by Kokate, Purohit and Gokhale	
	4.	Essential of Pharmacognosy by Dr.S.H.Ansari	
	5.	Pharmacognosy & Phytochemistry by V.D.Rangari	
	6.	Pharmacopoeal standards for Ayurvedic Formulation	
		(Council of Research in Indian Medicine & Homeopathy)	
	7.	Mukherjee, P.W. Quality Control of Herbal Drugs: An	
		Approach to Evaluation of Botanicals. Business Horizons	
		Publishers, New Delhi, India, 2002.	
Other Referenc			
es			





So	chool:	SOP		
Programme:		B.Pharm		
Branch:		Semester: VI		
1	CourseCode	BP604T		
2	Course Title	Biopharmaceutics & Pharmacokinetics – Theory		
3	Credits	4		
4	Contact	3-1-0		
	Hours			
	(L-T-P)			
	Course Type	Compulsory		
5	Course	Upon completion of the course student shall be able to:		
	Objective	1. Understand the basic concepts in biopharmaceutics and		
		pharmacokinetics and		
		their significance.		
		2. Use of plasma drug concentration-time data to calculate the		
		pharmacokinetic parameters to describe the kinetics of drug absorption,		
		distribution, metabolism, excretion, elimination.		
		3. To understand the concepts of bioavailability and bioequivalence of drug products and their significance.		
		1		
		Understand various pharmacokinetic parameters, their significance & applications.		
6	Course	CO1: Students will be able to define and differentiate the meaning of		
	Outcomes	Biopharmaceutics and Pharmacokinetics		
		CO2: Students will be able to plan about basic concepts and importance of plasma drug concentration-time data to calculate the pharmacokinetic parameters to describe the kinetics of drug absorption, distribution, metabolism, excretion, elimination. CO3: Students will be able to categorize, sketch and relate various compartment models and their orientation while learning the parameters involved in the biopharmaceutical expression and infer the findings from such studies. CO4: Students will be able to correlate a study and interpret basic concepts, measurement and calculation of zero order and first order absorption rate constant involved in various biopharmaceutical and pharmacokinetics measurements. CO5: Students will be able to interpret various constraints in developing data-base for individuals in diseased conditions and compare with the functioning of normal person while incorporating the concept of pharmacokinetic study. CO6: Students will be able to analyze Non linear Pharmacokinetics.		





7	Course Description	This subject is designed to impart knowledge and skills of Biopharmaceutics and pharmacokinetics and their applications in pharmaceutical development,
8	-	
Unit I Introduction to Biopharmaceutics a. Absorption; Mechanisms of drug absorption influencing drug absorption though GIT, absoral extra-vascular routes, b. Distribution Tissue permeability of drugs, be volume of drug distribution, plasma and tissue factors affecting protein-drug binding. c. Kinetics of protein binding, Clinical significations		 Unit I Introduction to Biopharmaceutics a. Absorption; Mechanisms of drug absorption through GIT, factors influencing drug absorption though GIT, absorption of drug from Non per oral extra-vascular routes, b. Distribution Tissue permeability of drugs, binding of drugs, apparent, volume of drug distribution, plasma and tissue protein binding of drugs,
		UNIT- II 10 Hours
renal excretion of drugs, factors affecting renal excretion clearance, Non renal routes of drug excretion of drugs B. Bioavailability and Bioequivalence: Definition and bioavailability, absolute and relative bioavailability, bioavailability, <i>in-vitro</i> drug dissolution models, <i>in-vitro-in-v</i>		 B. Bioavailability and Bioequivalence: Definition and Objectives of bioavailability, absolute and relative bioavailability, measurement of bioavailability, <i>in-vitro</i> drug dissolution models, <i>in-vitro-in-vivo</i> correlations, C. bioequivalence studies, methods to enhance the dissolution rates and
Compartment models, Non compartment models, physiological mode compartment open model. (a). Intravenous Injection (Bolus) (b). Intravenous infusion and (c) Extra vascular administrations. B. Pharmacokinetics parameters - KE ,t1/2,Vd,AUC,Ka, Clt and C definitions methods of eliminations, C. understanding of their significance and application 4 UNIT-I A. Multicompartment models: Two compartment open model. B. Kinetics of multiple dosing, steady state drug levels,		A. Pharmacokinetics: Definition and introduction to Pharmacokinetics, Compartment models, Non compartment models, physiological models, One compartment open model. (a). Intravenous Injection (Bolus) (b). Intravenous infusion and (c) Extra vascular administrations. B. Pharmacokinetics parameters - KE ,t1/2,Vd,AUC,Ka, Clt and CLR- definitions methods of eliminations,
		 A. Multicompartment models: Two compartment open model. IV bolus B. Kinetics of multiple dosing, steady state drug levels, C. calculation of loading and mainetnance doses and their significance in





	T .		
Mode of examinat	UNIT- V Nonlinear Pharmacokinetics: a. Introduction, b. Factors causing Non-linearity. a. Michaelis-menton method of estimating parameters, c.Explanation with example of drugs. Theory		
ion			
Weighta ge	Continuous Mode Assessment	Sessional Exam	ESE
Distribut ion	10 Marks	15	75
Text book/s*	 Biopharmaceutics ar Applied biopharmaceutics and B.C.YU 4th edition, Bio pharmaceutics and Sunil B.Jaiswal, Pharmacokinetics: B Hand Book of Clinic by ADIS Health Scie Biopharmaceutics; E Clinical Pharmacokinand Thomas, N. Tozen, I Dissolution Mack, Publishing Company Biopharmaceutics and Thomas, N. Tozen, I Biopharmaceutics and Biopharmaceutics and Thomas, N. Tozen, I Biopharmaceutics and Biopharmaceutics	nd Pharmacokinetics; ceutics and pharmacokinetics. Prentice-Hall Inernational Pharmacokinetics and Pharmacokinetics. Vallabh Prakashan Poy Milo Glbaldi Donate al Pharmacokinetics. By Swarbrick inetics, Concepts and Lea and Febrger, Philon, Bioavailability at paragraphy, Pennsylvaniaceutics and Clinical and expanded by Ref. 7. on's Pharmaceutica	cokinetics, Leon Shargel and Andrew tional edition.USA cs-A Treatise, By D. M. Brahmankar Pitampura, Delhi ald, R. Mercel Dekker Inc. s, By Milo Gibaldi and Laurie Prescott d Applications: By Malcolm Rowland ladelphia, 1995. and Bioequivalence, By Abdou H.M,
Other Referenc es			





Sc	chool:	SOP	
Pr	ogramme:	B.Pharm	
Bı	ranch:	Semester: VI	
1	Course Code	BP605T	
2	Course Title	Pharmaceutical Biotechnology – Theory	
3	Credits	4	
4	Contact	3-1-0	
	Hours		
	(L-T-P)		
	Course Type	Compulsory	
5	Course Objective	Upon completion of the subject student shall be able to;	
	Objective	1. Understanding the importance of Immobilized	
		enzymes in Pharmaceutical Industries	
		2. Genetic engineering applications in relation to production of	
		pharmaceuticals	
		3. Importance of Monoclonal antibodies in Industries	
		4. Appreciate the use of microorganisms in fermentation technology	
6	Course	CO1: Students will be able to understand the importance of	
	Outcomes	Immobilized enzymes in Pharmaceutical Industries.	
		CO2: Students will be able to describe genetic engineering applications in relation to production of Pharmaceuticals	
CO3: Students will be able to conclude about antibodies in Industries		CO3: Students will be able to conclude about importance of Monoclonal antibodies in Industries	
		CO4: Students will be able to understand the use of microorganisms in fermentation technology	
		CO5: Students will be able to plan various fermentation methods	
		CO6: Students will be able to explore recent developments in fermentation technology.	





7	Course Description	• Biotechnology has a long promise to revolutionize the biological sciences and technology.
		• Scientific application of biotechnology in the field of genetic engineering, medicine and fermentation technology makes the subject interesting.
		• Biotechnology is leading to new biological revolutions in diagnosis, prevention and cure of diseases, new and cheaper pharmaceutical drugs.
		• Biotechnology has already produced transgenic crops and animals and the future promises lot more.
		• It is basically a research-based subject.
8	Outline syl	labus
	1	
	_	Unit I
		a) Brief introduction to Biotechnology with reference to Pharmaceutical Sciences.
		b) Enzyme Biotechnology- Methods of enzyme immobilization and applications.
		c) Biosensors- Working and applications of biosensors in Pharmaceutical Industries.
		Brief introduction to Protein Engineering.
		Use of microbes in industry. Production of Enzymes-General consideration - Amylase, Catalase, Peroxidase, Lipase, Protease, Penicillinase.
		Basic principles of genetic engineering.
	2	
		Unit II
		Types of immunity- humoral immunity, cellular immunity
		a). Structure of Immunoglobulins
		b). Structure and Function of MHC
		c). Hypersensitivity reactions, Immune stimulation and Immune suppressions.
		General method of the preparation of bacterial vaccines, toxoids, viral vaccine, antitoxins, serum-immune blood derivatives and other products relative to immunity. Storage conditions and stability of official vaccines
		Hybridoma technology- Production, Purification and Applications
		Blood products and Plasma Substituties.





3	Unit III a. Types of immunity- humoral immunity, cellular immunity		
b. Structure of Immunoglobulins Structure and Function of MHC Hypersensitivity reactions, Immune stimulation and Immun c. General method of the preparation of bacterial vacci vaccine, antitoxins, serum-immune blood derivative relative to immunity. Storage conditions and stability of official vaccines Hybridoma technology- Production, Purification and App Blood products and Plasma Substituties.			f bacterial vaccines, toxoids, viral blood derivatives and other products accines
4	Unit IV a) Immuno blotting te	chniques- ELISA, Wo	estern blotting, Southern blotting.
	b) Genetic organization	n of Eukaryotes and	Prokaryotes
c) Microbial genetics including transformation, transduction, plasmids and transposons. Introduction to Microbial biotransformation and applications. Mutation: Types of mutation/mutants.			ation, transduction, conjugation,
			and applications.
5	5 Unit V a. Fermentation methods and general requirements, study of media equipments, sterilization methods, aeration process, stirring.		
	b. Large scale pro	duction fermenter des	sign and its various controls.
	c. Study of the production of - penicillins, citric acid, Vitamin B12, Glutamic acid, Griseofulvin,		
	Blood Products: Collection, Processing and Storage of whole human blood, dried human plasma, plasma Substituties.		
Mode of examinat ion			
Weighta	Continuous Mode	Sessional Exam	ESE
ge Distribut	Assessment	15	75
ion	10 Marks	13	13





 B.R. Glick and J.J. Pasternak: Molecular Biotechnology: Principles and Applications of RecombinantDNA: ASM Press Washington D.C. RA Goldshy et. al.,: Kuby Immunology. J.W. Goding: Monoclonal Antibodies. Zaborsky: Immobilized Enzyme CRC Press, Degraland, Ohio. S.B. Primrose: Molecular Biotechnology (Second Edition) Blackwell Scientific Publication. Stanbury F., P., Whitakar A., and Hall J., S., Principles of fermentation technology, 2nd edition, Aditya books Ltd., New Delhi 	





School:		SOP	
Pi	ogramme:	B.Pharm	
Branch:		Semester: VI	
1	1 Course Code BP606T		
2	Course Title	Pharmaceutical Quality Assurance – Theory	
3	Credits	4	
4	Contact	3-1-0	
	Hours		
	(L-T-P)		
	Course Type	Compulsory	
5	Course	Upon completion, students will be familiar with various aspects of quality	
	Objective	control and quality assurance aspects of pharmaceutical industries. It deals	
		with the important aspects like cGMP, QC tests, documentation, quality	
		certifications and	
		regulatory affairs	
6	Course	CO1: Students will be able to describe about upon completion of the course student	
	Outcomes	shall be able to understand the cGMP aspects in a pharmaceutical industry	
		CO2: Students will be able to associate basic concepts and importance of appreciate the importance of documentation	
		CO3: Students will be able to interpret andunderstand basic concepts on quality certifications applicable to pharmaceutical industries	
CO4: Students will be able to summarize and understand the responsible & QC departments along with GLP and validation aspects		CO4: Students will be able to summarize and understand the responsibilities of QA & QC departments along with GLP and validation aspects	
		CO5: Students will be able to analyze Complaints and evaluation of complaints in addition to Handling of return goods	
		CO6: Students will be able to compare about all the documents used in the Pharmaceutical industry.	
_	Course	This course deals with the various aspects of quality control and quality	
7	Description	assurance aspects of pharmaceutical industries. It deals with the important	
		aspects like cGMP, QC tests, documentation, quality certifications and	
	regulatory		
8	Outline syllabus		





	UNIT – I
	 A. Quality Assurance and Quality Management concepts: Definition and concept of Quality control, Quality assurance and GMP B. Total Quality Management (TQM): Definition, elements, philosophies C. ICH Guidelines: purpose, participants, process of harmonization, Brief overview of QSEM, with special emphasis on Q-series guidelines, ICH stability testing guidelines Quality by design (QbD): Definition, overview, elements of QbD Programme, tools ISO 9000 & ISO14000: Overview, Benefits, Elements, steps for registration NABL accreditation: Principles and procedures
2	UNIT - II A. Organization and personnel: Personnel responsibilities, training,
	hygiene and personal records. Premises: Design, construction and plant layout, maintenance, sanitation, environmental control, utilities and maintenance of sterile areas, control of contamination. B. Equipments and raw materials: Equipment selection, purchase specifications, maintenance,
	C. purchase specifications and maintenance of stores for raw materials.
3	 UNIT - III A. Quality Control: Quality control test for containers, rubber closures and secondary packingmaterials. B. Good Laboratory Practices: General Provisions, Organization and Personnel, Facilities, Equipment, Testing Facilities Operation, Test and Control Articles, C. Protocol for Conduct of a Nonclinical Laboratory Study, Records and Reports, Disqualification of Testing Facilities
4	 UNIT - IV A. Complaints: Complaints and evaluation of complaints, Handling of return good, recalling and waste disposal. B. Document maintenance in pharmaceutical industry: Batch Formula Record, Master Formula Record, SOP, Quality audit, C. Quality Review and Quality documentation, Reports and documents, distribution records.





				www.sharda.ac.in
	5 UNIT - V A. Calibration and Validation: Introduction, definition and			
				roduction, definition and general
		principles of c	alibration, qualification	on and validation, importance and
		scope of validate	tion,	
		B. types of validation	ation, validation mast	ter plan. Calibration of pH meter,
		Qualification of	of UV-Visible spectro	ophotometer, General principles of
		Analytical meth	nod Validation.	
		C. Warehousing:	Good warehousing pr	ractice, materials management
	Mode of	Theory		
	examinat			
	ion	Cartina Mada	Sessional Exam	ECE
	Weighta ge	Continuous Mode Assessment	Sessional Exam	ESE
	Distribut	Assessment	15	75
	ion	10 Marks	13	13
	Text	1. Quality	Assurance Guide b	by organization of Pharmaceutical
	book/s*	Products	of India.	
		2. Good La	boratory Practice Reg	ulations, 2 nd Edition, Sandy Weinberg
		Vol. 69.		,
		3. Quality	Assurance of Pharma	ceuticals- A compendium of
				als Vol I WHO Publications.
		_	to Total Quality Mana	gement- Kushik Maitra and Sedhan K
		Ghosh		
		_	Practice GMP's – P P S	
				anagement – Sadhank G Ghosh
				poeia – Vol I, II, III, IV-
			•	and Quality specification for
			oratory Practices – M	cipients and Dosage forms
			lelines, ISO 9000 and	
$ldsymbol{\sqcup}$). 1011 guild		





Sc	hool:	SOP	
Pr	ogramme:	B. Pharm	
Bı	anch:	Semester: V	
1	Course	BP607P	
	Code		
2	Course	Medicinal Chemistry III - Practical	
	Title		
3	Credits	2	
4	Contact	0-0-4	
	Hours		
	(L-T-P)		
	Course	Compulsory	
5	Туре		
3	Course Objective	Upon completion of the course student shall be able to	
	_	Understand the importance of drug design and different techniques of drug design.	
		2. Understand the chemistry of drugs with respect to their biological activity.	
		3. Know the metabolism, adverse effects and therapeutic value of drugs.	
		4. Know the importance of SAR of drugs.	
6	Course Outcomes	CO1: Student will be able to understand the structure, chemistry and its correlation with the therapeutic value of drugs.	
		CO2: Student will be able to formulate of drugs and preparation of drugs.	
		CO3: Student will be able to apply the basic knowledge about the synthesis of sulphanilamide.	
		CO4: Student will be able to generalize about the synthesis of Chlorobutanol metronidazole.	
		CO5: Student will be able to conclude about the synthesis of chloroquine, dapsone	
CO6: Stu		CO6: Student will be able to plan about the synthesis of Tolbutamide.	





7	Course	I Pre	paration of drugs an	d intermediates
	Description	1 Sulp	hanilamide	
		2 7-Н	ydroxy, 4-methyl cou	marin
		3 Chle	orobutanol	
		4 Trip	henyl imidazole	
		5 Toll	outamide	
		6 Hex	amine	
		II Ass	y of drugs	
		1 Ison	icotinic acid hydrazid	le
		2 Chle	proquine	
		3 Met	ronidazole, Dapsone	
		4 Chle	orpheniramine maleat	e, Benzyl penicillin
8	Outline Sylla	DUS		
	1	Preparation of Sulphanilamide		
	2		hlorpheniramine male	eate
	3	Preparation of 7-Hydroxy, 4-methyl coumarin		
	4	Preparation of C		
	5	Preparation of C		
	6	Preparation of T	riphenyl imidazole	
	7	Preparation of B	enzyl penicillin	
	8	Preparation of P	reparation of Preparat	ion of Tolbutamide
	9	Preparation of M	1	
	10	Preparation of D		
	Mode of	Practical/Viva		
	examination			
	Weightage	Continuous	Sessional Exam	ESE
	Distribution	Mode		
		Assessment		
		05 Marks	10	35





Text book/s*	1. Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry.
	2. Foye's Principles of Medicinal Chemistry.
	3. Burger's Medicinal Chemistry, Vol I to IV.
	4. Introduction to principles of drug design- Smith and Williams.
	5. Remington's Pharmaceutical Sciences.
	6. Martindale's extra pharmacopoeia.





School:		SOP
Pr	ogramme:	B. Pharm
Br	anch:	Semester: VI
1	Course	BP608P
	Code	
2	Course	Pharmacology III- Practical
	Title	
3	Credits	2
4	Contact	0-0-4
	Hours	
	(L-T-P)	
	Course	Compulsory
	Type	
5	Course	Upon completion of this course the student should be able to
	Objective	1. Understand the mechanism of drug action and its relevance in the
		treatment of different diseases
		2. Demonstrate isolation of different organs/tissues from the laboratory
		animals by simulated experiments
		3. Demonstrate the various receptor actions using isolated tissue
preparation Appreciate correlation of pharmacology with relati		Appreciate correlation of pharmacology with related medical sciences
6	Course	CO1: Students would be able to define various instruments and
0	Outcomes	methods used in the evaluation of <i>in vitro</i> and <i>in vivo</i> evaluation of
	Outcomes	various drugs.
		various drugs.
		CO2: Students would be able to understand the working principles of the
		instruments used and actions of various drugs on biological systems.
		CO3: Students would be able to demonstrate the effects of various
		categories of drugs and bioassays of physiological substances.
		edicgories of drugs and bloussays of physiological substances.
		CO4: Students would be able to analyze the outcomes of experiments
		through simulation studies.
		CO5: Students would be able to evaluate the normal and abnormal
		physiological processes, and various drugs that can be employed for different
		treatment protocols.
		CO6: The student will be able to perform bioassays of various drugs.





7	Course			
	Description	1. Dose	e calculation in pha	rmacological experiments
	_	2. Anti	allergic activity by	mast cell stabilization assay
		3. Stud	ly of anti-ulcer acti	vity of a drug using pylorus
		ligar	nd (SHAY) rat m	odel and NSAIDS induced
		ulce	r model.	
		4. Stud	ly of effect of drugs	on gastrointestinal motility
		5. Effe	ct of agonist and ar	tagonists on guinea pig ileum
			mation of serum bi analyser	ochemical parameters by using semi-
		8. Effe	•	ve on frog intestine Insulin hypoglycemic
		9. Test	for pyrogens (rabl	pit method)
			ermination of acute n data	oral toxicity (LD50) of a drug from a
		11. Dete		skin irritation / corrosion of a test
			ermination of acute tance	eye irritation / corrosion of a test
		13. Calc	culation of pharmac	okinetic parameters from a given data
			-	n experimental pharmacology(
			ent's t test, ANOV	= = = = = = = = = = = = = = = = = = = =
			statistics methods in Wilcoxon Signed l	n experimental pharmacology (Chi square Rank test)
8	Outline Sylla	bus		
	1	Dose calculation in	pharmacological e	xperiments
	2			sing pylorus ligand (SHAY) rat model
		and NSAIDS induc		
	3	Study of effect of d		
	4	Effect of agonist an		
	5			meters by using semi- autoanalyser
	6			meters by using semi- autoanalyser
	8		rgative on frog inte	sune
	9	Insulin hypoglycemic effect in rabbit		
	_	Test for pyrogens ((D50) 6 1 6 1 1
	10		cute oral toxicity (LD50) of a drug from a given data
	Mode of examination	Practical/Viva		
	Weightage	Continuous	Sessional Exam	ESE
	Distribution	Mode	Sessional Laum	
		Assessment		
1	Ī			





	05 Marks	10	35
'Text book/s*	Paras medical Applications 2. Ghosh Market Hilton & Con 3. Kulkarni VallabhP	al publisher Mode, by Charles R.Crain IN. Fundamentals mpany, Kolkata, SK. Handboomakashan,	K., Principles of Pharmacology, ern Pharmacology with clinical ig& Robert, of Experimental Pharmacology. ok of experimental pharmacology.
Other References			





School:		SOP		
Pr	ogramme:	B. Pharm		
Br	anch:	Semester: VI		
1	Course	BP609P		
	Code			
2	Course	Herbal Drug Technology - Practical		
	Title			
3	Credits	2		
4	Contact	0-0-4		
	Hours			
	(L-T-P)			
	Course	Compulsory		
	Type			
5	Course	1. Explain correct use of various equipments in Pharmacognosy		
	Objective	laboratory.		
		2. Evaluation of drugs and various formulations		
		3. Demonstrate various herbal preparations		
		4. Formulations of various herbal cosmetics		
		5. Analyse the phytoconstituents		
6	Course	CO1: Students would able to prepare crude drug extract, identify extract		
	Outcomes	through phytochemical screening and also identify herbal drugs through		
		chemical test		
		CO2: Students would be able to distinguish excipients of natural origin and		
		estimate alcohol content in alcoholic formulations		
		CO3: Students would be able to formulate herbal creams, shampoo		
		CO4: Students would be able to analyze evaluation parameters for herbal		
		shampoo and creams		
		CO5: Students would be able to formulate and evaluate syrups.		
		CO6: Students will be able to understand about herbal drug preparations.		
7	Course	1. To perform preliminary phytochemical screening of crude drugs.		
	Description	2. Determination of the alcohol content of Asava and Arista		
		3. Evaluation of excipients of natural origin		
		4. Incorporation of prepared and standardized extract in cosmetic		
		formulations like creams, lotions and shampoos and their evaluation.		
		5. Incorporation of prepared and standardized extract in formulations like		
		syrups, mixtures and tablets and their evaluation as per Pharmacopoeial		
		requirements.		
		6. Monograph analysis of herbal drugs from recent Pharmacopoeias		
		7. Determination of Aldehyde content		
		8. Determination of Phenol content		
		Determination of total alkaloids		
8	Outline Sylla			
	1	To determine phenol content in clove oil.		
	2	To determine total alkaloids in crude drug sample.		





3	Evaluation of ex	cipients of natural ori	igin
4	Incorporation of	prepared and standar	dized extract in cosmetic formulations
	like creams, lotions and shampoos and their evaluation.		
5	To perform Preli	minary phytochemic	al screening of crude drugs.
6	To determine the	alcohol content of A	Asava and Arishta.
7	To formulate and	evaluate herbal crea	um containing Curcuma longa.
8	To formulate and	l evaluate Polyherbal	shampoo.
9	To formulate and	evaluate herbal cou	gh syrup
10	To perform Preli	minary phytochemic	al screening of crude drugs.
Mode of examination	Practical/Viva		
Weightage	Continuous	Sessional Exam	ESE
Distribution	Mode		
	Assessment		
	05 Marks	10	35
Text	1. Te	extbook of Pharmacognosy by Trease & Evans.	
book/s*	2. Te	xtbook of Pharmacog	gnosy by Tyler, Brady & Robber.
	3. Pharmacognosy by Kokate, Purohit and Gokhale		
	4. Es	ential of Pharmacogn	nosy by Dr.S.H.Ansari
	5. Ph	armacognosy & Phyt	ochemistry by V.D.Rangari





School:		SOP
	gramme:	B.Pharm
Bra	nch:	Semester: VII
1	Course Code	BP701T
2	Course Title	Instrumental Methods of Analysis
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Compulsory
5	Course Objective	Upon completion of this course the student should be able to 1. Understand the interaction of matter with electromagnetic radiations and its applications in drug analysis. 2. Understand various techniques in Analysis of various Pharmaceuticals. 3. Study the applications of various Instruments in analysis of Pharmaceuticals. 4. Perform quantitative & qualitative analysis of drugs using various analytical instruments.
6	Course	CO1: Students will be able to understand about Analytical techniques in
7	Outcomes	Instrumental Methods of Analysis. CO2 Students will be able to apply the principle of Modern instruments using in Analysis of Pharmaceuticals CO3: Students will be able to compare about applications of Modern instruments using in Analysis of Pharmaceuticals CO4: Students will be able to identify how to operate the Modern instruments in analysis of Pharmaceuticals. CO5: Students will be able to interpret the chromatographic separation and analysis of drugs. CO6: Students will be able to analyze the chromatographic separation and analysis of drugs
7	Course Description	This subject deals with the application of instrumental methods in qualitative and quantitative analysis of drugs. This subject is designed to impart a fundamental knowledge on the principles and instrumentation of spectroscopic and chromatographic technique. This also emphasizes on theoretical and practical knowledge on modern analytical instruments that are used for drug testing.
8	Outline syllabus	3





	www.sharda.ac.in
	 A. UV Visible spectroscopy Electronic transitions, chromophores, auxochromes, spectral shifts, solvent effect on absorption spectra, Beer and Lambert's law, Derivation and deviations. B. Instrumentation - Sources of radiation, wavelength selectors, sample cells, detectors- Photo tube, Photomultiplier tube, Photo voltaic cell, Silicon Photodiode. C. Applications - Spectrophotometric titrations, Single component and multi component analysis D. Fluorimetry
	A. IR spectroscopy Introduction, fundamental modes of vibrations in poly atomic molecules, sample handling, factors affecting vibrations Instrumentation - Sources of radiation, wavelength selectors, detectors - Golay cell, Bolometer, Thermocouple, Thermister, Pyroelectric detector and applications
	B. Flame Photometry- Principle, interferences, instrumentation and
	applications
	C. Atomic absorption spectroscopy- Principle, interferences,
	instrumentation and applications
	Nepheloturbidometry- Principle, instrumentation and applications
3	A. Introduction to chromatography Adsorption and partition column chromatography-Methodology,
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications
4	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications
4	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation,
4	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization,
4	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications
4	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction,
4	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory,
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications.
5	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography- Introduction, classification, ion
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography- Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process,
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography- Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography- Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications B. Gel chromatography- Introduction, theory, instrumentation and
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography—Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications B. Gel chromatography—Introduction, theory, instrumentation and applications Affinity
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography- Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications B. Gel chromatography- Introduction, theory, instrumentation and
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography—Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications B. Gel chromatography—Introduction, theory, instrumentation and applications Affinity
	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography- Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications B. Gel chromatography- Introduction, theory, instrumentation and applications Affinity C. chromatography- Introduction, theory, instrumentation and applications
5	advantages, disadvantages and applications. B. Thin layer chromatography- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications. C. Paper chromatography-Introduction, methodology, development techniques, advantages, disadvantages and applications Electrophoresis— Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications A. Gas chromatography - Introduction, theory, instrumentation, derivatization, temperature Programmeming, advantages, disadvantages and applications B. High performance liquid chromatography (HPLC)-Introduction, theory, C. instrumentation, advantages and applications. A. Ion exchange chromatography—Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications B. Gel chromatography—Introduction, theory, instrumentation and applications Affinity C. chromatography—Introduction, theory, instrumentation and applications of Theory





Weightage	Continuous	Sessional	ESE		
Distribution	Mode	Exam			
	Assessment				
	10 Marks	15	75		
Text book/s*					
Other	Recommended Books (Latest Editions)				
References	1. Instrur	nental Method	ds of Chemical Analysis by B.K Sharma		
	2. Organic spectroscopy by Y.R Sharma				
	3. Text b	ook of Pharma	aceutical Analysis by Kenneth A. Connors		
	4. Vogel Vogel	's Text book	of Quantitative Chemical Analysis by A.I.		
	5. Practic Stenlake	cal Pharmace	utical Chemistry by A.H. Beckett and J.B.		
	6. Organi	ic Chemistry b	oy I. L. Finar		
	7. Organi	ic spectroscop	y by William Kemp		
	8. Quanti	itative Analysi	is of Drugs by D. C. Garrett		
	9. Quanti P. D. Seth	•	is of Drugs in Pharmaceutical Formulations by		
	10. Spectr Silverstein		identification of Organic Compounds by		
			identification of Organic Compou		





School:		SOP
Pro	gramme:	B.Pharm
Bra	nch:	Semester: VII
1	Course Code	BP-702T
2	Course Title	Industrial Pharmacy-II Theory
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Compulsory
5	Course	Upon completion of the course, the student shall be able to:
	Objective	Know the process of pilot plant and scale up of pharmaceutical dosage forms
		2. Understand the process of technology transfer from lab scale to commercial batch
		3. Know different Laws and Acts that regulate pharmaceutical industry
		4. Understand the approval process and regulatory requirements for drug products
6	Course Outcomes	CO1: Students shall be able to understand about the process of pilot plant and scale up of pharmaceutical dosage forms
		CO2: Students shall be able to describe the process of technology transfer from lab scale to commercial batch.
		CO3: Students shall be able to plan about stepwise product development process from NDA filing to final FDA submission
		CO4 Students shall be to able to analyze the different laws and acts that regulate pharmaceutical industry in India and US
		CO5: Students shall be to able to Develop the concept of quality management and knowledge of required certifications
		CO6: Students will be able to design about all the process and regulatory requirements for drug product.
7	Course Description	This course is designed to impart fundamental knowledge on pharmaceutical product development and translation from laboratory to market
8	Outline syllabı	ıs
	Unit 1	A. Pilot plant scale up techniques:





- B. Pilot plant scale up techniques: General considerations including significance of personnel requirements, space requirements, raw materials, Pilot plant scale up considerations for solids, liquid orals, semi solids
- C. relevant documentation, SUPAC guidelines, Introduction to platform technology.

Unit 2

Technology development and transfer

- A. Technology development and transfer: WHO guidelines for Technology Transfer(TT): Terminology, Technology transfer protocol,
- B. Quality risk management, Transfer from R & D to production (Process, packaging and cleaning), Granularity of TT Process (API, excipients, finished products, packaging materials) Documentation, Premises and equipments, qualification and validation, quality control, analytical method transfer,
- C. Approved regulatory bodies and agencies, Commercialization practical aspects and problems (case studies), TT agencies in India APCTD, NRDC, TIFAC, BCIL, TBSE / SIDBI; TT related documentation confidentiality agreement, licensing, MoUs, legal issues

Unit 3

Regulatory affairs

- A. Regulatory affairs: Introduction, Historical overview of Regulatory Affairs, Regulatory authorities, Role of Regulatory affairs department,
- B. Responsibility of Regulatory Affairs Professionals Regulatory requirements for drug approval: Drug Development Teams, Non-Clinical Drug Development, Pharmacology, Drug Metabolism and Toxicology, General considerations of Investigational New Drug (IND) Application,
- C. Investigator's Brochure (IB) and New Drug Application (NDA), Clinical research / BE studies, Clinical Research Protocols, Biostatistics in Pharmaceutical Product Development, Data Presentation for FDA Submissions, Management of Clinical Studies

Unit 4

Quality management systems

- A. Quality management systems: Quality management & Certifications: Concept of Quality,
- B. Total Quality Management, Quality by Design (QbD), Six Sigma concept, Out of Specifications (OOS),
- C. Change control, Introduction to ISO 9000 series of quality systems standards, ISO 14000, NABL, GLP

Unit 5

Indian Regulatory Requirements

- A. Indian Regulatory Requirements: Central Drug Standard Control Organization (CDSCO) and State Licensing Authority:
- B. B. Organization, Responsibilities, Certificate of Pharmaceutical Product (COPP),
- C. Regulatory requirements and approval procedures for New Drugs

Mode of examination

Theory





Weightage	Continuous	Sessiona	ESE
Distribution	Mode	1 Exam	
	Assessment		
	10 Marks	15	75
		22 : 2 ***	
Text			ikipedia, the free encyclopedia modified
book/s	on 7th April a	available at 1	http,//en.wikipedia.org/wiki/Regulatory_
	Affairs.		
	2. International	Regulatory	Affairs Updates, 2005. available at
	http://www.iraup	•	•
			d S. Mantus. Text book of FDA Regulatory
	_		on Drugs, Medical Devices, and Biologics'
		ioi Piescripu	on Drugs, Medical Devices, and Biologics
	Second Edition.		
	4. Regulatory A	Affairs broug	tht by learning plus, inc. available at
	http://www.cgmp	o.com/ra.htm.	





Sch	ool:	SOP				
Pro	gramme:	B. Pharm				
Bra	inch:	Semester:VII				
1	Course Code	BP703T				
2	Course Title	Pharmacy Practice Theory				
3	Credits	4				
4	Contact	3-1-0				
	Hours					
	(L-T-P)					
	Course Type	Compulsory				
5	Course Objective	This course has been designed to impart the fundamental knowledge of pharmacy practice and ethics along with the aspects of hospital organization. Objectives: Upon the completion of this course the students shall be able to				
		1. know various drug distribution methods in a hospital				
		2. appreciate the pharmacy stores management and inventory control				
		3. monitor drug therapy of patient through medication chart review and				
		clinical review				
		4. obtain medication history interview and counsel the patients				
		5. identify drug related problems				
		6. detect and assess adverse drug reactions				
		7. interpret selected laboratory results (as monitoring parameters in therapeutics) of specific disease states				
		8. know pharmaceutical care services				
		9. do patient counselling in community pharmacy;				
		10. Appreciate the concept of rational drug therapy.				
6	Course	CO1: Student will be able to understand Ability to discuss the controversies				
	Outcomes	in drug therapy				
		CO2: Student will be able to apply the therapeutic approach to management				
		of hospital				
		CO3: Student will be able to identify the patient specific parameters relevant				
		in monitoring therapy				
		CO4: Student will be able to conclude the importance of individualized				
		therapeutic plans based on diagnosis CO5: Student will be able to analyze data collected at their research work.				
		· ·				
		CO6: Students shall be able to generalize the role of a Pharmacist in a community.				





7	Course Description	In the changing scenario of pharmacy practice in India, for successful practice of Hospital Pharmacy, the students are required to learn various skills like drug distribution, drug information, and therapeutic drug monitoring for improved patient care. In community pharmacy, students will
		be learning various skills such as dispensing of drugs, responding to minor
		ailments by providing suitable safe medication, patient counselling for
		improved patient care in the community set up.
8	Outline syllabi	
	Unit 1	Hospital and it's organization
		on, Classification of hospital- Primary, Secondary and Tertiary hospitals,
		sification based on clinical and non- clinical basis, Organization
	Structure of a l	Hospital, and Medical staffs involved in the hospital and their functions.
	staff required Classification idiosyncrasy sudden with interactions, interactions,	functions of hospital pharmacy, Organization structure, Location, Layout and rements, and Responsibilities and functions of hospital pharmacists ons - Excessive pharmacological effects, secondary pharmacological effects, y, allergic drug reactions, genetically determined toxicity, toxicity following thdrawal of drugs, Drug interaction- beneficial interactions, adverse and pharmacokinetic drug interactions, Methods for detecting drug spontaneous case reports and record linkage studies, and Adverse drug orting and management.
	requirement	n and structure of retail and wholesale drug store, types and design, Legal s for establishment and maintenance of a drug store, Dispensing of products, maintenance of records of retail and wholesale drug store.
	Unit 2	Drug distribution system in a hospital
	and la	sing of drugs to inpatients, types of drug distribution systems, charging policy belling, Dispensing of drugs to ambulatory patients, and Dispensing of lled drugs.
		ntents of hospital formulary, Differentiation of hospital formulary and Drug n and revision, and addition and deletion of drug from hospital formulary.
	Therap Monito	or Therapeutic Drug Monitoring, Factors to be considered during the beutic Drug Monitoring, and Indian scenario for Therapeutic Drug bring. Causes of medication non-adherence, pharmacist role in the medication nce, and monitoring of patient medication adherence.
		r the patient medication history interview, medication interview forms. erials, staff, and infrastructure requirements
	i manciai, mat	indis, sair, and infrastructure requirements

Pharmacy and therapeutic committee

Unit 3





- A. Organization, functions, Policies of the pharmacy and therapeutic committee in including drugs into formulary, inpatient and outpatient prescription, automatic stop order, and emergency drug list preparation. Drug and Poison information centre, Sources of drug information, Computerised services, and storage and retrieval of information.
- B. Definition of patient counseling; steps involved in patient counseling, and Special cases that require the pharmacist.Role of pharmacist in the education and training Programme, Internal and external training Programme, Services to the nursing homes/clinics, Code of ethics for community pharmacy, and Role of pharmacist in the interdepartmental communication and community health education.
- C. Prescribed medication order- interpretation and legal requirements, and Communication skills- communication with prescribers and patients.

Unit 4 Budget preparation and implementation

- A. Budget preparation and implementation.
- B. Introduction to Clinical Pharmacy, Concept of clinical pharmacy, functions and responsibilities of clinical pharmacist, Drug therapy monitoring medication chart review, clinical review, pharmacist intervention, Ward round participation, Medication history and Pharmaceutical care.

Dosing pattern and drug therapy based on Pharmacokinetic & disease pattern.

C. Introduction and sale of over the counter, and Rational use of common over the counter medications.

Unit 5 Drug store management and inventory control

- A. Organisation of drug store, types of materials stocked and storage conditions, Purchase and inventory control: principles, purchase procedure, purchase order, procurement and stocking, Economic order quantity, Reorder quantity level, and Methods used for the analysis of the drug expenditure Description, principles involved, classification, control, identification, role of hospital pharmacist, advisory committee.
- B. Description, principles involved, classification, control, identification, role of hospital pharmacist, advisory committee.
- C. Blood chemistry, hematology, and urinalysis.

	Mode of examination	Theory		
	Weightage Distribution	Continuous Mode Assessment	Sessional Exam	ESE
		10 Marks	15	75





Text book/s*	
Other	1. Merchant S.H. and Dr. J.S.Quadry. A textbook of hospital
References	pharmacy, 4th ed. Ahmadabad: B.S. Shah Prakakshan; 2001.
	2. Parthasarathi G, Karin Nyfort-Hansen, Milap C Nahata. A textbook of Clinical Pharmacy Practice- essential concepts and
	skills, 1 st ed. Chennai: Orient Longman Private Limited; 2004.
	3. William E. Hassan. <i>Hospital pharmacy</i> , 5th ed. Philadelphia: Lea & Febiger; 1986.
	4. Tipnis Bajaj. <i>Hospital Pharmacy</i> , 1 st ed. Maharashtra:
	Career Publications; 2008.
	5. Scott LT. Basic skills in interpreting laboratory data, 4thed. American
	Society of Health System Pharmacists Inc; 2009.
	5. Parmar N.S. Health Education and Community Pharmacy, 18th ed.
	India: CBS Publishers & Distributers; 2008.





School:		SOP			
Programme:		B. Pharm			
Bra	nch:	Semester: VII			
1	Course Code	BP 704T			
2	Course Title	Novel drug delivery systems- Theory			
3	Credits	4			
4	Contact Hours (L-T-P)	3-1-0			
	Course Type	Compulsory			
5	Course Objective	After the successful completion of this course, the student shall be able to: 1. The course aims to provide an understanding of basic knowledge on the area of novel drug delivery systems. 2. To understand various approaches for development of novel drug delivery systems. 3. To understand the criteria for selection of drugs and polymers for the development of Novel drug delivery systems, their formulation and evaluation			
6	Course Outcomes	CO1: The students will understand the concepts and applications of Novel Drug Delivery Systems and to study various properties for sustained and controlled drug delivery systems. CO2: The student will be able to apply knowledge in developing various novel formulations as per requirements and to learn mucosal and Implantable drug delivery. CO3: The student will be able to analyze various evaluation parameters for oral, parenteral, topical etc. drug delivery systems. CO4: The students will be able to formulate industrially feasible, cost effective strategy for development of new dosage forms CO5: The students will be able to plan about site specific drug delivery. CO6: The Students will be able to develop ocular drug delivery, its issues and challenges, drug selection.			
7	Course Description	This subject is designed to impart basic knowledge on the area of novel drug delivery systems. 1. To understand various approaches for development of novel drug delivery systems. 2. To understand the criteria for selection of drugs and polymers for the development of Novel drug delivery systems, their formulation and evaluation			
8	Outline syllabus	5			





	www.snarga.ac.in				
	A. Controlled drug delivery systems: Introduction, terminology/definitions and rationale, advantages, disadvantages, selection of drug candidates. Approaches to design controlled release formulations based on diffusion, dissolution and ion exchange principles. Physicochemical and biological properties of drugs relevant to controlled release formulations				
	B. Polymers: Introduction, classification, properties,				
	C. advantages and application of polymers in formulation of controlled release drug delivery systems.				
	A. Microencapsulation: Definition, advantages and disadvantages, microspheres /microcapsules, microparticles, methods of microencapsulation, applications				
	 B. Mucosal Drug Delivery system: Introduction, Principles of bioadhesion / mucoadhesion, concepts, advantages and disadvantages, transmucosal permeability and formulation considerations of buccal delivery systems C. Implantable Drug Delivery Systems: Introduction, advantages and disadvantages, concept of implantsand osmotic pump 				
3	 A. Transdermal Drug Delivery Systems: Introduction, Permeation through skin, factors affecting permeation, permeation enhancers, basic components of TDDS, formulation approaches B. Gastroretentive drug delivery systems: Introduction, advantages, disadvantages C. approaches for GRDDS – Floating, high density systems, inflatable and gastroadhesive systems and their applications 				
4	 A. Targeted drug Delivery: Concepts and approaches advantages and disadvantages, B. introduction to liposomes, niosomes, nanoparticles, C. monoclonal antibodies and their applications 				
5	 A. Ocular Drug Delivery Systems: Introduction, intra ocular barriers and methods to overcome –Preliminary study, ocular formulations and ocuserts B. Intrauterine Drug Delivery Systems: Introduction, advantages and disadvantages, C. development of intra uterine devices (IUDs) and applications 				
	c. actophicit of milital aternic devices (1003) and applications				





Mode of	Theory		
examination	-		
Weightage	Continuous	Sessional	ESE
Distribution	Mode	Exam	
	Assessment		
	10 Marks	15	75
Text book/s*			
Other	Recommende	ed Books (Lat	est Editions)
References	1. Y W.	Chien, Novel 1	Drug Delivery Systems, 2nd edition, revised
	and ex	panded,	
	2. Marce	l Dekker, Inc.,	New York, 1992.
	3. Robins	son, J. R., Lee	e V. H. L, Controlled Drug Delivery
	Systen	ns, Marcel Del	kker,
	4. Inc., N	lew York, 199	2.
	5. Encyc	lopedia of Co	ntrolled Delivery. Edith Mathiowitz,
	Publis	hed by Wiley	•
	6 Intorga	ionas Dublica	tion John Quantitativa Analysis of Days in
			tion, John Quantitative Analysis of Drugs in nulations by P. D. Sethi
	1 1141111	accutical Polit	idiations by 1. D. Sedii
		-	identification of Organic Compounds by
	Silvers	stein.	





Sch	ool:	SOP			
Pro	gramme:	B.Pharm			
	nch:	Semester: VII			
1	Course Code	BP705P			
2	Course Title	Instrumental Methods of Analysis- theory			
3	Credits	2			
4	Contact	0-0-4			
	Hours				
	(L-T-P)				
	Course Type	Compulsory			
5	Course	Upon completion of this course the student should be able to			
	Objective	1. Understand applications of instruments in drug analysis			
		2. Understand operation of instruments in Analysis of various			
		Pharmaceuticals.			
		3. Study the preparation of various analytes			
		4. Perform quantitative & qualitative analysis of drugs using various			
		analytical instruments.			
6	Course	CO1: Students will be able to understand Analytical techniques in			
	Outcomes	Instrumental Methods of Analysis.			
		CO2: Students will be able to describe conceptual knowledge about			
		operation of Modern instruments using in Analysis of Pharmaceuticals			
		CO3: Students will be able to apply the basic knowledge about applications of Modern instruments using in Analysis of Pharmaceuticals			
		applications of Wodern instruments using in Analysis of Pharmaceuticals			
		CO4: Student will be able to plan about how to operate the Modern			
		instruments in analysis of Pharmaceuticals.			
		modulients in analysis of Filarmaceuticals.			
		CO5: Students will be able to understand the chromatographic			
		separation and analysis of drugs.			
		CO6 Students will be able to interpret the chromatographic separation and			
		analysis of drugs.			
<u> </u>		minipolo of diago.			





	Text book/s*		•		
		05 Marks	10	35	
		Assessment	1.0		
	Distribution	Mode	Exam		
	Weightage	Continuous	Sessional	ESE	
	examination				
	Mode of	Practical	<u> </u>	, <u>, , , , , , , , , , , , , , , , , , </u>	
	12			layer chromatography	
	11			y paper chromatography	
	10			and sulphates by nephelo turbidometry	
	9	Determination of sodium by flame photometry Determination of potassium by flame photometry			
	8	Study of quenching of fluorescence			
	7	Estimation of quinine sulfate by fluorimetry.			
	5	Assay of paracetamol by UV- Spectrophotometry			
	5	spectroscopy.			
	4	Simultaneous		of ibuprofen and paracetamol by UV	
	3			by colorimetry	
	2	Estimation of		_	
	1			otion maxima and effect of solvents on ic compounds.	
3	Outline syllabu	ıs			
		2. Estima 3. Estima 4. Simult spectro 5 Assay of 6 Estimat 7 Study of 8 Determ 9 Determ 10 Determ 11 Separa 12 Separa 13 Separa 14 Demon	ation of dextro ation of sulfan caneous estimates oscopy of paracetamolation of quinine f quenching of ination of sodi ination of pota nination of chation of amino ation of sugars ation of plant paracetrism	of organic compounds. ose by colorimetry ilamide by colorimetry nation of ibuprofen and paracetamol by UV by UV- Spectrophotometry sulfate by fluorimetry f fluorescence ium by flame photometry assium by flame photometry lorides and sulphates by nephelo turbidometry acids by paper chromatography by thin layer chromatography oigments by column chromatography riment on HPLC riment on Gas Chromatography	
7	Course Description	Pharmaceutica	als	various modern instruments used in analysis of absorption maxima and effect of solvents on	
7	Course	Subject cover	s operation of	various modern instruments used in analy	





Other	Recommended Books (Latest Editions)
References	1. Instrumental Methods of Chemical Analysis by B.K Sharma
	2. Organic spectroscopy by Y.R Sharma
	3. Text book of Pharmaceutical Analysis by Kenneth A. Connors
	4. Vogel's Text book of Quantitative Chemical Analysis by A.I. Vogel
	5. Practical Pharmaceutical Chemistry by A.H. Beckett and J.B. Stenlake
	6. Organic Chemistry by I. L. Finar
	7. Organic spectroscopy by William Kemp
	8. Quantitative Analysis of Drugs by D. C. Garrett
	9. Quantitative Analysis of Drugs in Pharmaceutical Formulations by
	P. D. Sethi
	Spectrophotometric identification of Organic Compounds by Silverstein





School:		SOP			
	gramme:	B.Pharm			
	nch:	Semester: VIII			
1	Course Code	BP801T			
2	Course Title	Biostatistics & Research Methodology- Theory			
3	Credits	4			
4	Contact Hours (L-T-P)	3-1-0			
	Course Type	Compulsory			
5	Course Objective	 Know the operation of M.S. Excel, SPSS, R and MINITAB[®], DoE (Design of Experiment) Know the various statistical techniques to solve statistical problems Appreciate statistical techniques in solving the problems 			
6	Course Outcomes	CO1:Students will be able to Describe statistics, Biostatistics and interpretation of frequency distribution table, and their pharmaceutical examples CO2: Students will be able to Calculate the measures of central tendency and dispersion of a data and describe the method used for analysis, including a discussion of advantages, disadvantages, and necessary assumptions. CO3: Students will be able to Describe the properties of Regression, Curve fitting, Multiple regression and their pharmaceutical examples. CO4: Students will be able to Calculate and interpret the correlation between two variables. CO5: Students will be able to Understand the concept of Probability and different types of distributions, Poisson's distribution and its properties. CO6: Students will be able to plan about the concept of Probability and different types of distributions, Poisson's distribution and its properties.			
7	Course Description	To understand the applications of Biostatics in Pharmacy. This subject deals with descriptive statistics, Graphics, Correlation, Regression, logistic regression Probability theory, Sampling technique, Parametric tests, Non Parametric tests, ANOVA, Introduction to Design of Experiments, Phases of Clinical trials and Observational and Experimental studies, SPSS, R and MINITAB statistical software's, analyzing the statistical data using Excel.			
8	Outline syllabus				





 T	
1	 Unit-I A. Introduction: Statistics, Biostatistics, Frequency distribution B. Measures of central tendency: Mean, Median, Mode-Pharmaceutical examples
	C. Measures of dispersion: Dispersion, Range, standard deviation, Pharmaceutical problems
2	D. Correlation: Definition, Karl Pearson's coefficient of correlation, Multiple correlation - Pharmaceuticals examples
2	 Unit-II A. Regression: Curve fitting by the method of least squares, fitting the lines y= a + bx and x = a + by, Multiple regression, standard error of regression—Pharmaceutical Examples
	B. Probability: Definition of probability, Binomial distribution, Normal distribution, Poisson's distribution, properties - problems Sample, Population, large sample, small sample, Null hypothesis, alternative hypothesis, sampling, essence of sampling, types of sampling, Error-I type, Error-II type, Standard error of mean (SEM) - Pharmaceutical examples
	C. Parametric test: t-test(Sample, Pooled or Unpaired and Paired), ANOVA, (One way and Two way), Least Significance difference
3	Unit-III A. Non Parametric tests: Wilcoxon Rank Sum Test, Mann-Whitney U test, Kruskal-Wallis test, Friedman Test B.Introduction to Research: Need for research, Need for design of Experiments, Experiential Design Technique, plagiarism C. Graphs: Histogram, Pie Chart, Cubic Graph, response surface plot, Counter Plot graph Designing the methodology: Sample size determination and Power of a study, Report writing and presentation of data, Protocol, Cohorts studies,
	Observational studies, Experimental studies, Designing clinical trial, various phases.
4	Unit-IV A. Blocking and confounding system for Two-level factorials Regression modeling: Hypothesis testing in Simple and Multiple regression models B. Introduction to Practical components of Industrial and Clinical
	 Trials Problems: ®Statistical Analysis Using Excel, SPSS, MINITAB DESIGN OF EXPERIMENTS, R C. Online Statistical Software's to Industrial and Clinical trial approach
5	Unit-V A. Design and Analysis of experiments: Factorial Design: Definition, 2, 2 design. Advantage of factorial design B. Response Surface methodology: Central composite design,
	C. Historical design, Optimization Techniques





Mode	of	Theory		
examination	1			
Weightage		Continuous	Sessional	ESE
Distribution	l	Mode	Exam	
		Assessment		
		10 Marks	15	75
Text book/s	*			
Other		Recommende	ed Books (Lat	test edition):
References		1. Pharm	aceutical star	tistics- Practical and clinical applications,
		Sanfor	d Bolton, pub	lisher Marcel Dekker Inc. NewYork.
		2. Funda	mental of S	Statistics – Himalaya Publishing House-
		S.C.G	uptha	
		3. Design	n and Analys	sis of Experiments –PHI Learning Private
		Limite	ed, R.	
		Panne	rselvam,	
		4. Design	n and Analysis	of Experiments – Wiley Students Edition,
		Dougl	as and C. Mon	tgomery





School:		SOP			
Pro	gramme:	B.Pharm			
Branch:		Semester: VIII			
1	Course Code	BP802T			
2	Course Title	Social & Preventive Pharmacy- Theory			
3	Credits	4			
4	Contact Hours (L-T-P)	3-1-0			
	Course Type	Compulsory			
5	Course Objective	 Upon completion of this course the student should be able to 1. Acquire high consciousness/realization of current issues related to health and pharmaceutical problems within the country and worldwide. 2. Have a critical way of thinking based on current healthcare development. 3. Evaluate alternative ways of solving problems related to health and pharmaceutical issues 			
6	Course Outcomes	CO1: The students will understand the issues related to social issues of health. CO2: The student will be able to summarize the impact of govt. health policies run for various health issues. CO3: The student will be able to apply the knowledge of understanding the health issues of the society on finding the effective solution for eradication of diseases. CO4: The students will compare the correlation of various factors affecting the health status of common people and will assess the action plans to combat the health issues. CO5: The students would evaluate the processes of various national Programmes related to social health and prevention of various diseases. CO6: The students would elaborate the processes of various .preventions of diseases			
7	Course Description	The purpose of this course is to introduce to students a number of health issues and their challenges. This course also introduced a number of national health Programmemes. The roles of the pharmacist in these contexts are also discussed.			
8	Outline syllabus	3			





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	1	UnitI:
		A. Concept of health and disease: Definition, concepts and
		evaluation of public health. Understanding the concept of prevention and
		control of disease, social causes of diseases and social problems of the
		sick.
		D. Social and health education: Food in relation to nutrition and
		health, Balanced diet, Nutritional deficiencies, Vitamin
		deficiencies, Malnutrition and its prevention.
		Sociology and health: Socio cultural factors related to health and disease,
		Impact of urbanization on health and disease, Poverty and health
		Hygiene and health: personal hygiene and health care; avoidable habits
	2	Unit II:
	-	A. Preventive medicine: General principles of prevention and control
		of diseases such as cholera, SARS,
		B. Ebola virus, influenza, acute respiratory infections, malaria,
		chicken guinea, dengue, lymphatic
		C. filariasis, pneumonia, hypertension, diabetes mellitus, cancer, drug
		addiction-drug substance abuse
	3	Unit III:
		A. National health Programmes, its objectives, functioning and
		outcome of the following: HIV AND AIDS control
		Programmeme, TB, Integrated disease surveillance Programme
		(IDSP),
		B. National leprosy control Programmeme, National mental health
		Programme, National Programmeme for prevention and control of
		deafness,
		C. Universal immunization Programmeme, National Programmeme
		for control of blindness, Pulse polio Programmeme.
1	4	Unit IV:
	'	A. National health intervention Programmeme for mother and child,
		National family welfare Programmeme, National tobacco control
		•
		Programmeme,
		B. National Malaria Prevention Programme, National Programmeme
		for the health care for the elderly,
		C. Social health Programmeme; role of WHO in Indian national
_		Programme
	5	Unit V:
		A. Community services in rural, urban and school health: Functions of
		PHC, Improvement in rural sanitation, national urban health
		mission, Health promotion and education in school.
	M. 1. C	
	Mode of	Theory
	examination	
	Weightage	Continuous Sessional ESE
	Distribution	Mode Exam
		Assessment
		1 1





	10 Marks	15	75			
Text book/s*						
Other	Recommen	ded Books (La	test Edition	ıs)		
References	1. Shor	t Textbook of	Preventive	and Social	Medicine,	Prabhakara
	GN,	2 2010, ISI	BN: 978938	80704104, J	JAYPEE P	ublications
	Edit	ion,				
		book of Preven			` •	1 /
		ed by Royth Ed	ition, 2013, 1	ISBN: 97893	,	
			lath,	Saha	Indranil,	, 4
		ications				
		ew of Prev				`
		tatistics), Jain V		ition, 2014, IS	SBN: 97893	351522331,
		PEE Publication				
		ntials of Cor	•	Iedicine—A	Practical	Approach,
		math Lalita D, 1				
		math Dhananja	•			·
		PEE Publication			Preventive	and Social
		icine, K Park, 2	,	*		D
		N-14: 9788				
		SLISHERS. Co	•	harmacy Prac	ctice, Rame	esh Adepu,
	BSP	publishers, Hy	derabad			





School:		SOP
Pro	gramme:	B.Pharm
	nch:	Semester: VIII
1	Course Code	BP803ET
2	Course Title	Pharma marketing management- Theory
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Compulsory
5	Course Objective	After the successful completion of this course, the student shall be able to:
		 The course aims to provide an understanding of marketing concepts and techniques and their applications in the pharmaceutical industry Have a critical way of thinking based on different marketing strategy for product development. The aim here is to develop a community around the brand whereby audiences can interact with certain content. In the pharmaceutical industry, more so big pharma, and just like most other consumer-facing industries, there are more products and more messages subsequently meaning more noise.
6	Course Outcomes	CO1: The students will understand the Marketing concepts and techniques and the application of the same in the pharmaceutical industry. CO2: The student will be able to summarize the Market research and distribution channels along with their implementation in the pharmaceutical industry. CO3: The student will be able to apply the knowledge regarding the Concepts of product line and product mix decisions, branding and product management. CO4: The students will analyze the Theories on promotion techniques for OTC Products, sales and pricing of a product.
7	Course Description	CO5: The students would evaluate the processes of issues in price management in the pharmaceutical industry. CO6: The students will be able to conclude about the Emerging concepts in marketing and Global Marketing concept. The pharmaceutical industry not only needs highly qualified researchers, chemists and, technical people, but also requires skilled managers who can take the industry forward by managing and taking the complex decisions
		which are imperative for the growth of the industry. The Knowledge and Know-how of marketing management groom the people for taking a challenging role in Sales and Product management.





8	Outline syllabus	
	1	Unit I:
		A.Marketing:
		Definition, general concepts and scope of marketing; Distinction between
		marketing & selling; Marketing environment; Industry and competitive
		analysis; Analyzing consumer buying behavior; industrial buying
		behavior.
		A. Pharmaceutical market:
		Quantitative and qualitative aspects; size and composition of the market;
		demographic descriptions and socio-psychological characteristics of the
		consumer; market segmentation& targeting.Consumer profile;
		C.Motivation and prescribing habits of the physician; patients' choice of
		physician and retail pharmacist. Analyzing the Market; Role of market
		research.
	_	
	2	Unit II:
		A. Productdecision:
		Classification, product line and product mix decisions, product life
		cycle, product portfolio analysis; product positioning;
		B. New product decisions; Product branding, packaging and labeling
		decisions, C. Product management in pharmaceutical industry.
	3	Unit III:
	3	A. Promotion:
		Methods, determinants of promotional mix, promotional budget;
		B. An overview of personal selling, advertising, direct mail, journals,
		sampling, retailing,
		C. medical exhibition, public relations, online promotional techniques
		for OTC Products.
	4	Unit IV:
		A. Pharmaceutical marketing channels:
		Designing channel, channel members, selecting the appropriate
		channel, conflict in channels, physical distribution management:
		Strategic importance, tasks in physical distribution management.
		B. Professional sales representative (PSR):
		Duties of PSR, purpose of detailing, selection and training, supervising,
		norms for customer calls, motivating, evaluating,
		C. compensation and future prospects of the PSR.





5	Unit V:			
	A. Pricin	g:		
	Meaning, importance, objectives, determinants of price; pricing methods and strategies, issues in price management in pharmaceutical industry. An overview of DPCO (Drug Price Control Order) and NPPA (National Pharmaceutical Pricing Authority).			
	B. Emerg	ging concepts	in marketing:	
	Vertical & Ho	orizontal Mark	teting;	RuralMarketing;
	Consumerism	Consumerism;		
	C. Industr	rial Marketing	; Global Marketing.	
Mode of	Theory			
examination				
Weightage	Continuous	Sessional	ESE	
Distribution	Mode	Exam		
	Assessment			
	10 Marks	15	75	
Text book/s*				





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Other	Recommended Books: (Latest Editions)
References	1. Philip Kotler and Kevin Lane Keller: Marketing Management,
	Prentice Hall of India, New Delhi
	2. Walker, Boyd and Larreche: Marketing Strategy- Planning and
	Implementation, Tata MC GrawHill, New Delhi.
	3. Dhruv Grewal and Michael Levy: Marketing, Tata MC Graw Hill
	4. Arun Kumar and N Menakshi: Marketing Management, Vikas
	Publishing, India
	5. Rajan Saxena: Marketing Management; Tata MC Graw-Hill (India Edition)
	6. Ramaswamy, U.S & Nanakamari, S: Marketing
	Managemnt:Global Perspective,
	IndianContext, Macmilan India, New Delhi.
	7. Shanker, Ravi: Service Marketing, Excell Books, New Delhi
	8. Subba Rao Changanti, Pharmaceutical Marketing in India (GIFT –
	Excel series) Excel
	7. Rabindra Nath, Saha Indranil, 4
	Publications
	Review of Preventive and Social Medicine (Including
	Biostatistics), Jain Vivek, 6 Edition, 2014, ISBN: 9789351522331,
	JAYPEE Publications
	8. Essentials of Community Medicine—A Practical Approach,
	Hiremath Lalita D, nd
	9. Hiremath Dhananjaya A, 2 Edition, 2012, ISBN: 9789350250440,
	JAYPEE Publications Park Textbook of Preventive and Social
	Medicine, K Park, 21 Edition, 2011,
	10. ISBN-14: 9788190128285, BANARSIDAS BHANOT
	PUBLISHERS. Community Pharmacy Practice, Ramesh Adepu,
	BSP publishers, Hyderabad





School:		SOP
Pr	ogramme:	B.Pharm
Br	anch:	Semester: V
1	Course	BP804ET
	Code	
2	Course	Pharmaceutical Regulatory Science - Theory
	Title	
3	Credits	4
4	Contact	3-1-0
	Hours	
	(L-T-P)	
	Course	Compulsory
_	Type	
5	Course Objective	Upon completion of the subject student shall be able to;
		Know about the process of drug discovery and development
		Know the regulatory authorities and agencies governing the manufacture and sale of pharmaceuticals
		Know the regulatory approval process and their registration in Indian and international markets
6	Course Outcomes	CO1: Students would be able to understand about the process of drug discovery and development
		CO2: Students would be able to apply the regulatory authorities and agencies governing the manufacture and sale of pharmaceuticals CO3: Students would be able to explain the regulatory approval process CO4: Students would be able to analyze the registration of drug in Indian and international market. CO5 Students would be able to plan about clinical trials, ethical committee and protocol designing.
7	Course Descriptio n	This course is designed to impart the fundamental knowledge on the regulatory requirements for approval of new drugs, and drug products in regulated markets of India & other countries like US, EU, Japan, Australia,UK etc. It prepares the students to learn in detail on the regulatory requirements, documentation requirements, and registration procedures for marketing the drug products.
8	Outline sylla	abus





	T			
1	Unit I			
		scovery and devel	opment	
	Stages of drug disco	•	· P······	
		•	linical studies, non-clinical activities,	
	clinical studies,	1 / 1	,	
		generics, Concept	of generics, Generic drug product	
	development.	, ,		
2	Unit II			
	A. Regulatory A	pproval Process		
	Approval processes	and timelines invol	ved in Investigational New Drug (IND),	
	New Drug Applicat	ion (NDA), Abbre	viated New Drug Application (ANDA).	
	Changes to an appro	, ,	- , , , ,	
		uthorities and age		
			ndia, United States, European Union,	
	_	•	on structure and types of applications)	
3	Unit III	eunada (erganizaa	on survey and types of approximas)	
		of Indian drug nra	oduct in overseas market	
		0 1	eal products, Technical documentation,	
			Technical Document (CTD), electronic	
	Common Technical	(Bivii), Common	recimical Bocament (C1B), electronic	
		D) ASEAN Con	nmon Technical Document (ACTD)	
	research.	<i>5)</i> ; <i>HSE/H</i> (Con	minori Teelimeer Becament (TeTB)	
	pelletization process			
			s	
	C.equipments for manufacture of pellets Unit IV			
4	A. Clinical trials			
•			titutional Review Board / Independent	
	1 2	<u> </u>	ing procedures, Informed consent process	
	and procedures,		mis process, micrimes consent process	
	_	of Investigators, sp	ponsors & Monitors, C.Managing and	
	_		ce - safety monitoring in clinical trials	
5	Unit V			
	A. Regulatory C	Concepts		
	Basic terminology,	guidance, guideline	es, regulations, Laws and Acts, Orange	
	book, Federal Register, Code of Federal Regulatory, Purple book			
Mode of	Theory			
examinatio				
n				
Weightage	Continuous Mode	Sessional Exam	ESE	
Distributio	Assessment			
n	10 Marks	15	75	
			•	





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Text book/s*	1. Drug Regulatory Affairs by Sachin Itkar, Dr. N.S. Vyawahare, Nirali Prakashan.
	2. The Pharmaceutical Regulatory Process, Second Edition Edited by Ira
	R. Berry and Robert P. Martin, Drugs and the Pharmaceutical
	Sciences, Vol. 185. Informa Health care Publishers.
	3. New Drug Approval Process: Accelerating Global Registrations By
	Richard A Guarino, MD, 5 th edition, Drugs and the Pharmaceutical
	Sciences, Vol. 190.
	4. Guidebook for drug regulatory submissions / Sandy Weinberg. By
	John Wiley & Sons. Inc.
	5. FDA Regulatory Affairs: a guide for prescription drugs, medical
	devices, and biologics /edited by Douglas J. Pisano, David Mantus.
	6. Generic Drug Product Development, Solid Oral Dosage forms, Leon
	Shargel and Isader Kaufer, Marcel Dekker series, Vol.143
	7. Clinical Trials and Human Research: A Practical Guide to Regulatory
	Compliance By Fay A. Rozovsky and Rodney K. Adams
	8. Principles and Practices of Clinical Research, Second Edition Edited by
	John I. Gallin and Frederick P. Ognibene
	9. Drugs: From Discovery to Approval, Second Edition By Rick Ng





Sch	ool:	SOP
Pro	gramme:	B.Pharm
Bra	nch:	Semester: VIII
1	Course Code	BP805ET
2	Course Title	Pharmacovigilance Theory
3	Credits	4
4	Contact Hours (L-T-P)	3-1-0
	Course Type	Elective
5	Course Objective	This course has been designed to impart the fundamental knowledge of pharmacy practice and ethics along with the aspects of hospital organization. Objectives: Upon the completion of this course the students shall be able to Tell Why drug safety monitoring is important? History and development of pharmacovigilance National and international scenario of pharmacovigilance Dictionaries, coding and terminologies used in pharmacovigilance Detection of new adverse drug reactions and their assessment International standards for classification of diseases and drugs Adverse drug reaction reporting systems and communication in Methods to generate safety data during pre clinical, clinical and post approval phases of drugs' life cycle Drug safety evaluation in paediatrics, geriatrics, pregnancy and lactation Pharmacovigilance Programme of India (PvPI) requirement for ADR reporting in India ICH guidelines for ICSR, PSUR, expedited reporting, pharmacovigilance planning CIOMS requirements for ADR reporting Writing case narratives of adverse events and their quality.
6	Course	Student will be able to
	Outcomes	CO1: Understant the history of Pharmacovigilance
		CO2: Apply about the Detection of new adverse drug reactions and their assessment CO3: Analyze the suspected drug events CO4: Interpret the importance of PV PI CO5: Understand reports like ICSR, PSUR CO6: Apply the ADR for vaccines





7	Course	This paper will provide an opportunity for the student to learn about		
	Description	development of pharmacovigilance as a science, basic terminologies used		
		in pharmacovigilance, global scenario of Pharmacovigilance, train		
		students on establishing pharmacovigilance Programmeme in an		
		organization, various methods that can be used to generate safety data and		
		signal detection. This paper also develops the skills of classifying drugs,		
		diseases and adverse drug reactions.		
8	Outline syllabus	3		
	1	A. Introduction to Pharmacovigilance		
		a. History and development of Pharmacovigilance		
		b. Importance of safety monitoring of Medicine		
		c. WHO international drug monitoring Programmeme		
		B. Pharmacovigilance Programme of India(PvPI)		
		C. Introduction to adverse drug reactions		
		Definitions and classification of ADR		
		Detection and reportin		
		Methods in Causality assessment		
		Severity and seriousness assessment		
		Predictability and preventability assessment prevention.		
		Management of adverse drug reactions		
		Basic terminologies used in pharmacovigilance		
		Terminologies of adverse medication related events		
		Regulatory terminologies		





2	Drug and disease classification
	A. Drug dictionaries and coding in pharmacovigilance
	Anatomical, therapeutic and chemical classification of drugs
	International classification of diseases
	Daily defined doses
	International Non proprietary Names for drugs
	B. Information resources in pharmacovigilance
	Basic drug information resources
	Specialised resources for ADRs
	WHO adverse reaction terminologies
	MedDRA and Standardised MedDRA queries
	-
	WHO drug dictionary Eudravigilance medicinal product dictionary
	Edulavignance medicinal product dictionally
	C. Establishing pharmacovigilance Programmeme
	Establishing in a hospital
	Establishment & operation of drug safety department in industry
	Contract Research Organisations (CROs)
	Establishing a national Programmeme
3	A. Vaccine safety surveillance
	Vaccine Pharmacovigilance
	Vaccination failureAdverse events following immunization
	B. Pharmacovigilance methodsPassive surveillance – Spontaneous
	reports and case series
	Stimulated reporting
	Active surveillance – Sentinel sites, drug event monitoring and registries
	Comparative observational studies – Cross sectional study, case
	control study and cohort study
	Targeted clinical investigations
	C. Communication in pharmacovigilance
	Effective communication in Pharmacovigilance
	Communication in Drug Safety Crisis management Communicating with
	Regulatory Agencies, Business Partners, Healthcare facilities & Media





4	A. Safety data generation			
	1. P	Pre clinical phase		
	2. C	2. Clinical phase		
	3. P	ost approval p	hase (PMS)	
	B. ICH (Guidelines for	· Pharmacovigilance	
	1. O	rganization ar	nd objectives of ICH	
	2. E	xpedited repor	rting	
	3. Ir	ndividual case	safety reports	
	4. P	eriodic safety	update reports	
	5. P	ost approval e	xpedited reporting	
	6. P	harmacovigila	ince planning	
		_	ractice in pharmacovigilance studies	
5			s of adverse drug reactions. Genetics related	
	ADR	with example	focusing PK parameters.	
	B. Drug	safety evalua	tion in special population	
	a. Pae	diatrics		
		b. Pregnancy and lactation		
	c. Ger			
	C. CIOMS			
		S Working Gr	oups	
		S Form		
			macovigilance	
		□□ D&C Act and Schedule Y		
	□□ Differences in Indian and global pharmacovigilance requirements			
Mode of	Theory			
examination	Theory			
Weightage	Continuous	Sessional	ESE	
Distribution	Mode	Exam		
	Assessment			
	10 Marks	15	75	
Text book/s*				





Other	Recommended Books (Latest edition)			
References	Textbook of Pharmacovigilance: S K Gupta, Jaypee Brothers, Medical			
	Publishers.			
	Practical Drug Safety from A to Z By Barton Cobert, Pierre Biron, Jones			
	andBartlett Publishers.			
	Mann's Pharmacovigilance:Elizabeth B. Andrews, Nicholas, Wiley			
	Publishers.			
	Stephens' Detection of New Adverse Drug Reactions: John Talbot, Patrick			
	Walle, Wiley Publishers.			
	An Introduction to Pharmacovigilance: Patrick Waller, Wiley Publishers.			
	Cobert's Manual of Drug Safety and Pharmacovigilance: Barton			
	Cobert, Jones & Bartlett Publishers.			





Sch	ool:	SOP		
Programme:		B.Pharm		
Branch:		Semester: VIII		
1	Course Code	BP808ET		
2	Course Title	Cell and molecular biology - Theory		
3	Credits	4		
4	Contact Hours (L-T-P)	3-1-0		
	Course Type	Elective		
5	Course	Upon the completion of the course student shall be able to:		
	Objective	1. Summarize cell and molecular biology history.		
		2. Summarize cellular functioning and composition.		
		3. Describe the chemical foundations of cell biology.		
		a. Summarize the DNA properties of cell biology.		
		b. Describe protein structure and function.		
		4. Describe cellular membrane structure and function.		
		5. Describe basic molecular genetic mechanisms.		
6	Course Outcomes	CO1: Students would be able to understand the concept of cell theory and basics of cellular structure and the mechanism of immune system.		
		CO2: Students would be able to apply cellular reproduction in eukaryotic cells and increase their knowledge about nucleic acids		
		CO3: Students would be able to interpret DNA, RNA and their role in central dogma of life, Protein synthesis and types of RNA.		
		CO4: Students would be able to apply Transgenics and Genomic Analysis, Cell Cycle analysis, role of genetics, mitosis and meiosis.		
		CO5: Students would be able to explain Cell Signals, Receptors for Cell Signals, Signaling Pathways, Misregulation of Signaling Pathways and Protein- Kinases: Functioning		
		CO6: Students will be able to analyze different aspect of cell function which will help them to think about research involving cell biology.		





		WWW.Jidi Ma.de.iii		
7	Course			
	Description	• Cell biology is a branch of biology that studies cells – their physiological		
		properties, their structure, the organelles they contain, interactions with		
		their environment, their life cycle, division, death and cell function.		
		• This is done both on a microscopic and molecular level.		
		• Cell biology research encompasses both the great diversity of single-		
		celled organisms like bacteria and protozoa, as well as the many		
		specialized cells in multi-cellular organisms such as humans, plants, and		
		sponges.		
8	Outline syllabus			
	1	Unit I		
		a) Cell and Molecular Biology: Definitions theory and basics and		
		Applications.		
		b) Cell and Molecular Biology: History and Summation.		
		c) Properties of cells and cell membrane.		
		d) Prokaryotic versus Eukaryotic		
		e) Cellular Reproduction		
		f) chemical Foundations – an Introduction and Reactions (Types)		
	2	Unit II		
		a). DNA and the Flow of Molecular Information		
		b). DNA Functioning		
		c). DNA and RNA		
		d). Types of RNA		
		e). Transcription and Translation		
	3	Unit III		
		a). Proteins: Defined and Amino Acids		
		b). Protein Structure		
		c). c)Regularities in Protein Pathways		
		d). Cellular Processes		
	4	e). Positive Control and significance of Protein Synthesis Unit IV		
	4	a). Science of Genetics		
		b). Transgenics and Genomic Analysis		
		c). Cell Cycle analysis		
		d). Mitosis and Meiosis		
		e). Cellular Activities and Checkpoints		
·	5	Unit V		
		a). Cell Signals: Introduction		
		b). Receptors for Cell Signals		
		c). Signaling Pathways: Overview		
		d). Misregulation of Signaling		
		e). Pathways Protein-Kinases: Functioning		
	Mode of	Theory		
	examination			





Weightage	Continuous	Sessional	ESE
Distribution	Mode	Exam	
	Assessment		
		15	75
	10 Marks		
Text book/s*			
Other	Recommended Books (latest edition):		
References	1. W.B. Hugo and A.D. Russel: Pharmaceutical Microbiology, Blackwell Scientific publications, Oxford London.		
		tt and Dunn., n, CBS Publish	Industrial Microbiology, Distributors, Delhi.
	3. Pelcza	r, Chan Kreig,	Microbiology, Tata McGraw Hill edn.
			Balliere Tindall and Cox: Pharmaceutical
	Micro	biology.	
	5. Rose: Industrial Microbiology.		
	6. Probisher, Hinsdill et al: Fundamentals of Microbiology, 9th ed.		
	Japan		
	_		s: Tutorial Pharmacy, CBS Publisher and
	Distrib		
		er: Microbial T	
			als of Microbiology.
			utical Microbiology, Vallabh Prakashan, Delhi
	11. Bergeys manual of systematic bacteriology, Williams and Wilkins-		
	A Waverly		
	a. company		
			J.J. Pasternak: Molecular Biotechnology:
		oles and	ambinantDNA. ACM Duage Weshington D.C.
			ombinantDNA: ASM Press Washington D.C.
	13. RA	A Goldshy et. a	ıl., : Kuby Immunology.





Scho	ool:	SOP				
Program:		B.Pharm				
Bra	,	Semester: 3				
1	Course Code	KVE 301T				
2	Course Title	Universal Human values and Professional Ethics				
3	Credits	3				
4	Contact Hours (L-T-P)	3-0-0				
	Course Type	Compulsory				
5	Course Objective	 To help the students appreciate the essential complementarily between 'VALUES' and 'SKILLS' to ensure sustained happiness and prosperity, which are the core aspirations of all human beings. To facilitate the development of a Holistic perspective among students towards life and profession as well as towards happiness and prosperity based on a correct understanding of the Human reality and the rest of Existence. Such a holistic perspective forms the basis of Universal Human Values and movement towards value-based living in a natural way To highlight plausible implications of such a Holistic understanding in terms of ethical human conduct, trustful and mutually fulfilling human behavior and mutually enriching interaction with Nature Thus, this course is intended to provide a much needed orientational input in value education to the young enquiring minds. 				
6	Course Outcomes	CO1: The students will Understand the concept of human aspirations, relationships, happiness and prosperity. CO2: The student will be able to summarize the fulfillment human aspirations and live in harmony at various levels. CO3: The student will be able to utilize the professional competence for augmenting universal human order. CO4: The students will analyze the correlation of various body systems and how they result in particular kind of functions. CO5: The students would Identify the scope and characteristics of people-friendly and eco-friendly production systems, technologies and management models. CO6: The students would socially and ecologically responsible and adopt ethical practices.				
7	Course Description	The students are able to sincerely evaluate the course and share with their friends. They are also able to suggest measures to make the course more effective and relevant. They are also able to make use of their understanding in the course for a happy and prosperous society.				





8	Outline syllabus				
0	1	 UNIT-I D. Course Introduction - Need, Basic Guidelines, Content and Process for Value Education. Understanding the need, basic guidelines, content and process for Value Education. Self Exploration—what is it? - its content and process; 'Natural Acceptance' and Experiential Validation- as the mechanism for self exploration E. Continuous Happiness and Prosperity- A look at basic Human Aspirations Right understanding, Relationship and Physical Facilities- the basic requirements for fulfillment of aspirations of every human being with their correct priority F. Understanding Happiness and Prosperity correctly- A critical appraisal of the current scenario. Method to fulfill the above human aspirations: understanding and living in harmony at various levels 			
	2	 UNIT-II D. Understanding Harmony in the Human Being - Harmony in Myself! Understanding human being as a co-existence of the sentient 'I' and the material 'Body' 8. Understanding the needs of Self ('I') and 'Body' - Sukh and Suvidha E. Understanding the Body as an instrument of 'I' (I being the doer, seer and enjoyer) 10. Understanding the characteristics and activities of 'I' and harmony in 'I' F. Understanding the harmony of I with the Body: Sanyam and Swasthya; correct appraisal of Physical needs, meaning of Prosperity in detail Programs to ensure Sanyam and Swasthya - Practice Exercises and Case Studies will be taken up in Practice Sessions. 			





3	UNIT-III
3	 D. Understanding Harmony in the Family and Society- Harmony in Human Human Relationship Understanding Harmony in the family – the basic unit of human interaction 14. Understanding values in human-human relationship; meaning of Nyaya and program for its fulfillment to ensure Ubhay-tripti; Trust (Vishwas) and Respect (Samman) as the foundational values of relationship E. Understanding the meaning of Vishwas; Difference between intention and competence 16. Understanding the meaning of Samman, Difference between respect and differentiation; the other salient values in relationship micturition reflex and role of kidneys in acid base balance, role of RAS in kidneyand disorders of kidney. F. Understanding the harmony in the society (society being an extension of family): Samadhan, Samridhi, Abhay, Sah-astitva as comprehensive Human Goals 18. Visualizing a universal harmonious order in society- Undivided Society (Akhand Samaj), Universal Order (Sarvabhaum Vyawastha)- from
4	family to world family! - Practice Exercises and Case Studies will be taken up in Practice Sessions. UNIT-IV B. Understanding Harmony in the Nature and Existence - Whole existence as Coexistence Understanding the harmony in the Nature 20. Interconnectedness and mutual fulfillment among the four orders of nature- recyclability and self-regulation in nature Understanding Existence as Co-existence (Sah-astitva) of mutually interacting units in all-pervasive space Holistic perception of harmony at all levels of existence - Practice Exercises and Case Studies will be taken up in Practice Sessions.





5	UNIT-V			
	D. Implications of the above Holistic Understanding of Harmony on Professional Ethics			
	Natural acceptance of human values 24. Definitiveness of Ethical Human Conduct			
	25. Basis for Humanistic Education, Humanistic Constitution and Humanistic			
	Universal Order			
	E. Competence in profession	onal ethics:		
	a) Ability to utilize th	e professional compet	ence for augmenting universal human	
	order b) Ability to idea	ntify the scope and cha	aracteristics of people-friendly and eco	
	friendly production s	systems, c) Ability t	to identify and develop appropriate	
	technologies and manage	gement patterns for abo	ove production systems.	
	F. Case studies of typical holistic technologies, management models and production systems 28. Strategy for transition from the present state to Universal Human Order: a) At the level of individual: as socially and ecologically responsible engineers, technologists and managers b) At the level of society: as mutually enriching institutions and organizations			
Mode of	Theory			
examination	-			
Weightage	Continuous Mode	Sessional Exam	ESE	
Distribution	Assessment			
TD . 1 1 / 16	10 Marks	15	50	
Text book/s* Reference Material	a. The text book R.R Gaur, R Sangal, G P Bagaria, A foundation course in Human Values and professional Ethics, Excel books, New Delhi, 2010, ISBN 978-8-174-46781-2			
	b. The teacher's manual R.R Gaur, R Sangal, G P Bagaria, A foundation course in Human Values and professional Ethics – Teachers Manual, Excel books, New Delhi, 2010			
	c. A set of DVDs containing Video of Teachers' Orientation Program. PPTs of Lectures and Practice Sessions. Audio-visual material for use in the practice sessions.			